

Master course: Astrochemistry I:

Lecture 7: Chemistry in star-forming regions

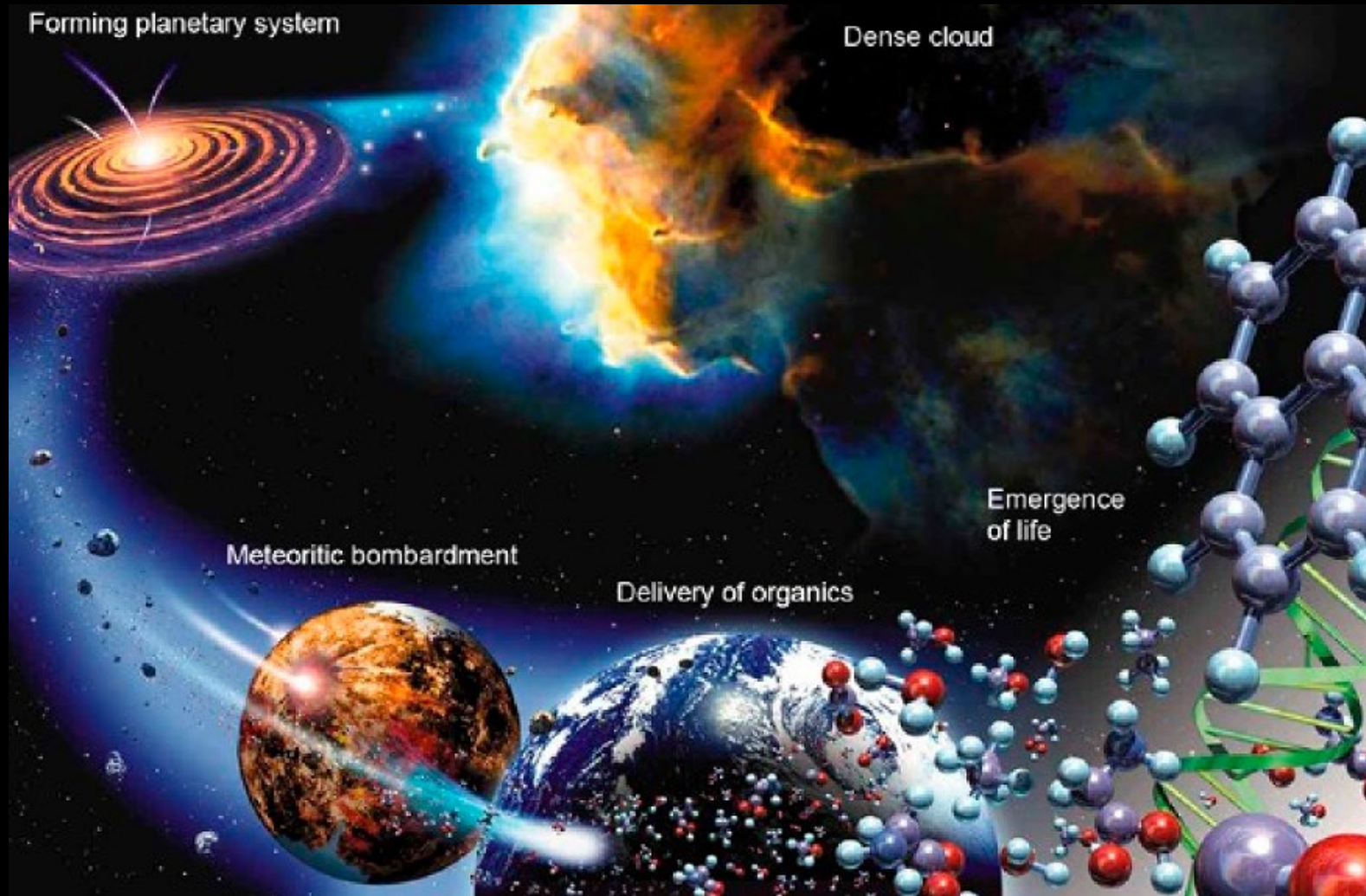
Reading list:

van Dishoeck Ch.8
“Astrochemistry of
Dense Protostellar and
Protoplanetary
Environments” in
Thronson, Stiavelli,
Tielens (2009)

van Dishoeck (2014)
Faraday discussions

Caselli & Ceccarelli
(2012)

Herbst & van Dishoeck
(2009) ARAA



Dr. Helgi Rafn Hrodmarsson – UPEC – 2026-2027

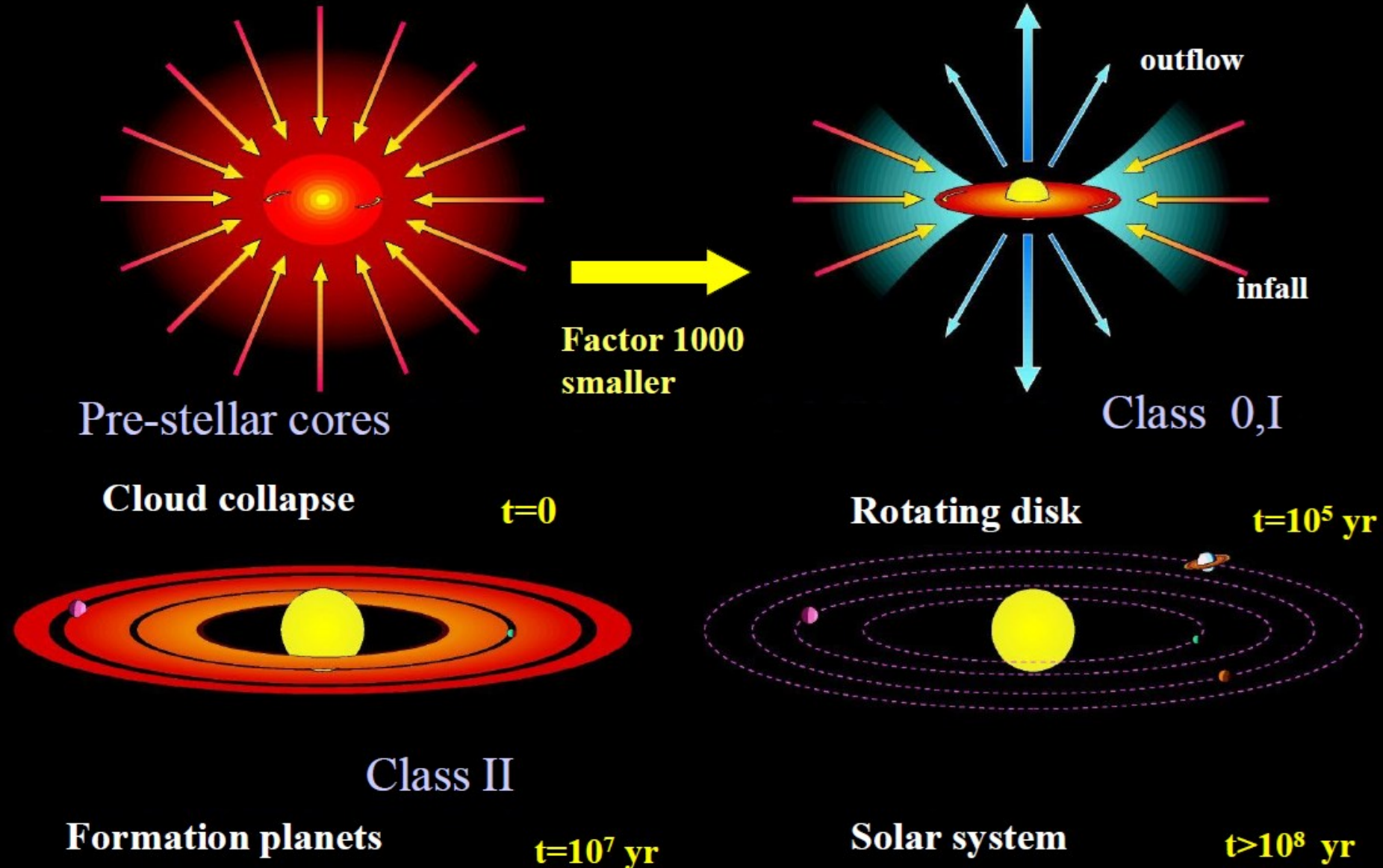
Outline

- Introduction
- Observational techniques
- Chemical scenario
- Cold outer envelopes around YSOs
- Warm inner envelopes around YSOs
- Hot cores
- Protoplanetary disks
- Comets

7.1 Introduction

- Star formation accompanied by enormous changes in physical parameters n , $T \Rightarrow$ response of chemistry?
- Which molecules are good diagnostics of various stages: collapse, outflow, hot cores?
- Molecular lines are used to probe physical conditions and dynamics $\Rightarrow$ need to choose the right molecule!
- In which form are molecules incorporated into circumstellar disks and new planetary systems?

Scenario for star- and planet formation



7.2 Observations

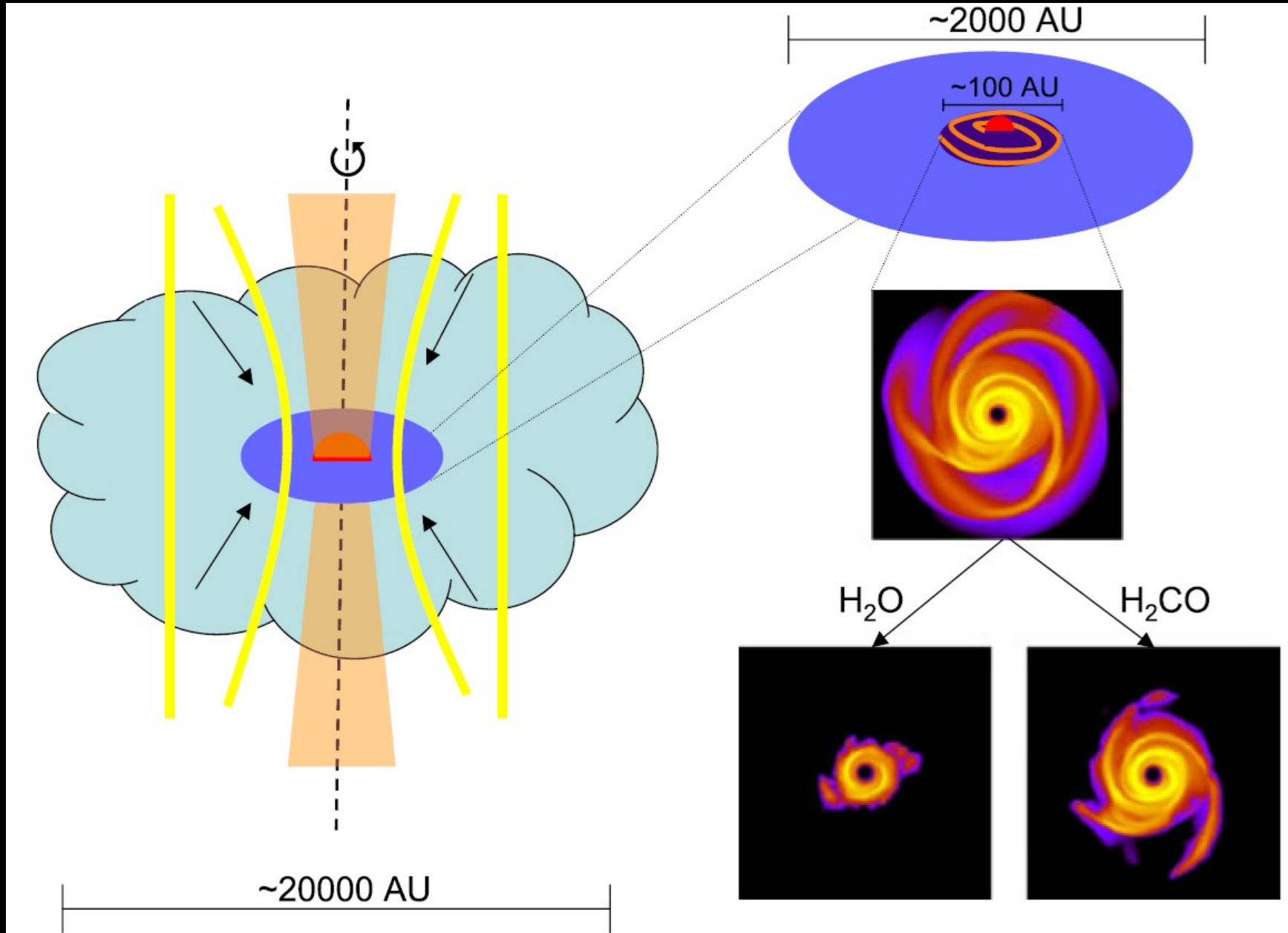
- **Submillimeter:**

- Very high spectral resolution ($R > 10^6$, < 0.1 km/s)
- Many gas-phase molecules with abundances down to 10^{-11} wrt H_2
- Emission $\Rightarrow$ map of region

- **Infrared:**

- Moderate spectral resolution ($R \sim 10^3\text{--}10^5$)
- Gases and solids with abundances down to $10^{-7}\text{--}10^{-8}$ wrt H_2
 - Probe major reservoirs of C, N, and O
- Molecules without permanent dipole moments (H_2 , C_2H_2 , CH_4 , CO_2 , CH_3 , ...)
- Absorption $\Rightarrow$ pencil beam line-of-sight; emission only from hot gas

Typical sizes



Typical sizes

Linear size	Angular size	
	Taurus 140 pc	Orion 450 pc
5 AU Inner disk	0.04''	0.01''
100 AU Outer disk	0.7''	0.2''
1000 AU YSO envelope	7''	2''
10000 AU = 0.05 pc Cloud core	74''	23''

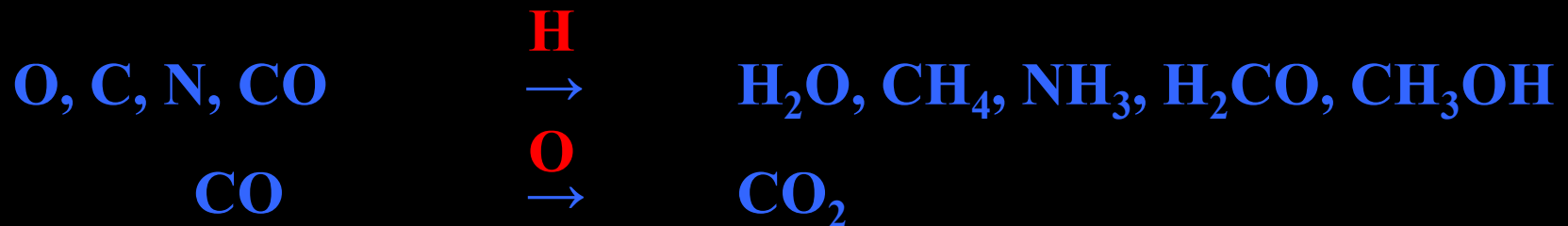
ALMA needed to resolve relevant scales; most data spatially unresolved pre-ALMA

Information from lines

- Line strength $\Rightarrow$ abundances
- Line ratios $\Rightarrow$ temperatures, densities
- Line profiles $\Rightarrow$ kinematics

7.3 Chemical scenario

- **Starless cores:** $T \approx 10\text{--}20\text{ K}$, $n \approx 10^4\text{ cm}^{-3}$
 - Quiescent ion-molecule chemistry
 - Radicals and carbon chains
 - Ices start forming
- **Pre-stellar cores and collapse:** $T \approx 10\text{ K}$, $n \approx 10^5\text{--}10^8\text{ cm}^{-3}$
 - Heavy freeze-out of molecules onto grains
 - Grain surface reactions produce new species:



Chemical scenario (continued)

- **Embedded YSO phase:** $T \approx 10\text{--}300\text{ K}$, $n \approx 10^5\text{--}10^9\text{ cm}^{-3}$
 - Gas and dust heated by accretion luminosity, UV + X rays from young star; production of more complex molecules
 - Sublimation of molecules from grains: H_2O , CH_3OH , CH_4 , complex organics...
 - Sequence according to sublimation temperatures
 - Hot core chemistry: complex organics, HCN, ...
- **Bipolar outflows:** $T \approx 200\text{--}2000\text{ K}$, $n \approx 10^5\text{--}10^7\text{ cm}^{-3}$
 - Interaction between outflow and envelope => shocks
 - High-T chemistry: H_2O , ...
 - Return of icy mantles to gas: H_2O , CH_3OH , CH_4 , ...
 - Destroy grain cores: SiO, ...

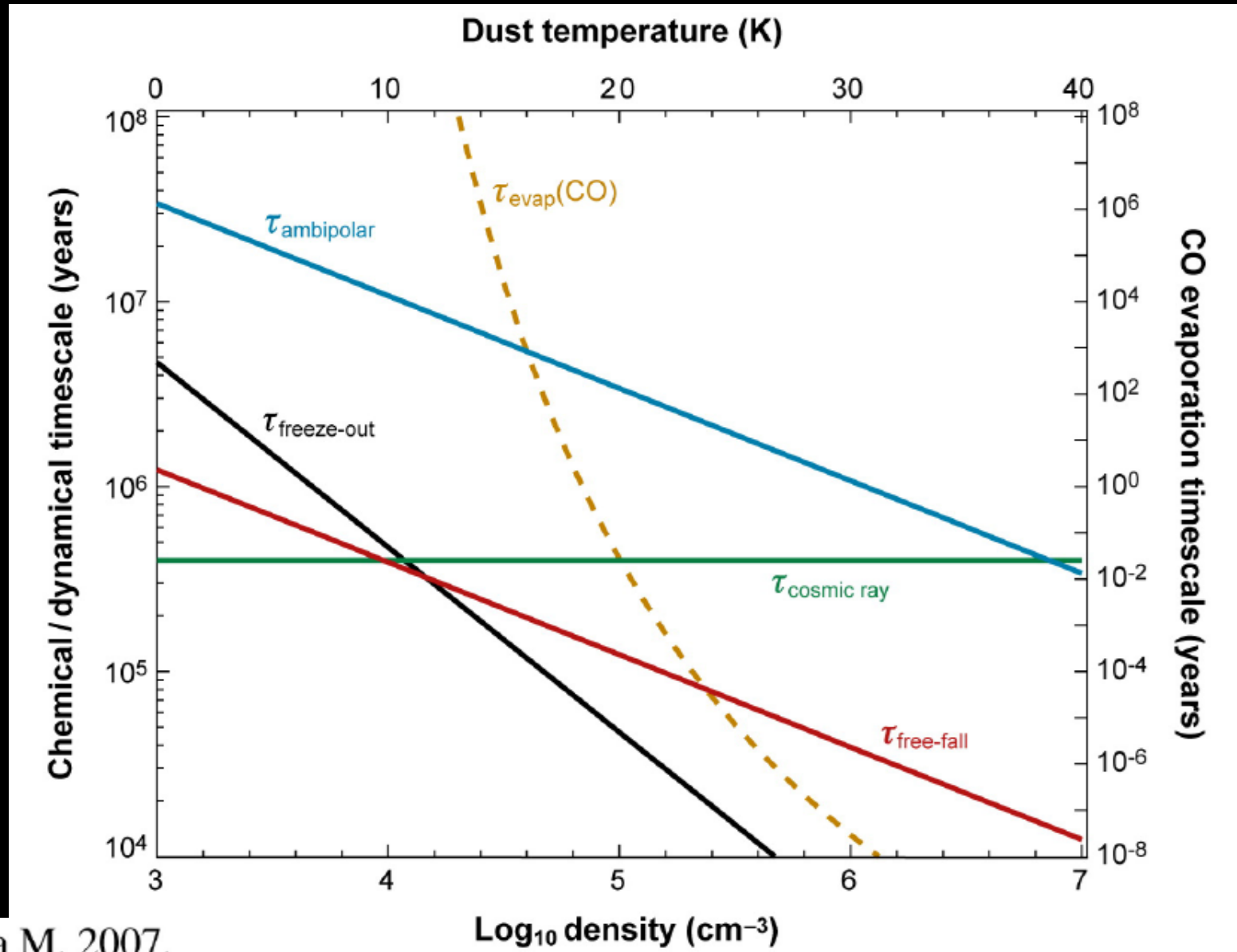
Lecture 5

7.4 Cold outer envelopes: freeze-out

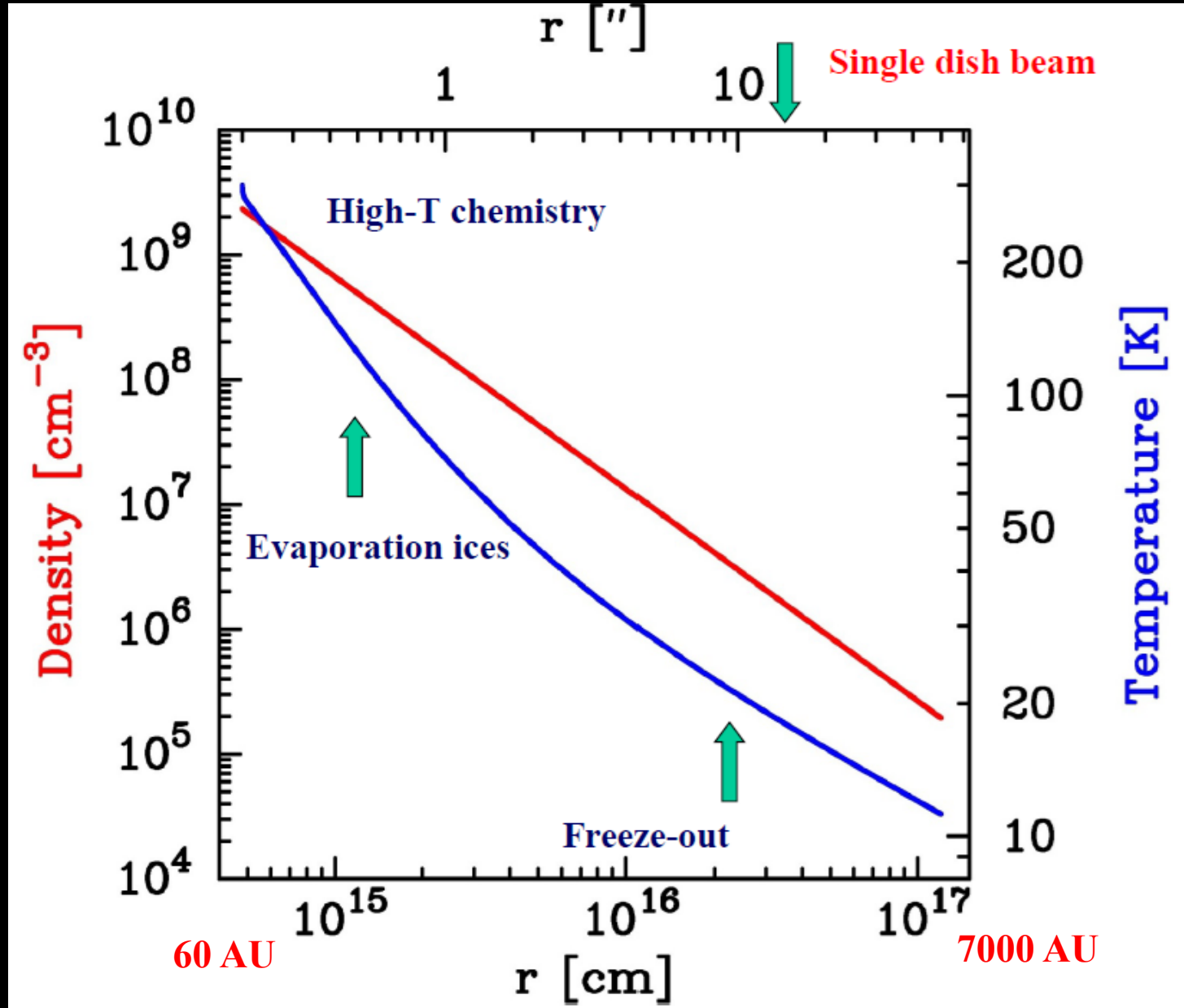
- When protostar starts to heat its surroundings, temperature gradient set up through envelope
 - Outer envelope cold $\Rightarrow$ significant freeze-out expected, as in pre-stellar cores
 - Inner envelope warm $\Rightarrow$ sublimation of ices
- Surveys of low- and high-mass YSOs
 - Constrain T , n from dust continuum data
 - Use line data to fit abundance
 - Simplest model: constant abundance with radius
 - More sophisticated models: drop, jump abundance profiles

Lecture 6

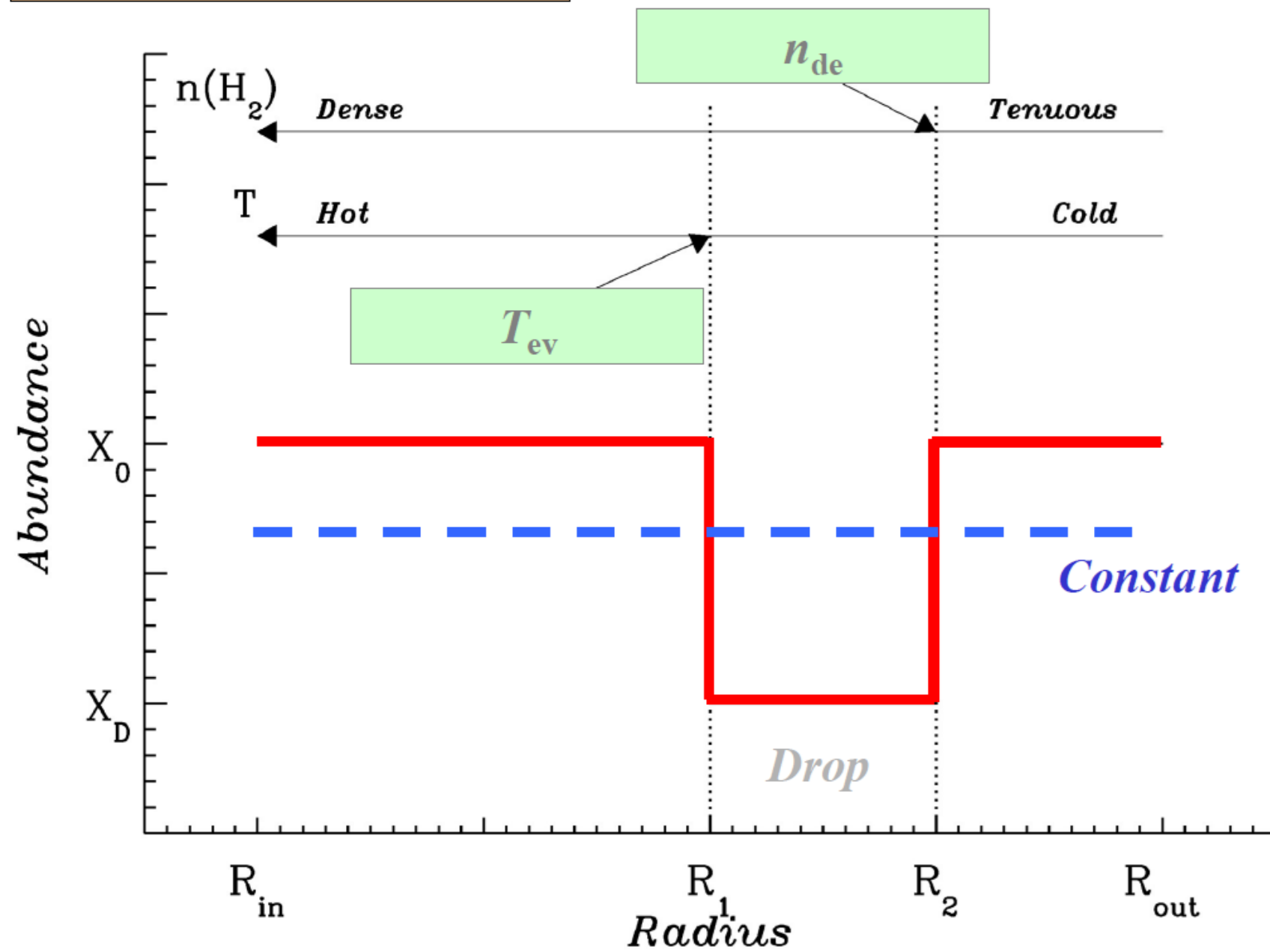
Time scales



Physical structure of protostellar envelope



"Drop" abundance model



Sublimation temperatures

Species	$T_{\text{evap-lab}}$ (K)
H ₂ O	150
CH ₃ OH	99
HCN	95
SO ₂	83
NH ₃	78
CO ₂	72
H ₂ CO	64
H ₂ S	57
CH ₄	31
CO	25
N ₂	22

Mumma et al. 1993,
PPIII

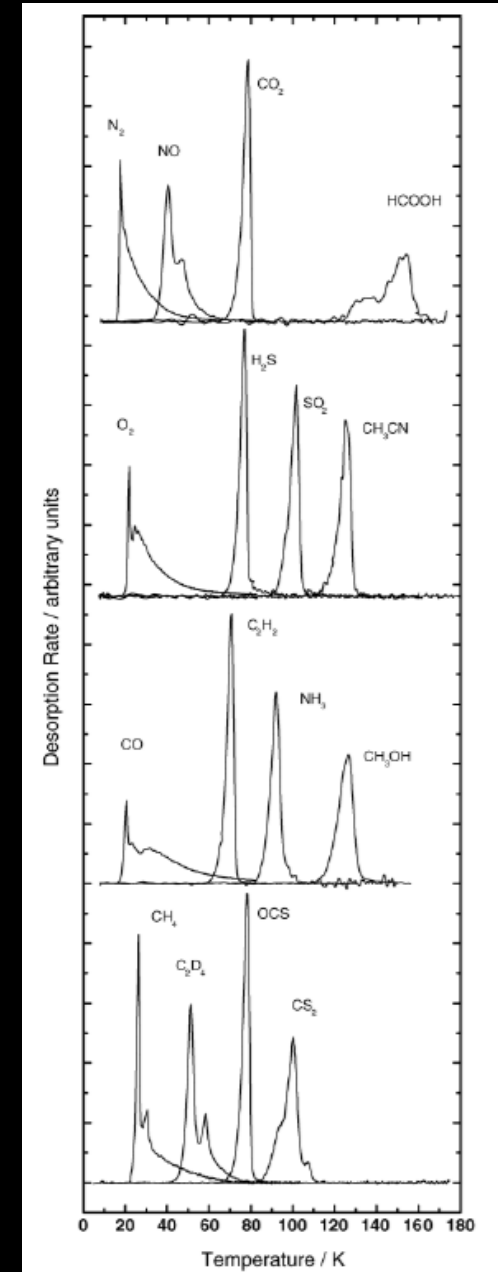
Collings et al. 2004

Sublimation temperatures for *pure* ices as measured in *lab*; values in space are lower because of lower densities

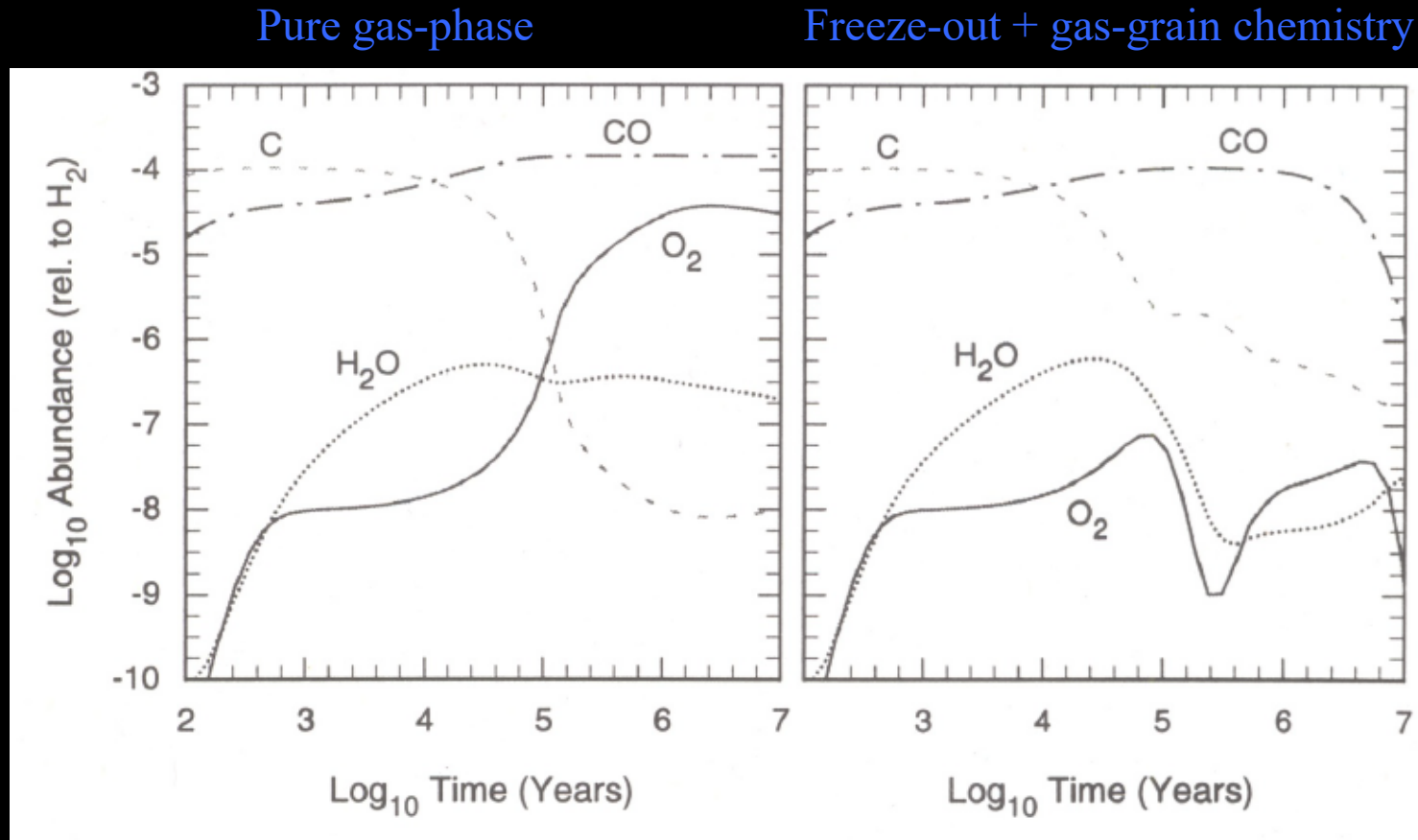
TPD experiments

- TPD = Temperature programmed desorption
- Experiments for pure ices
- Results for mixed ices depend on mixing ratio and environment
- Example: fraction ($\sim 5\%$) of CO mixed in H_2O ice evaporates at ~ 100 K in space rather than ~ 20 K.

Collings et al. 2004
Viti et al. 2004



Oxygen chemistry: freeze-out

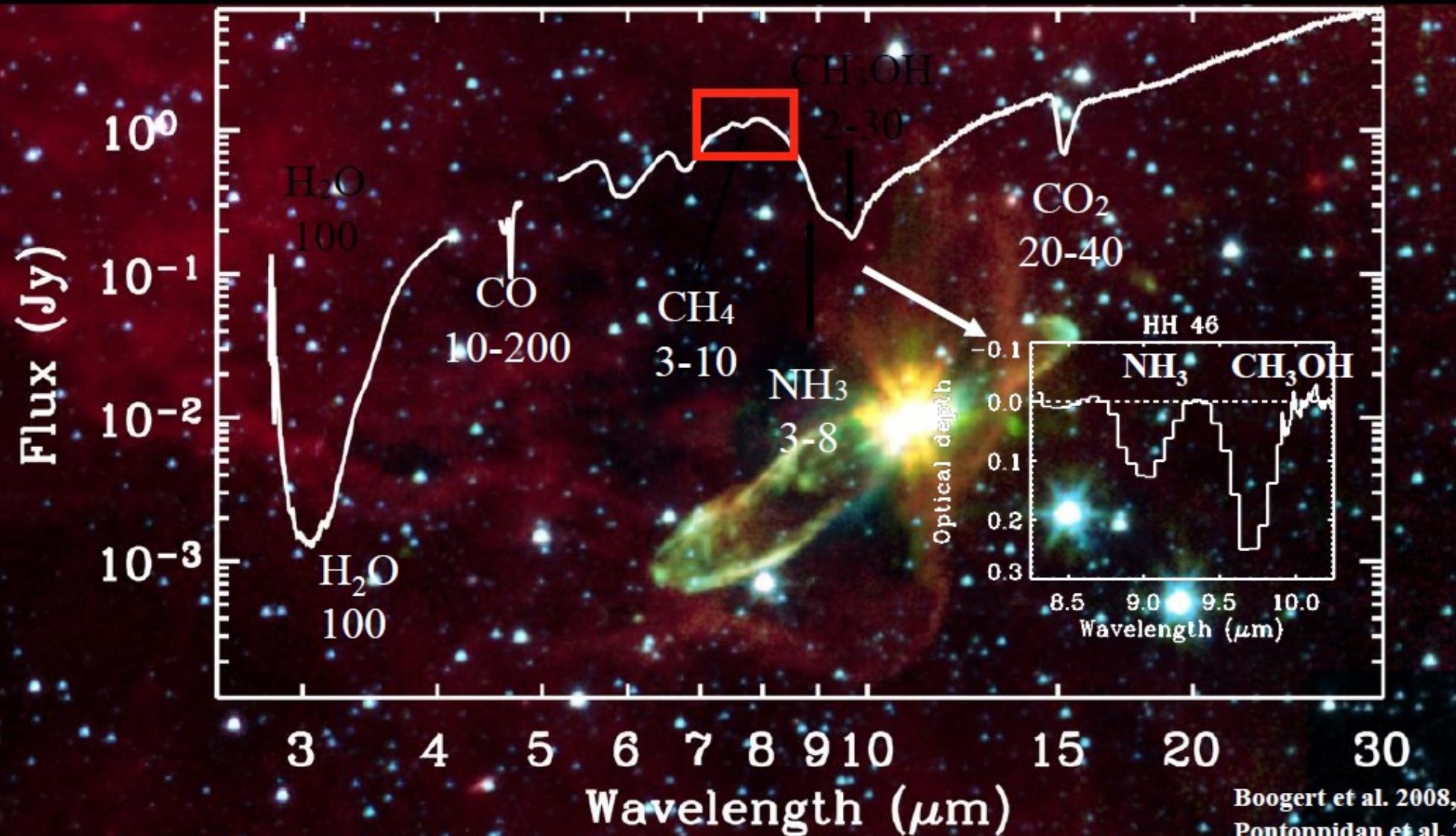


$\text{O} \rightarrow \text{H}_2\text{O}^{\text{gr}} \rightarrow \text{sticks}$
 $\text{C} \rightarrow \text{CH}_4^{\text{gr}} \rightarrow \text{evaporates}$

Bergin et al. 2000

- Gas-grain chemistry and differential evaporation can explain low H_2O and O_2 in dark clouds, as found by SWAS, ODIN, and Herschel

Ices around low-mass protostars



Montage: S. Bottinelli

- Variations seen in NH_3 , CH_3OH and OCN^-

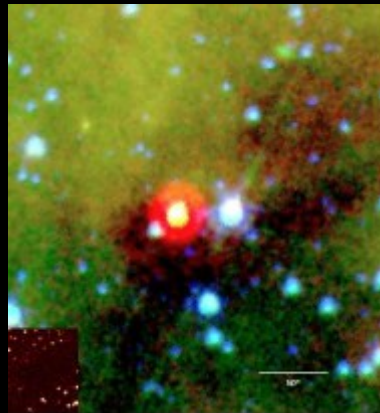
Boogert et al. 2008,
Pontoppidan et al. 2008,
Öberg et al. 2008
Bottinelli et al. 2010

Infrared observations of ices

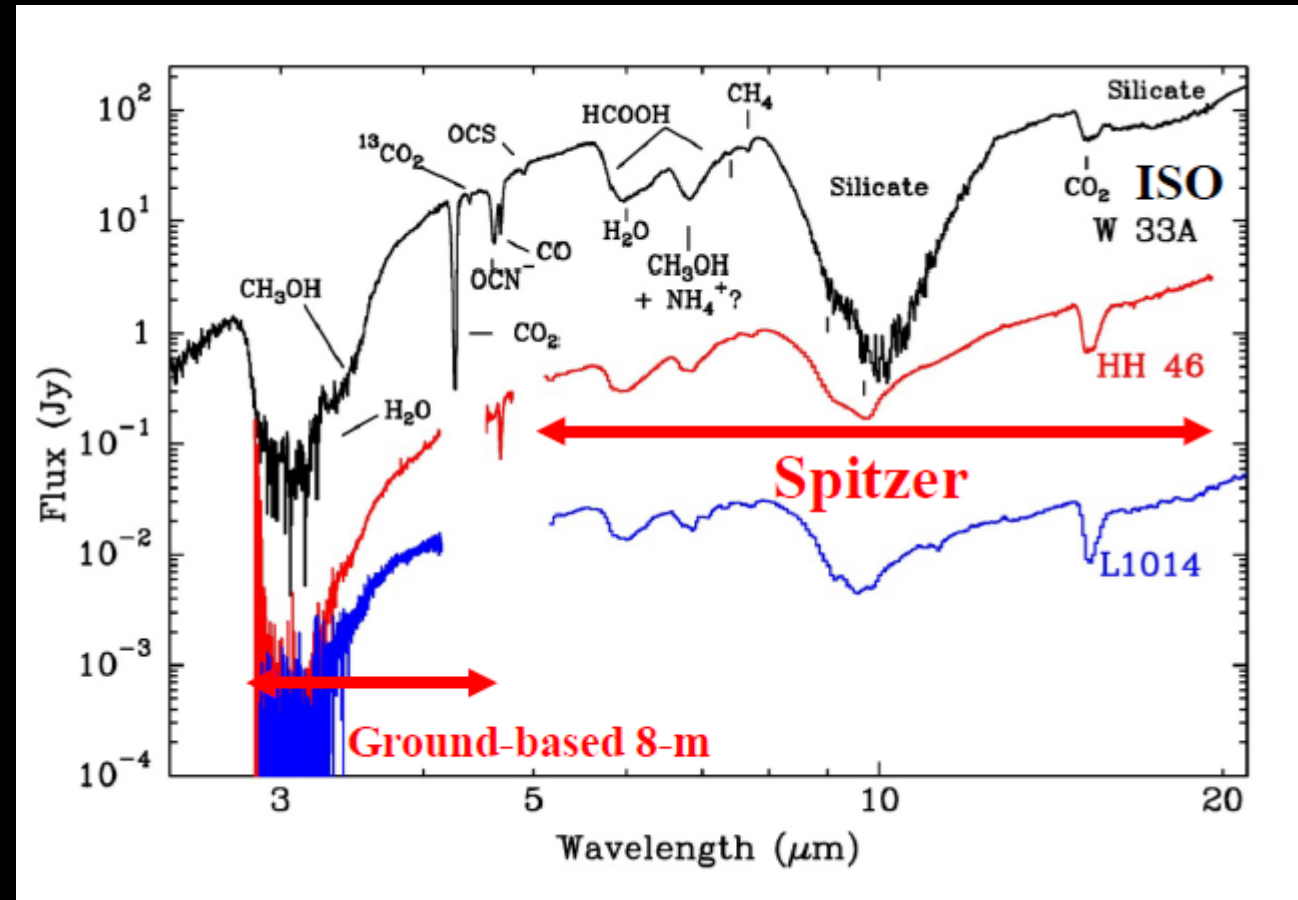
HH 46: solar-mass YSO



L1014: substellar YSO



From 10^5 to $< 0.1 L_{\odot}$ objects



Boogert et al. 2004, 2008
Pontoppidan et al. 2008
Öberg et al. 2008, 2011
Bottinelli et al. 2010, 2015

- Note similarity of ices over large range of luminosities
- Ices contain significant fraction of heavy elements

Median ice abundances

Ice feature	Low mass	High mass	Background
H ₂ O	100	100	100
CO	29	13	31
CO ₂	29	13	38
CH ₃ OH	3	4	4
NH ₃	5	5	-
CH ₄	5	2	-
OCN ⁻	0.3	0.6	-

Öberg et al. 2011

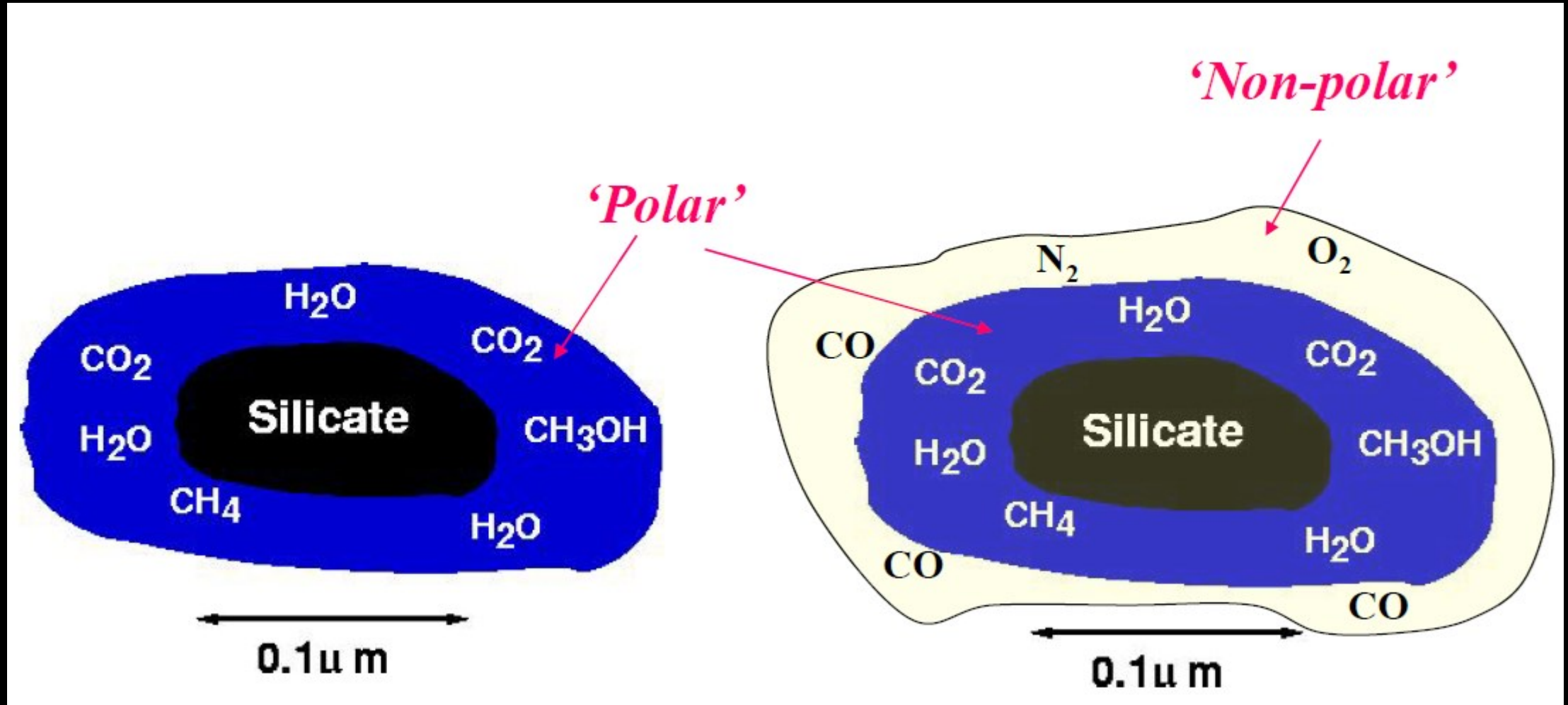
Boogert et al. 2015

$$\text{H}_2\text{O ice} / \text{H}_2 \sim (0.5\text{--}1) \times 10^{-4}$$

Different ice phases

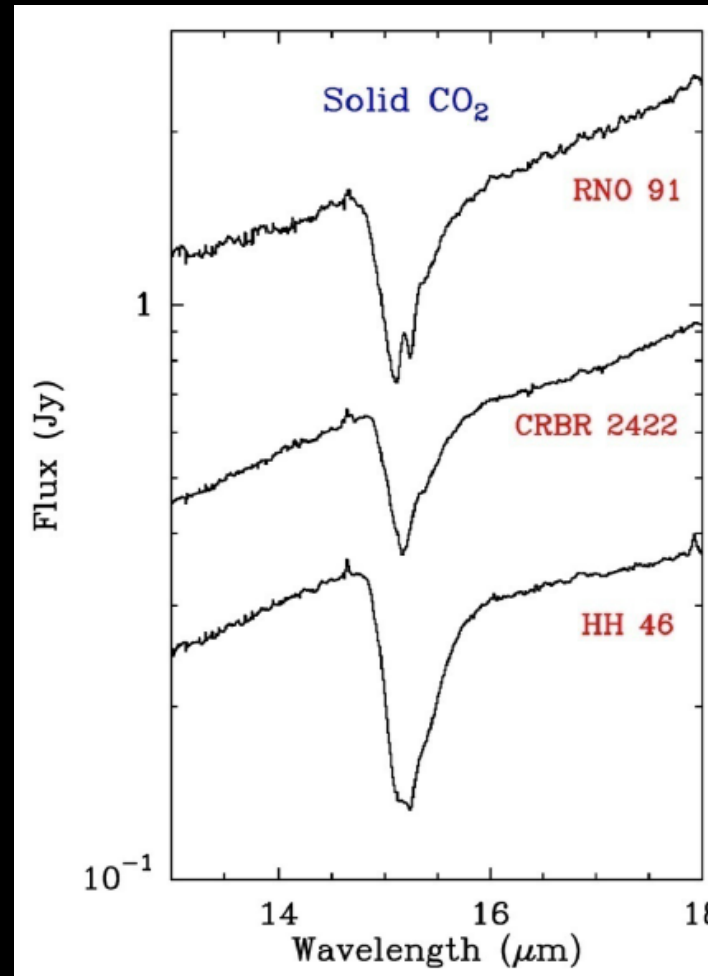
- Comparison of observed CO, CO₂ line profiles with lab spectra => multiple phases
 - **'Polar'**: H₂O-rich, hydrogen bonding
 - **'Apolar' or 'Non-polar'**: CO-rich, van der Waals bonding
 - **Segregated**: CO₂-rich, CH₃OH-rich
- Possible explanations
 - *Diverse condensation*: H-rich vs CO-rich gas
 - See Fig. Tielens **Lecture 6**
 - *Diverse sublimation*: temperature gradient results in differential outgassing and restructuring of ices

Different ice phases



- Different ice phases can be seen in IR spectra!

Observations CO₂ 15 μ m

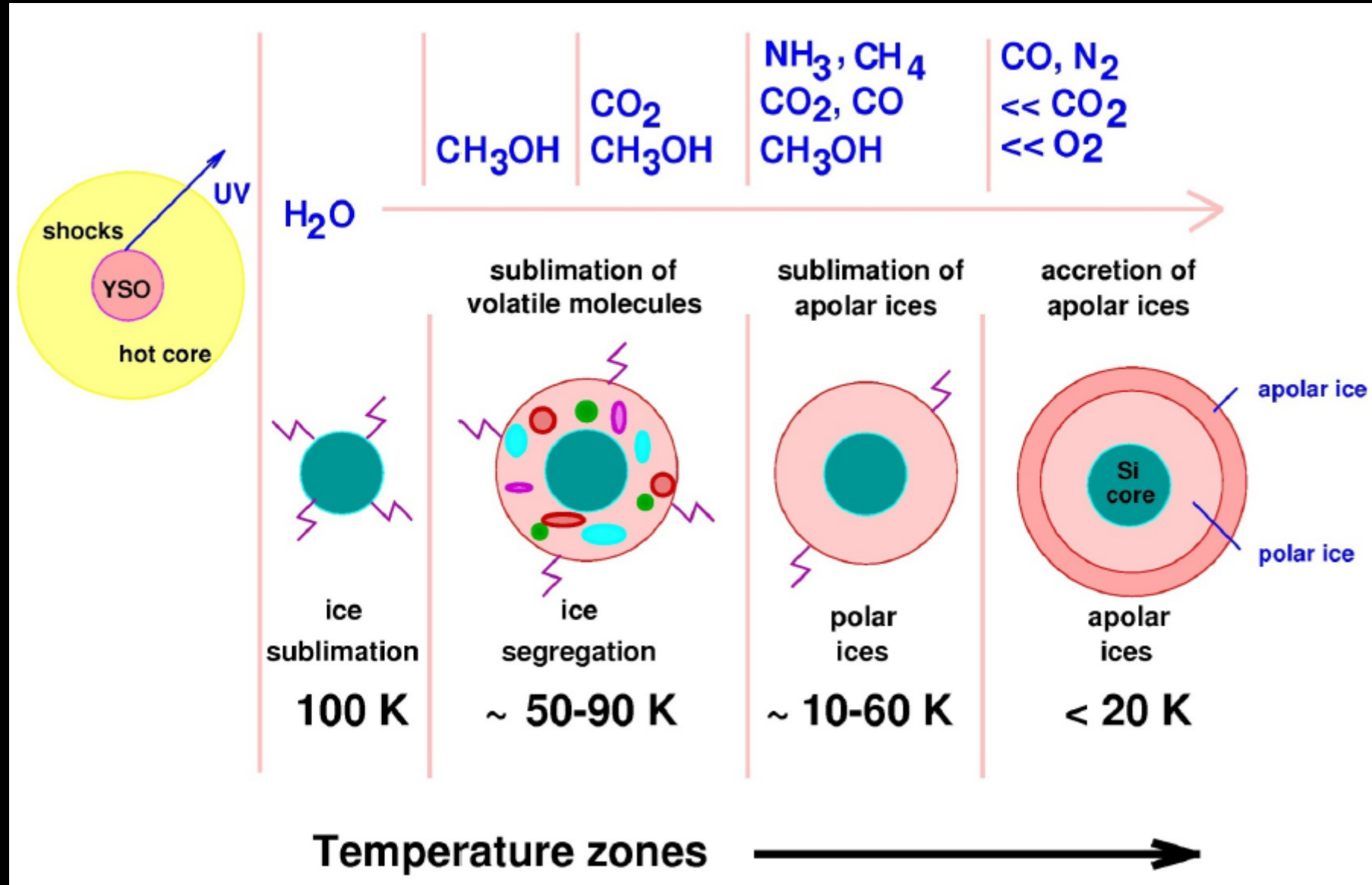


Low-mass YSOs

Obs: Pontoppidan et al. 2008
Lab: Ehrenfreund et al. 1997

- Shape of ice band depends on ice environment (polar or apolar)
- CO₂ ice develops double peak structure upon heating $\rightarrow$ ice segregation (irreversible!) $\Rightarrow$ temperature (history) diagnostic

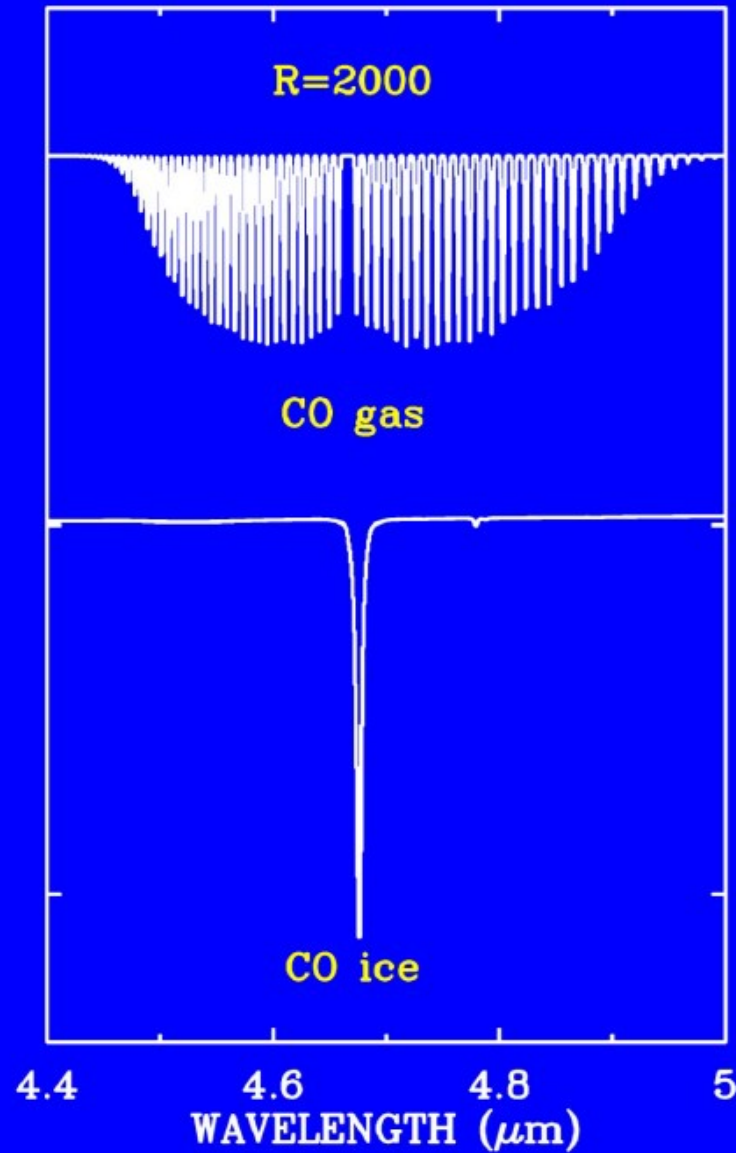
Different ice phases in YSO envelopes



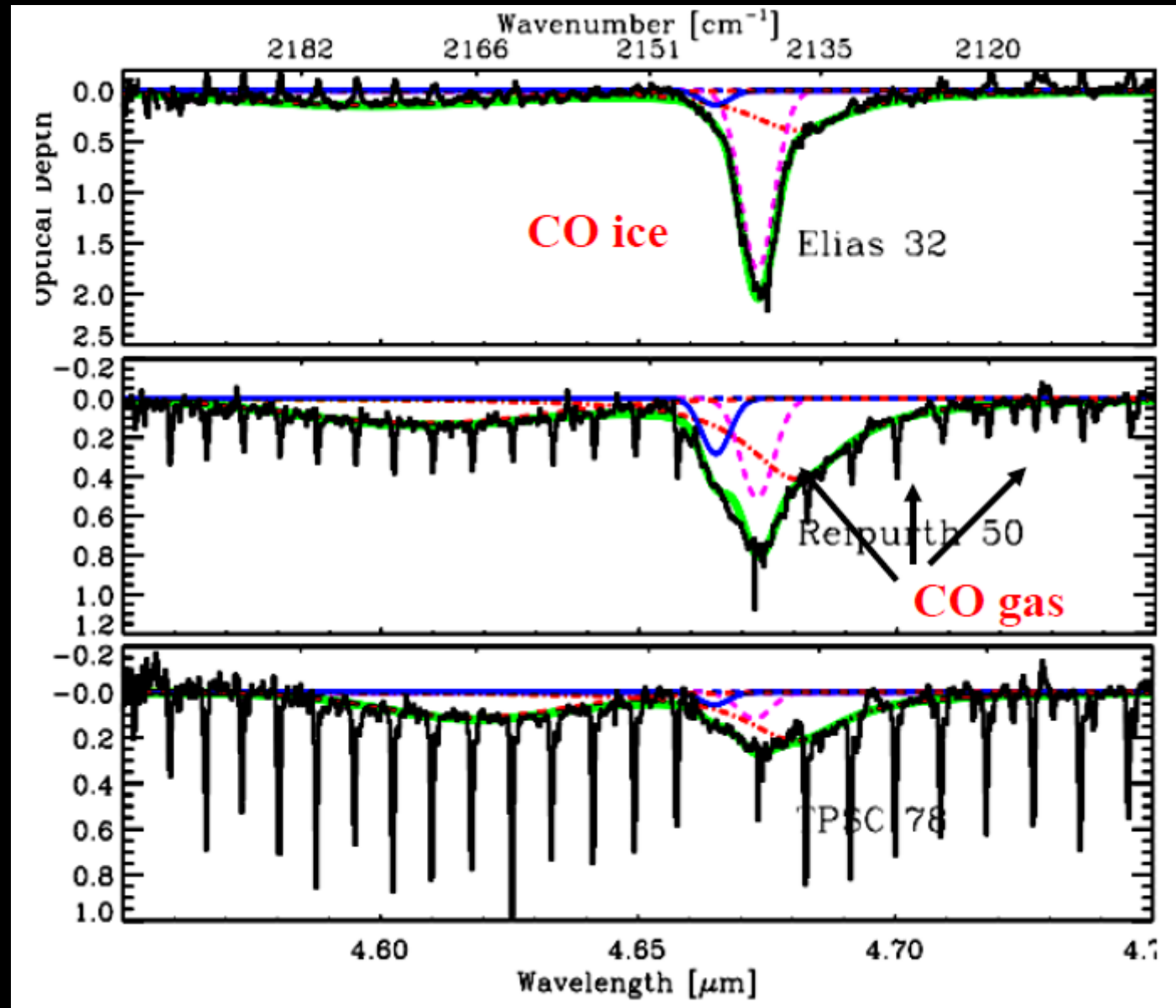
7.5 Warm inner envelope: ice sublimation

- Evaporation of ices directly observed in IR spectra
- Indirectly inferred from sub-mm data as a jump in abundance in inner envelope

Gas vs ice: IR



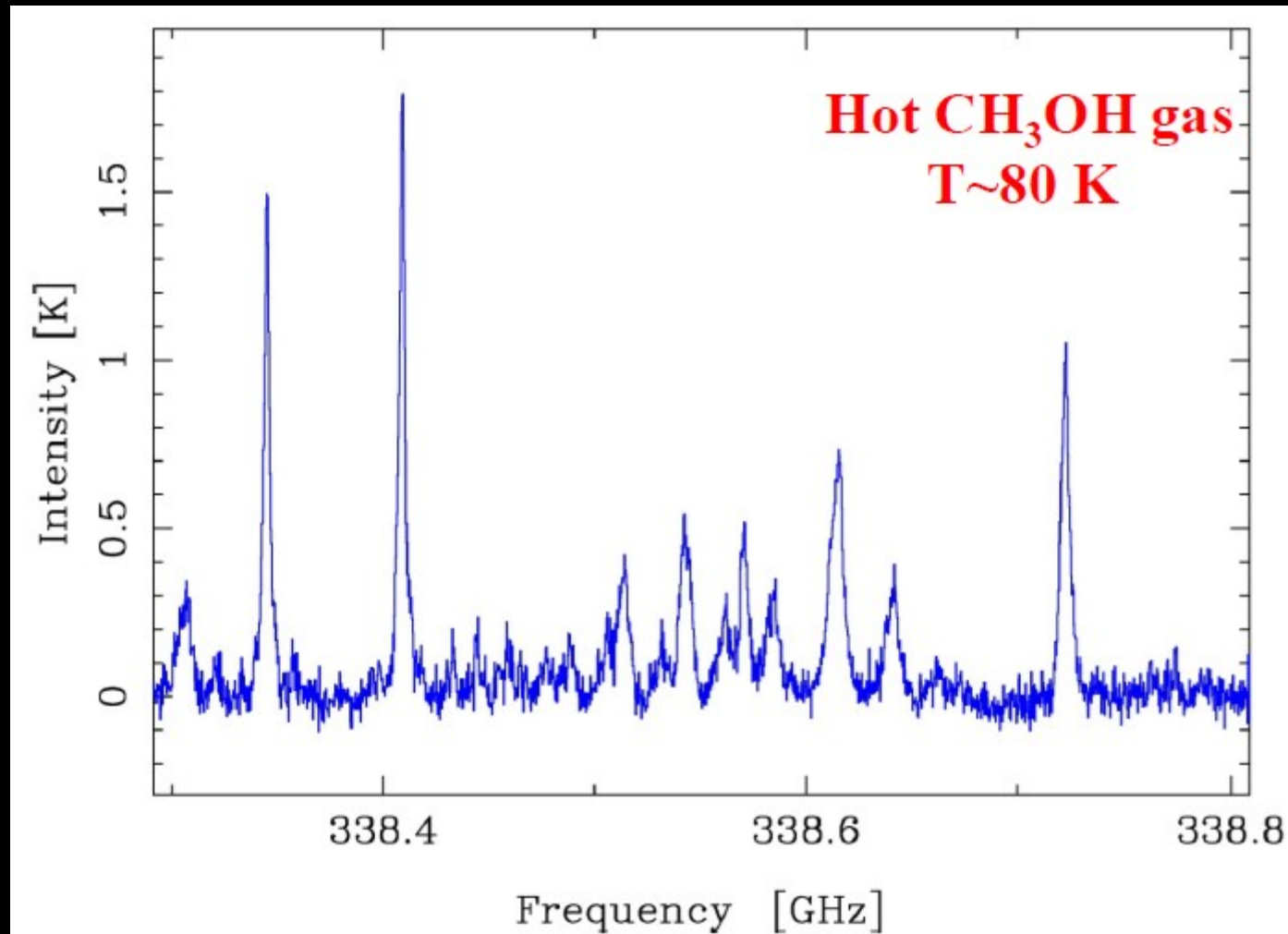
Ice sublimation low-mass YSOs



T

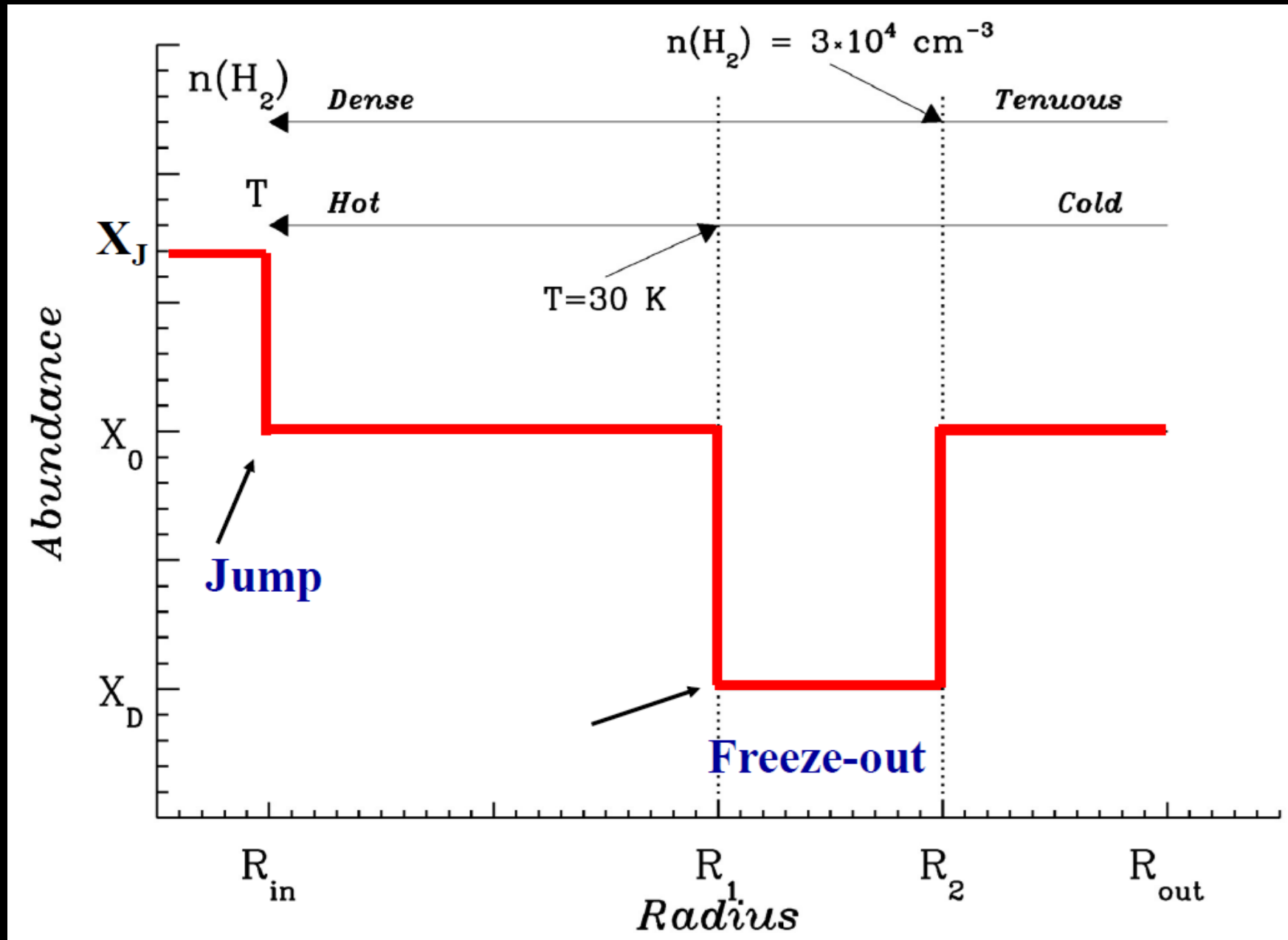


Hot methanol around solar-mass protostar

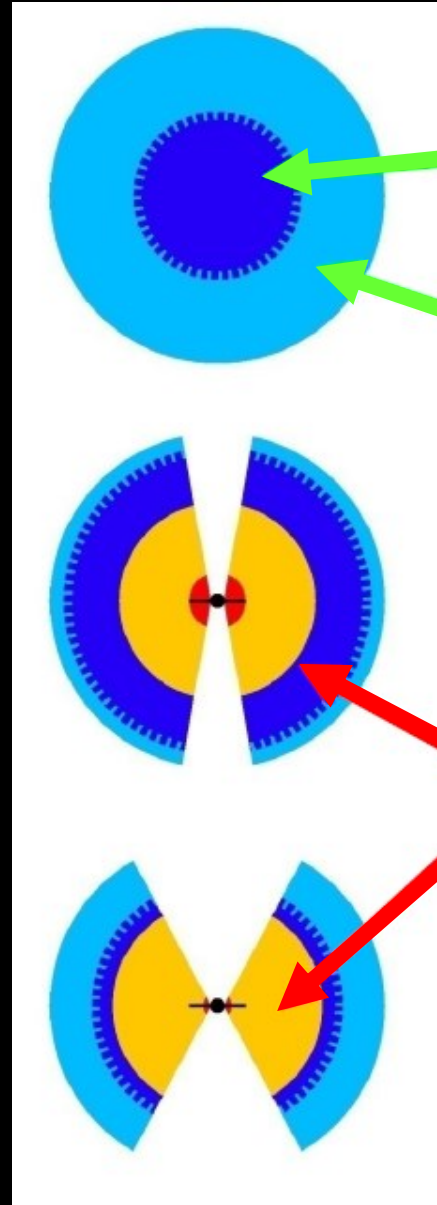
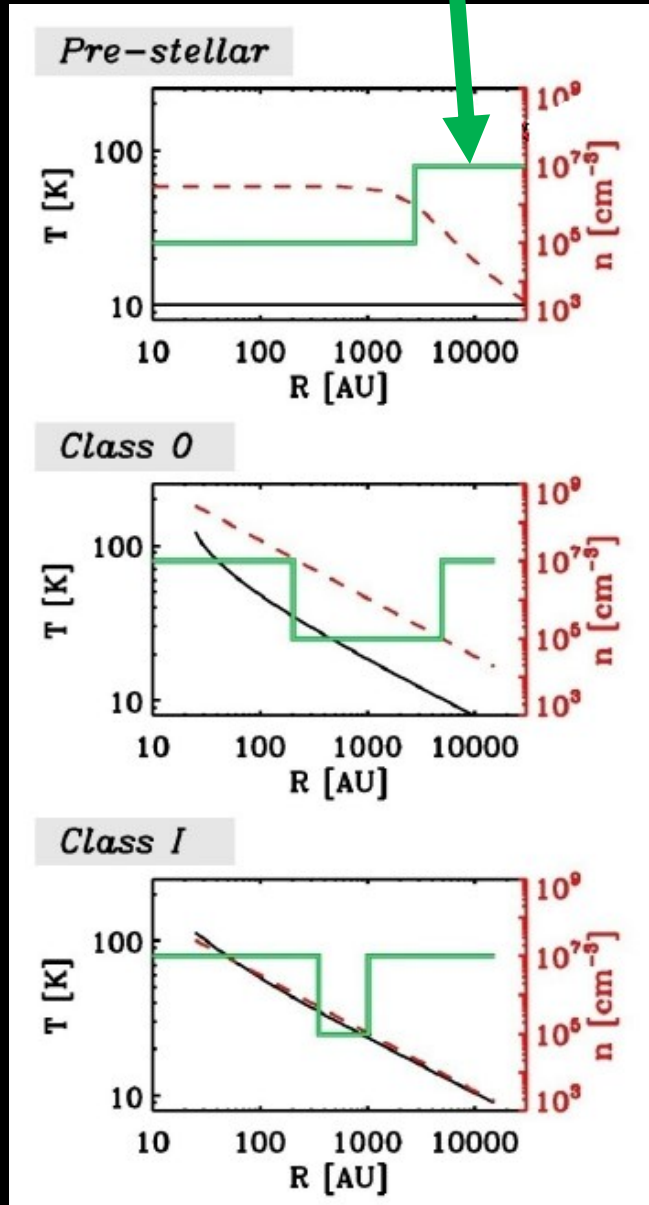


IRAS 16293-2422

Example: CH₃OH jump abundance structure



Abundances



- *Pre-stellar core:*
- Low temperature
- Depletion toward center
- ...but not edge
- *Protostellar core:*
- Central heating $\sim$ temperature gradient
- Thermal desorption
- ...outside (low T): depletion / no depletion regions as in pre-stellar stages

Jørgensen et al. 2005

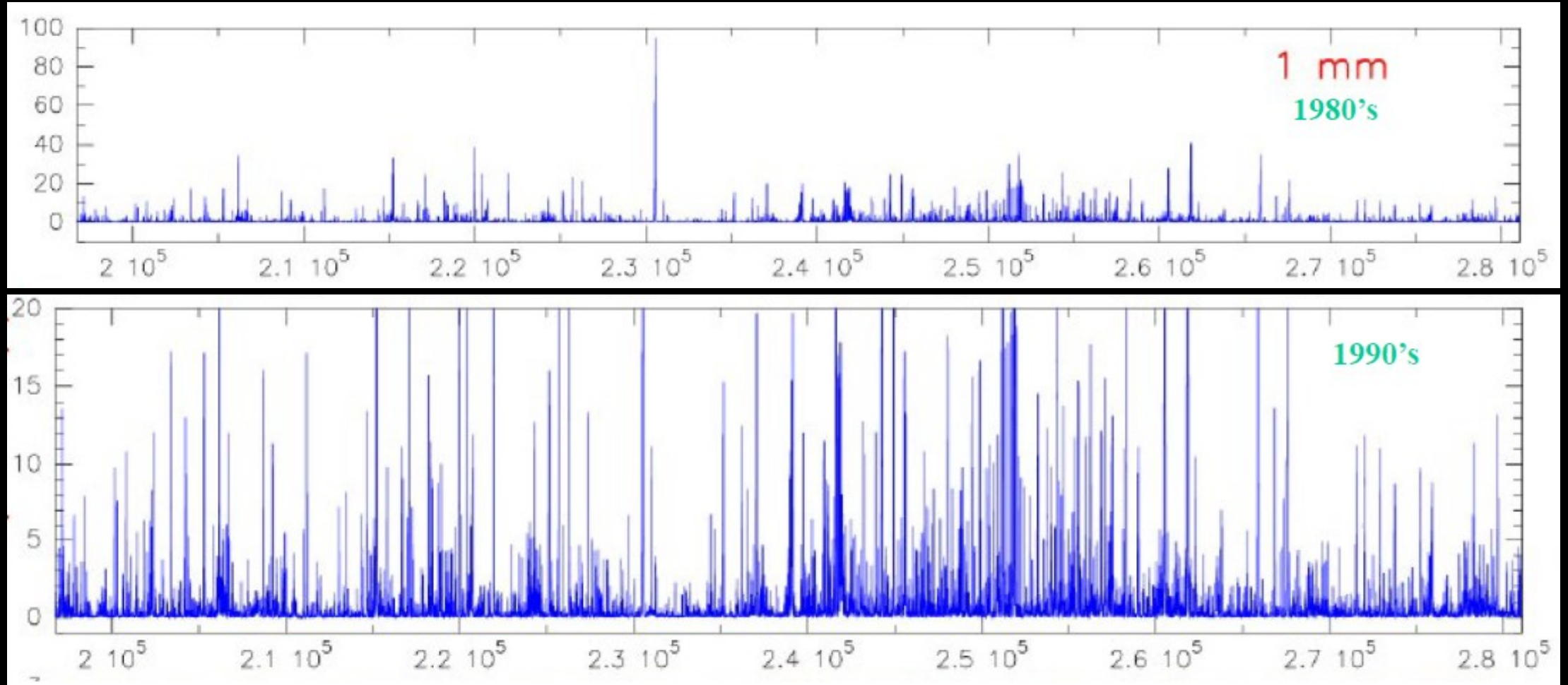
Q: Can chemistry constrain timescales? Current data suggest phase of heavy depletions (pre+ protostellar) lasts only $\sim 10^5$ yr

7.6 Hot cores

- Observations of most high-mass YSOs such as *Orion-KL*, *Sgr B2*... show high abundances of complex saturated organic molecules
- These are called '*hot cores*' with characteristics $T \approx 100\text{-}200\text{ K}$, $n \approx 10^7\text{ cm}^{-3}$, size $< 0.1\text{ pc}$
- Several low-mass YSOs show a similar chemical complexity
- Traditional view: molecules sublimated from grains into warm gas drive a rapid gas-phase chemistry for $\sim 10^5\text{ yr}$
 - Initial conditions: ice abundances, as constrained by observations

Orion line surveys

Antenna Temperature (K)

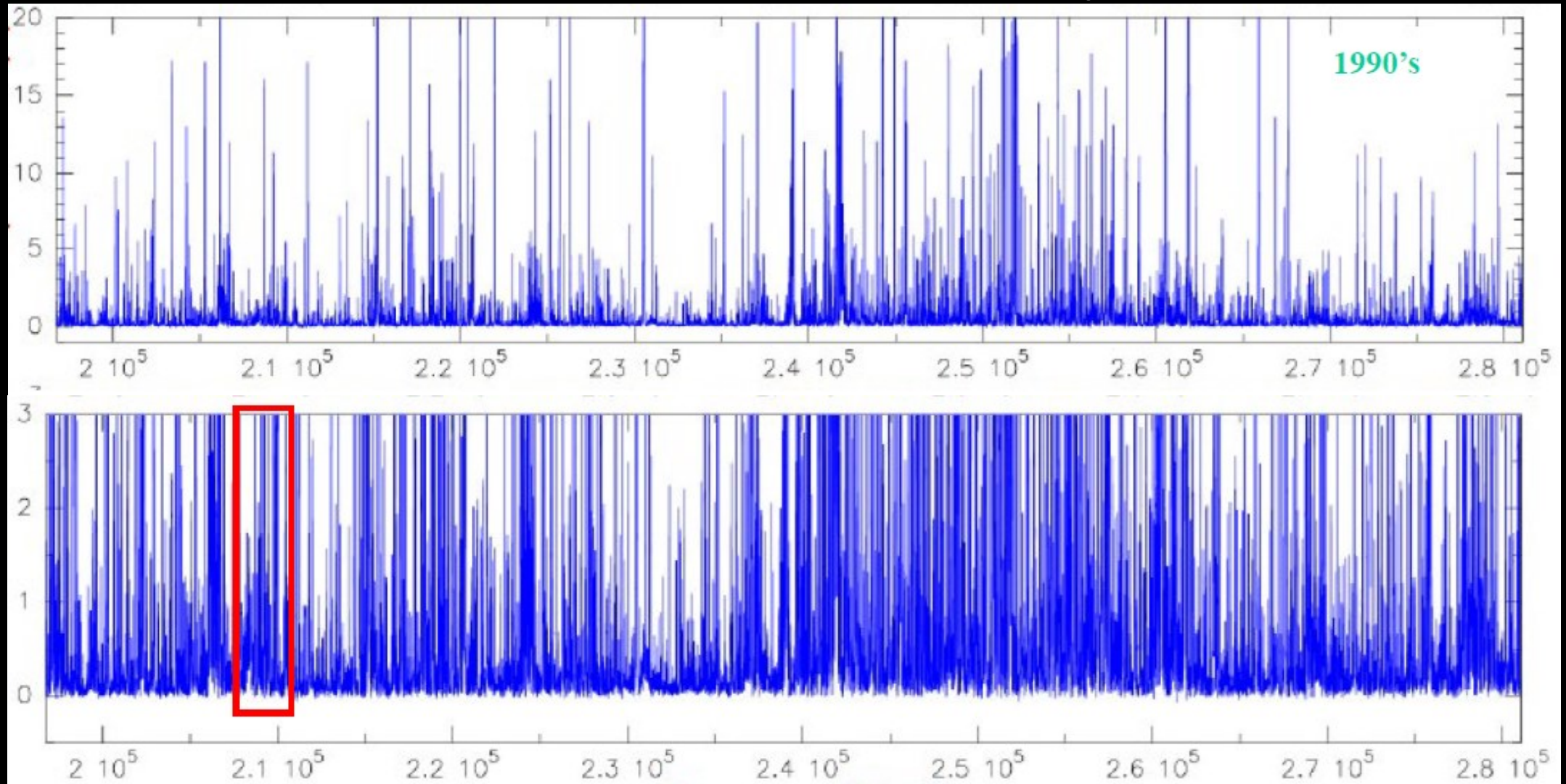


Frequency (MHz)

Tercero & Cernicharo 2007
Blake, Sutton et al. 1985
Schilke et al. 1993, 1997
Crockett et al. 2014 Herschel HIFI

Orion line surveys

Antenna Temperature (K)



Frequency (MHz)

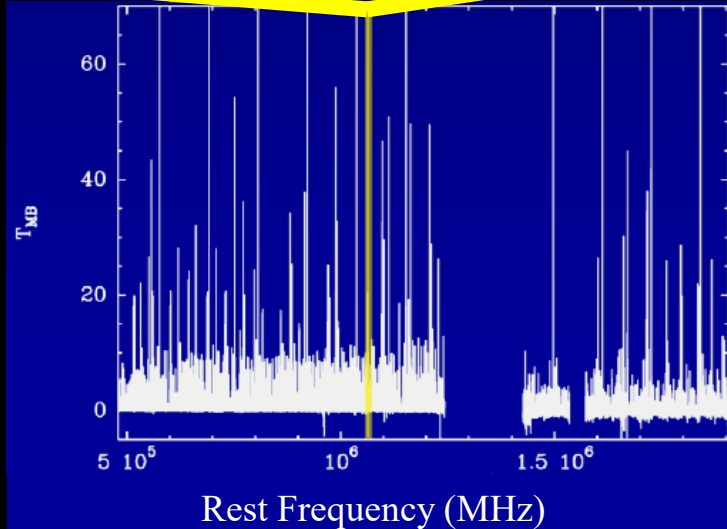
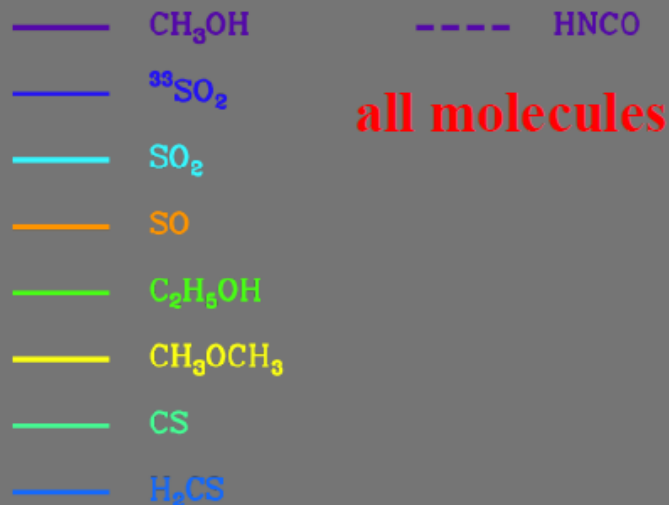
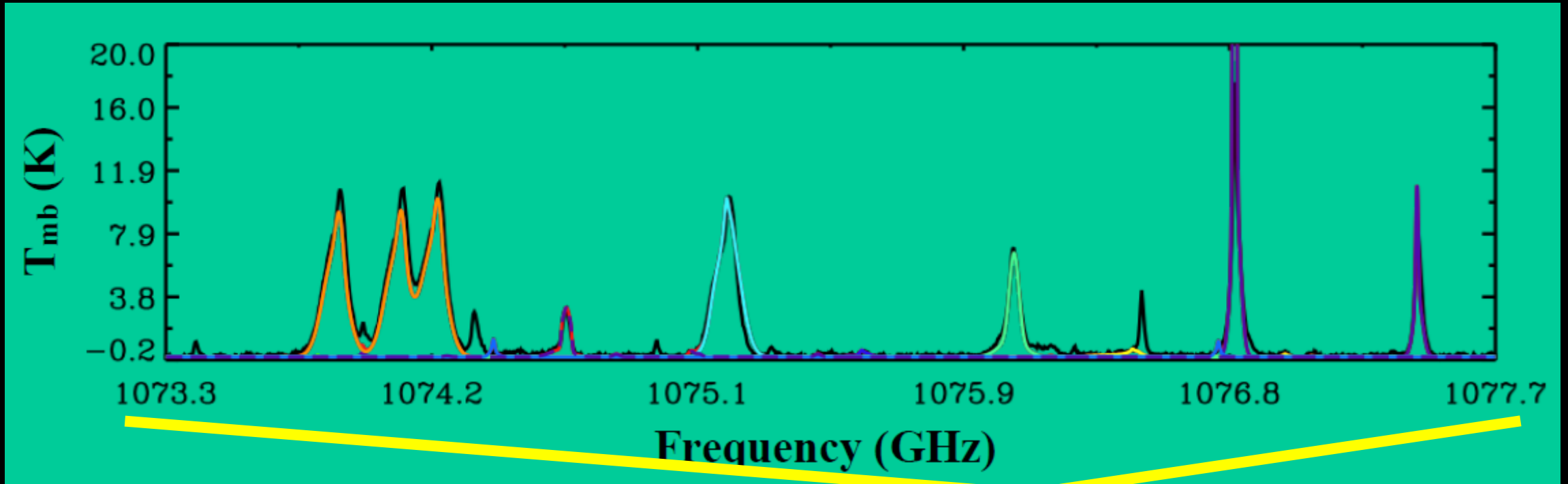
Tercero & Cernicharo 2007

Blake, Sutton et al. 1985

Schilke et al. 1993, 1997

Crockett et al. 2014 Herschel HIFI

Identifying and modeling emission



Crockett, Neill, Bergin et al.
2014, 2015

Solar mass protostars: *The ALMA PILS survey*

Ophiuchus d=125 pc low-mass star-forming region



IRAS16293-2422

Rich source of organic and deuterated molecules (vD+ 1995, Cazaux+03)

NASA/WISE

IRAS 16293-2422 low-mass protobinary star

ALMA: 0.4-3 mm

Source B
Face-on disk

Source A
Inclined disk

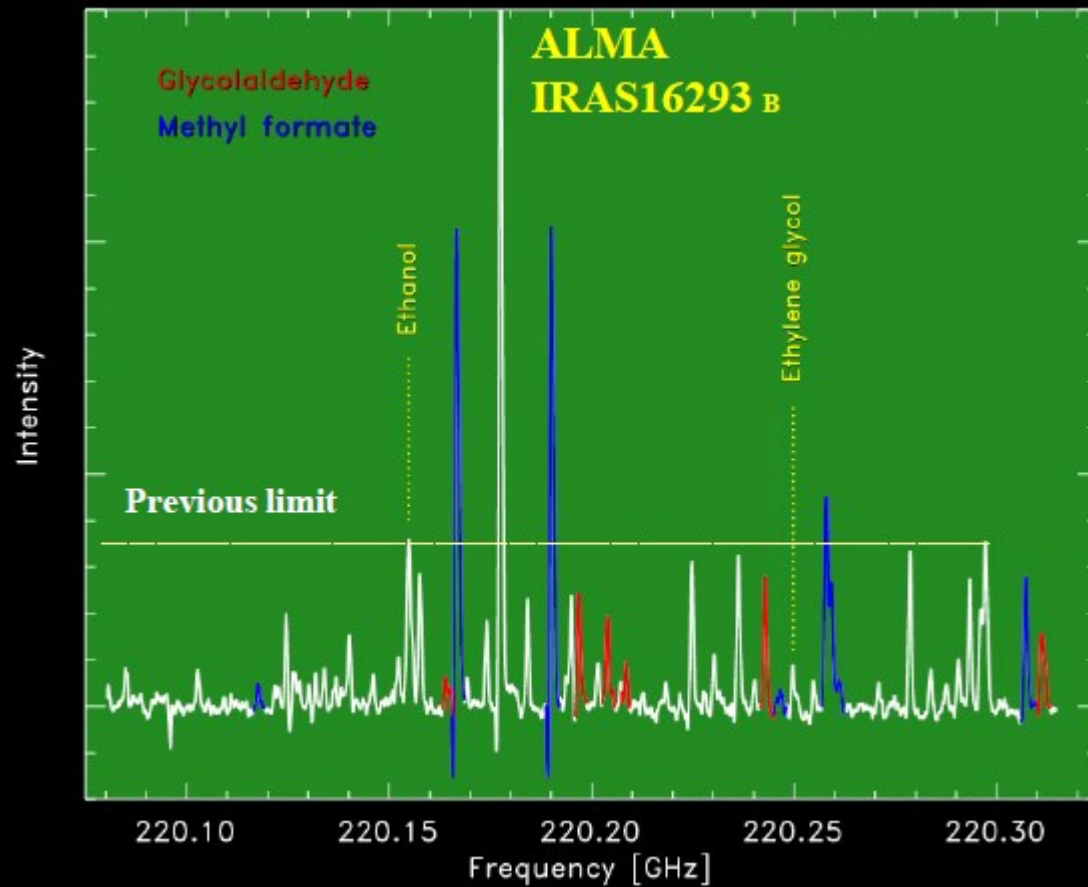
Pineda et al. 2012
Oya et al. 2016

*Protostellar Interferometric
Line Survey (PILS)*

60 AU

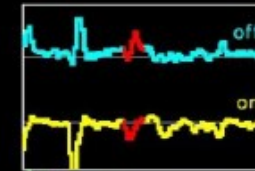
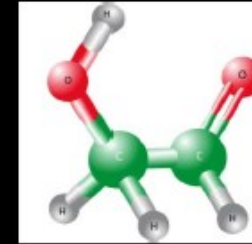


Detection of sugar near low-mass protostar



Detection of glycolaldehyde and ethylene glycol in first ALMA science verification data

Jørgensen et al. 2012



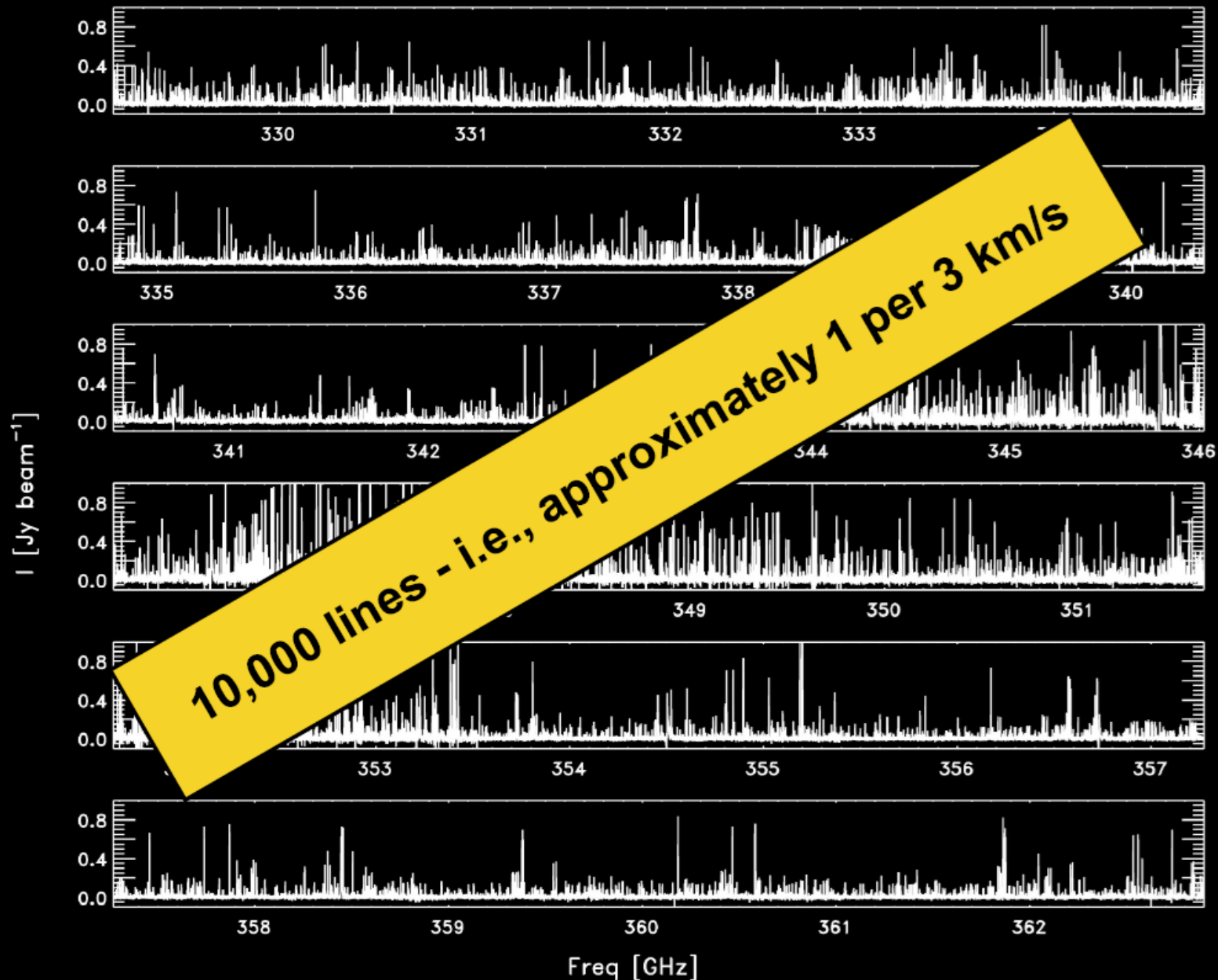
150AU

Band 9, 0.2''

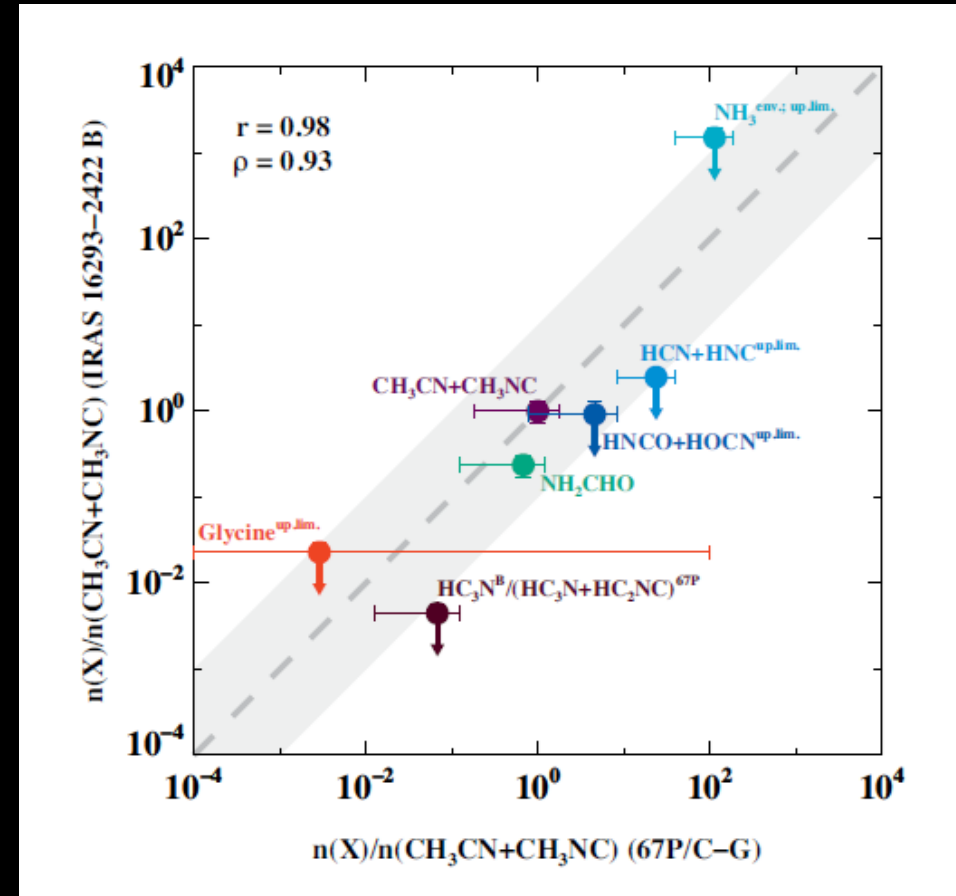
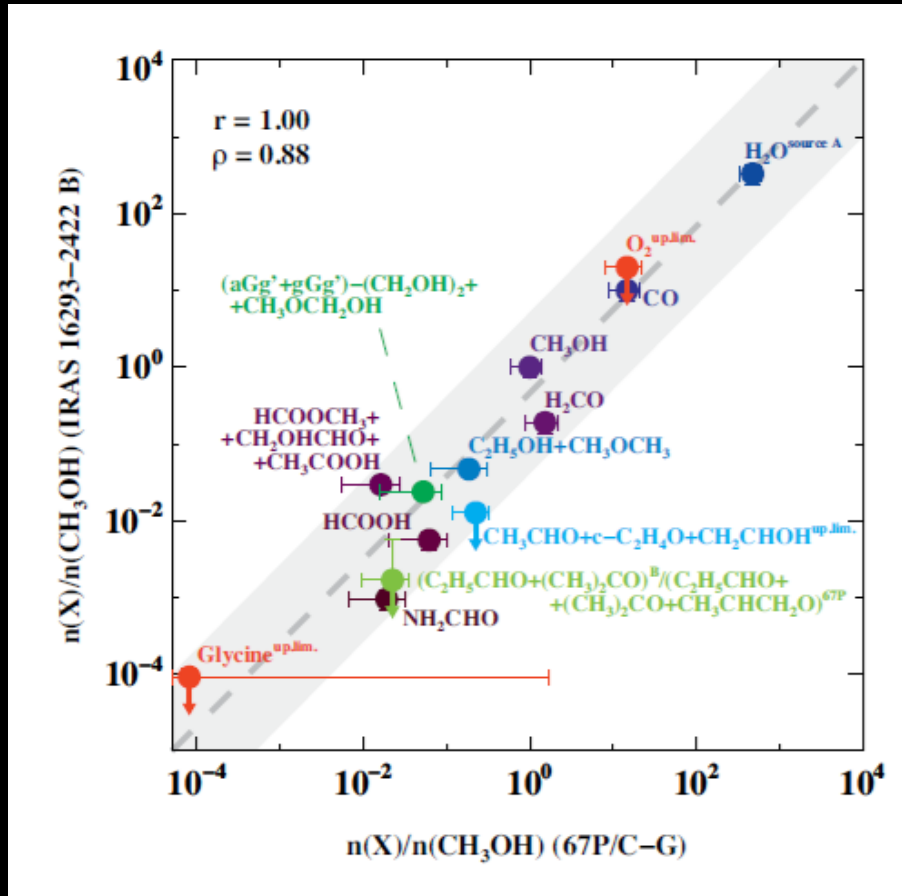


Complex molecules found on solar system scales (orbit of Uranus, 25 AU)

Full spectral survey of IRAS 16293-2422

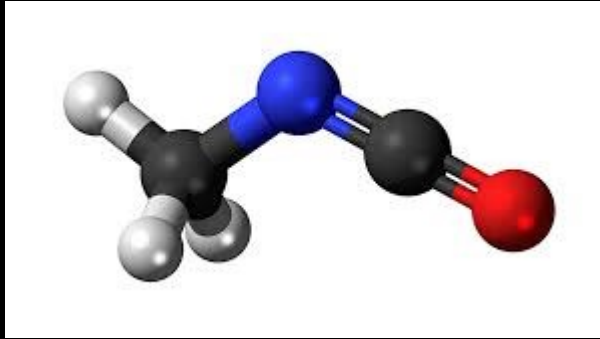


Protostellar chemical fingerprint

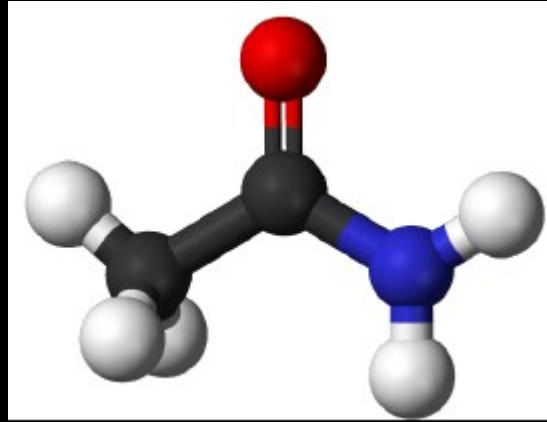


- More than 100 molecular species (including isotopologues)
- Dozens of new detections toward low-mass protostars (so far)
- Correlations with abundances from Comet 67P

Complex molecules on solar system scales

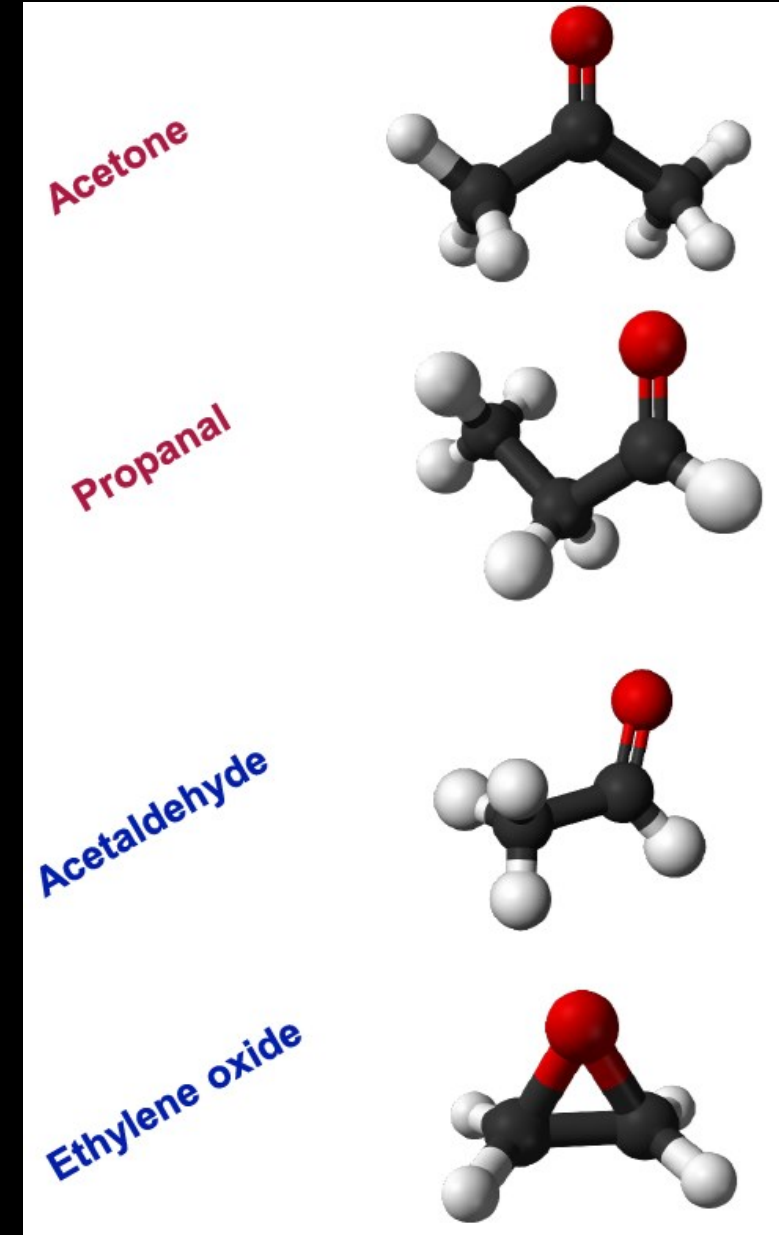


Methyl isocyanate
'Prebiotic' molecule

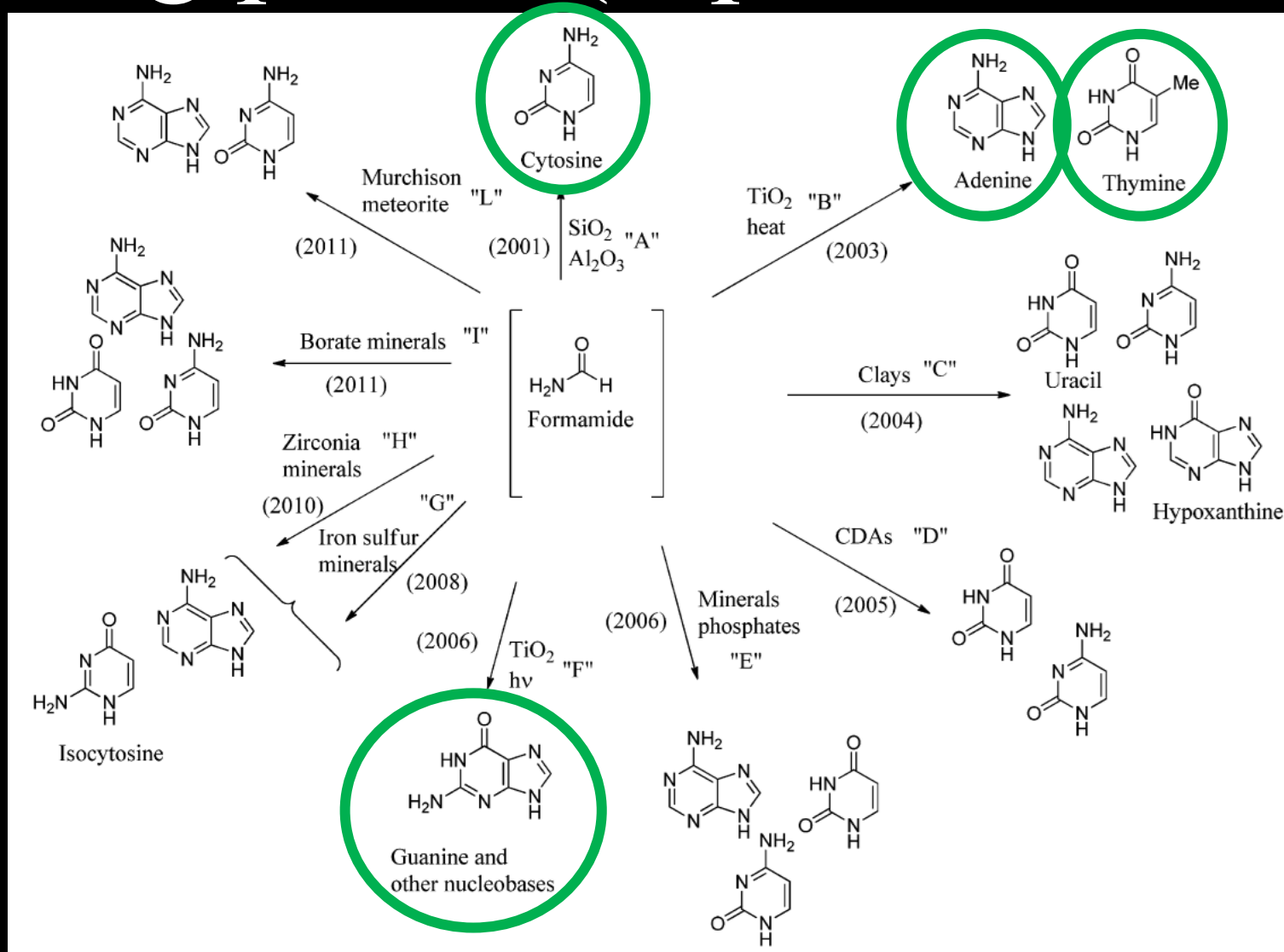


Acetamide
(but no glycine yet)

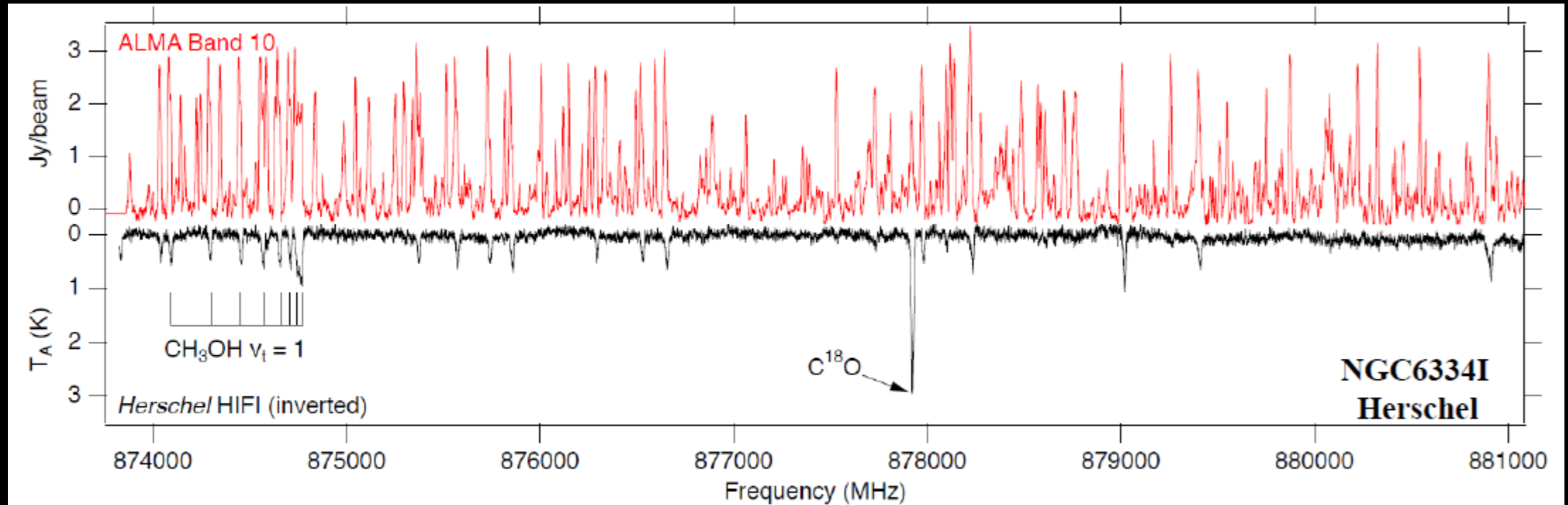
Lykke et al. 2017
Ligterink et al. 2017
Martín-Doménch et al. 2017
Ligterink et al. 2018



From formamide to amino acids and bases on a young planet (liquid water, clays...)



The ALMA revolution: sensitivity to small scale emission

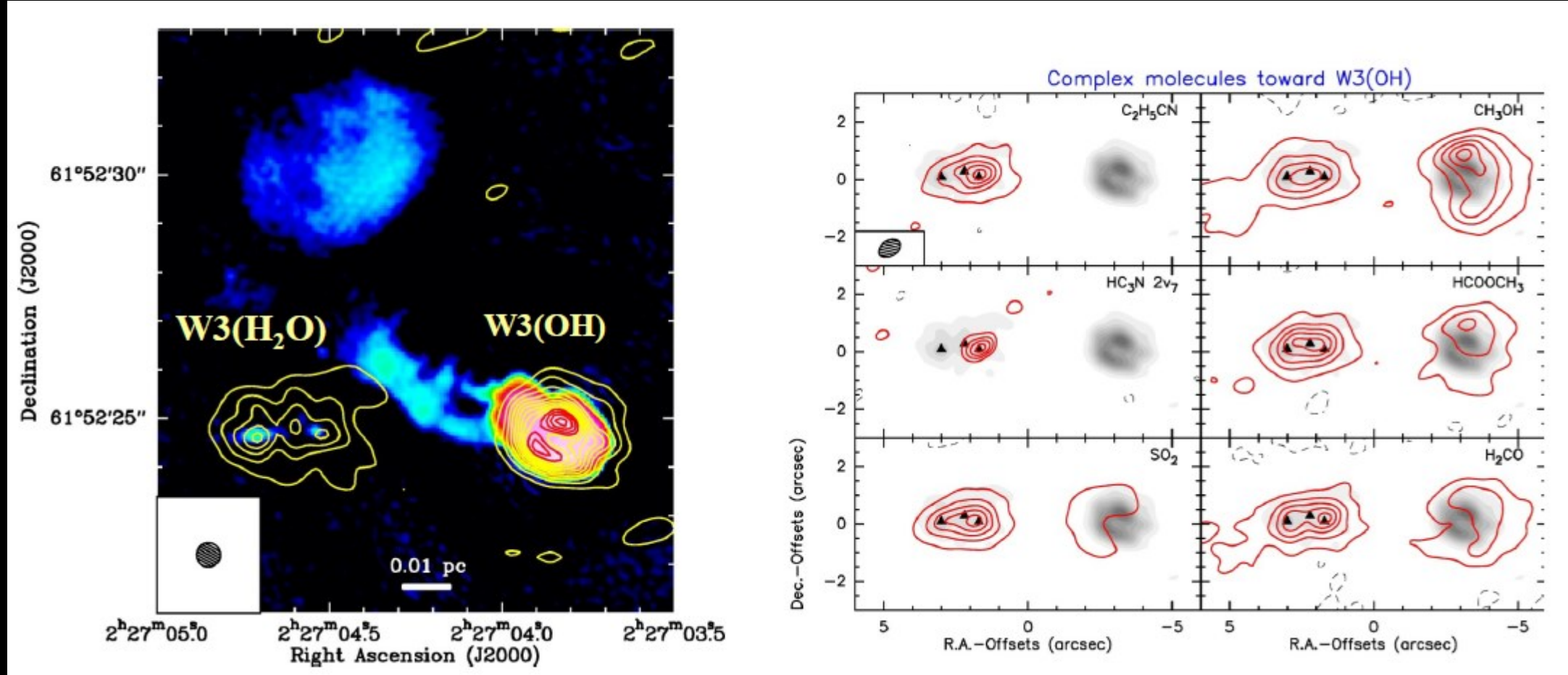


High-mass star-forming region

McGuire et al. 2019
Bogelund et al. 2019

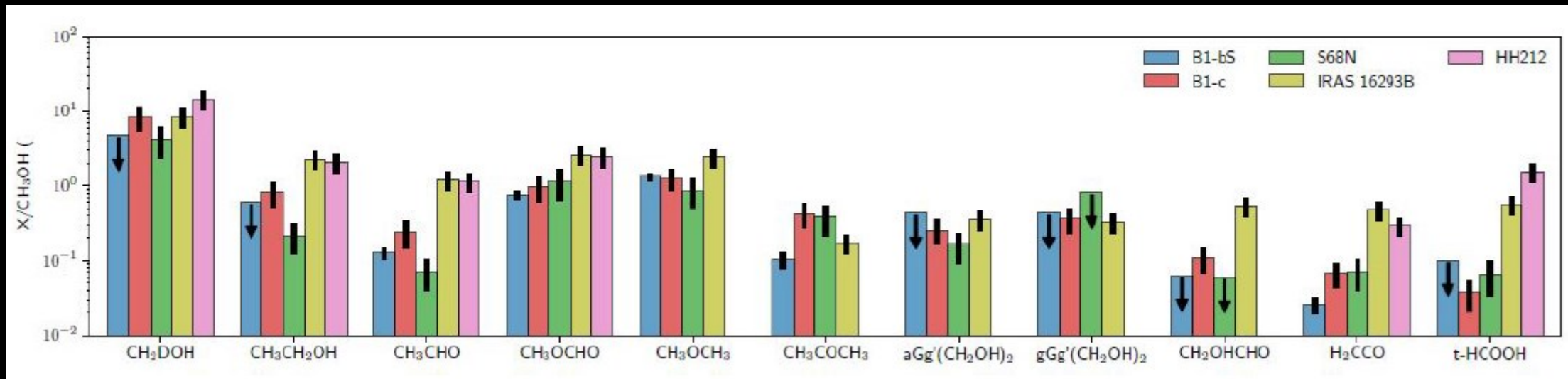
Chemical differentiation on small scales

High-mass protostars

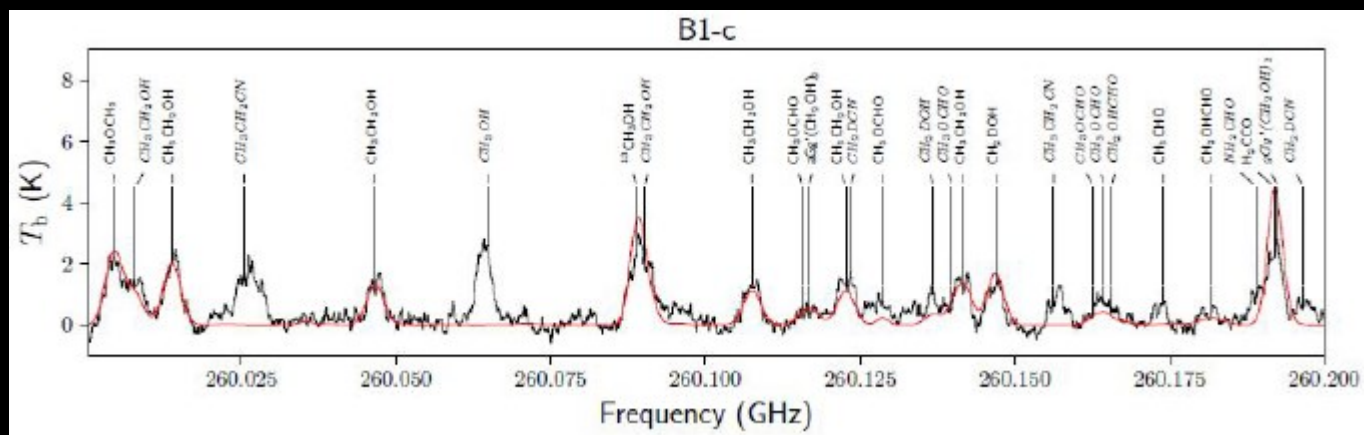


- Chemical differentiation between O- and N-rich complex organics on few arcsec scale (few hundred – few thousand AU) – different ages and formation

Similar COM abundances among low & high mass stars



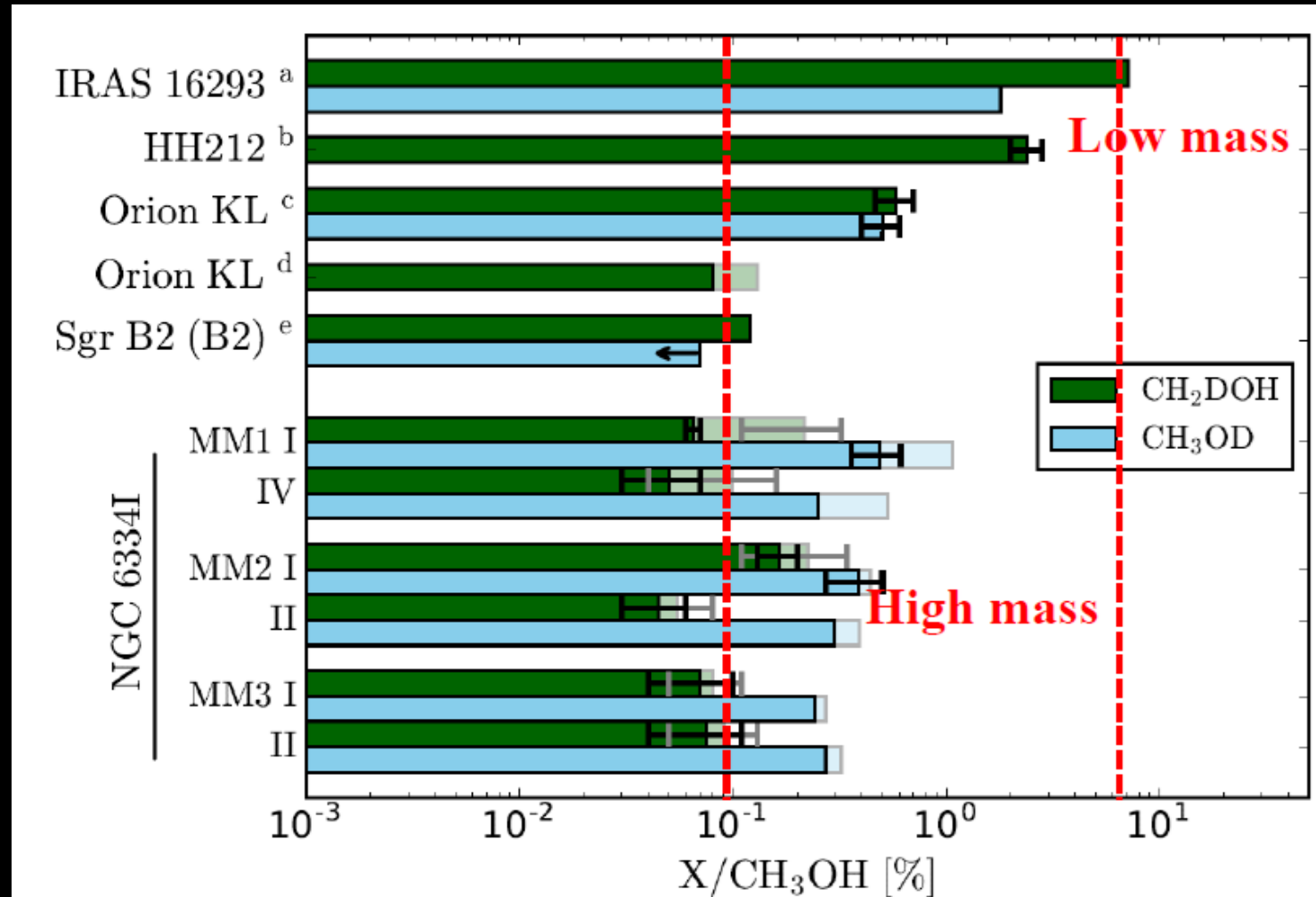
Perseus
Serpens
Ophiuchus
Orion



B1-c Low-mass protostar

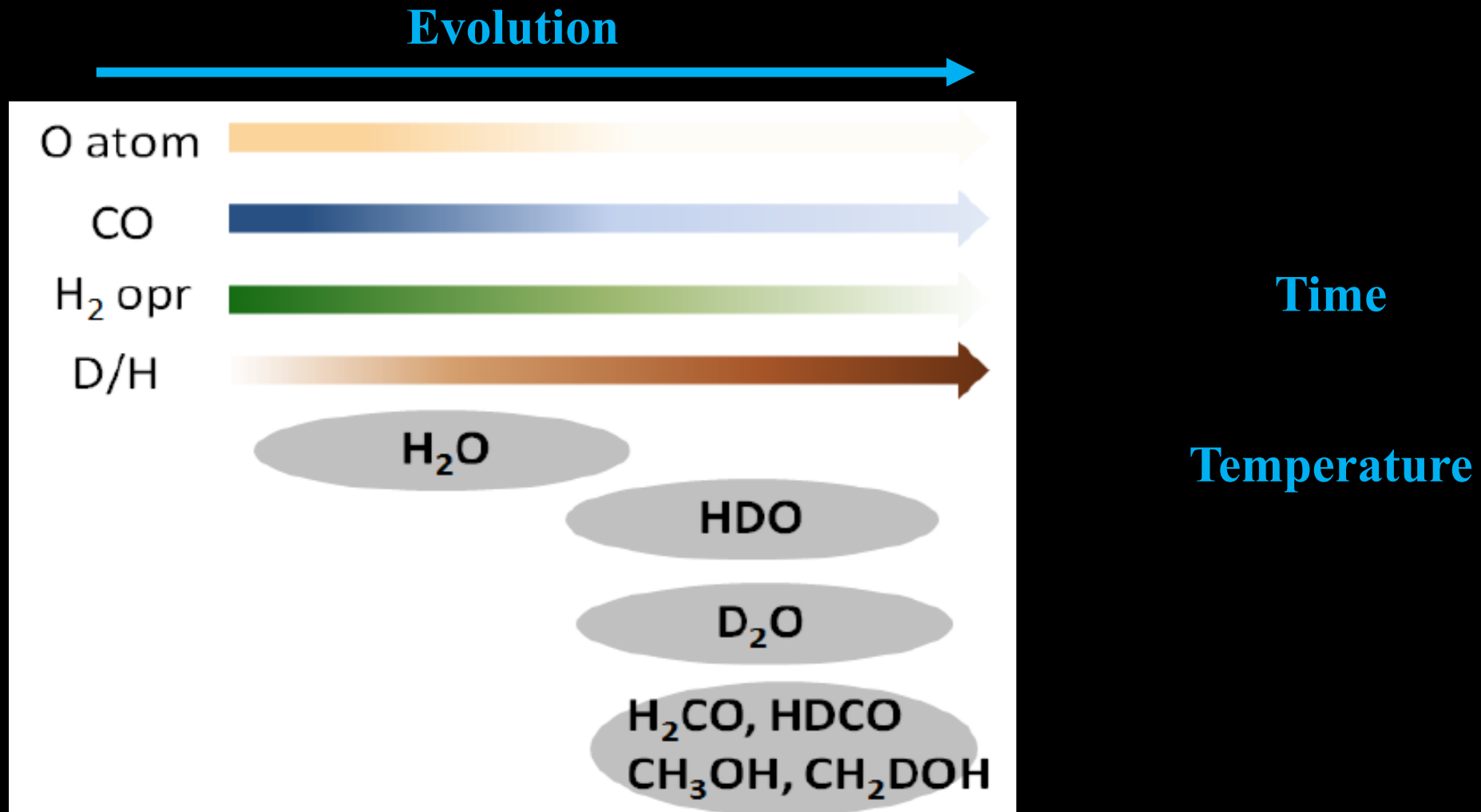
van Gelder et al. 2020
Bogelund et al. 2019

Low vs high mass deuteration



- Much lower deuteration in high mass sources
- Higher temperature effect?

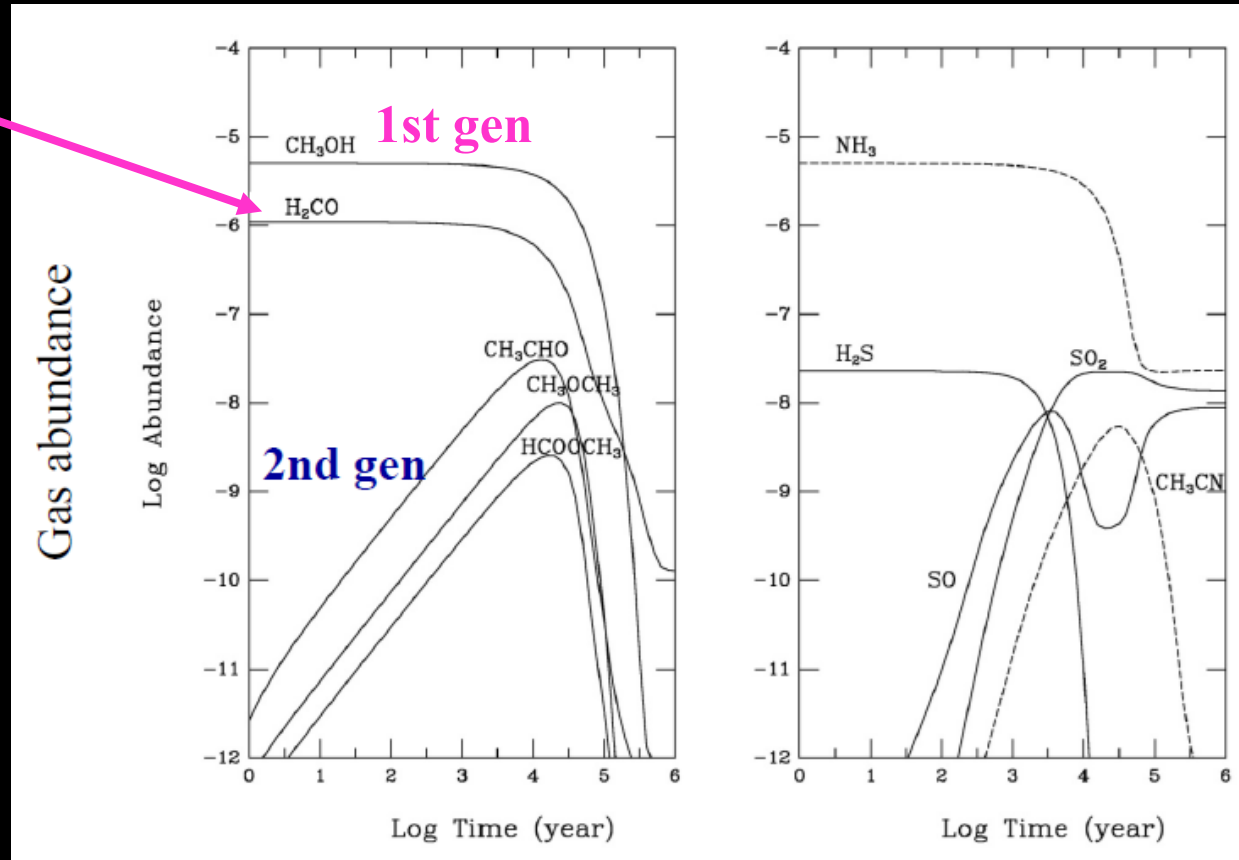
Highly deuterated COMs



- COMs much higher level of deuteration than water

Traditional hot core chemistry

Sublimated from
grains at $t = 0$



Charnley et al. 1992, 1997

- Hot gas-phase chemistry starting at $t = 0$ from sublimated ices
- Which complex organics are 'first generation' made in the ices, and which are 'second generation' produced in the gas?
- Can complex organics, e.g., CH₃OCH₃/CH₃OH be used as chemical clocks?
- In this scenario, differences between sources may be explained by different initial ice composition or different 'age' (evolutionary stage)

New insights

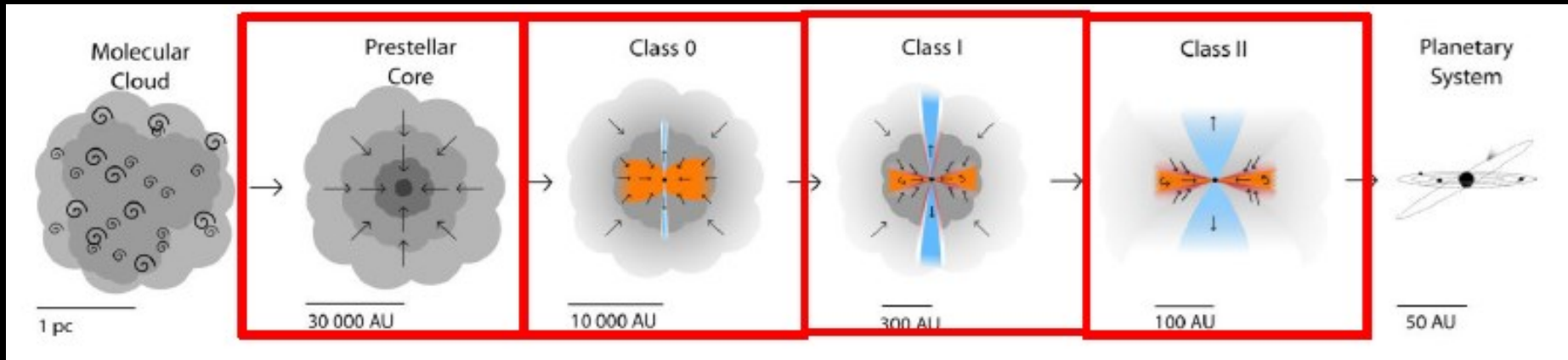
- Hot core gas-phase ‘**second generation**’ models can no longer quantitatively reproduce observations of some complex organics because of new laboratory rates (e.g., DR branching ratios)
- Similarity in abundance *ratios* across wide range of sources requires a mechanism which does not critically depend on specific time or conditions
- Several complex molecules can already be formed in cold clouds at ~ 10 K.
- Ice processing by UV ($\Rightarrow$ produce radicals) at somewhat elevated temperatures (20-40 K) also produces many complex molecules
 - Radicals become mobile at higher T

Origin of complex organics

- *Zeroth generation*: formed on ices in cold (~ 10 K) dark clouds: CH_3OH + more?
- *First generation*: formed in ices by reaction with radicals at slightly elevated temperatures (20-40 K)
 - Needs a little UV to produce radicals, e.g., by cosmic-ray induced photons
- *Second generation*: formed in hot (> 100 K) gas from reactions with sublimated species

Complex organic molecules

- Seen at all stages (pre-ALMA) !
 - Complex molecules found at *all* stages: in pre-stellar cores, protostars, shocks, disks
- New era with ALMA
 - ALMA can *image* each line
 - Sensitivity to solar-mass protostars, not just Orion-like



Forming complex organics on cold grains

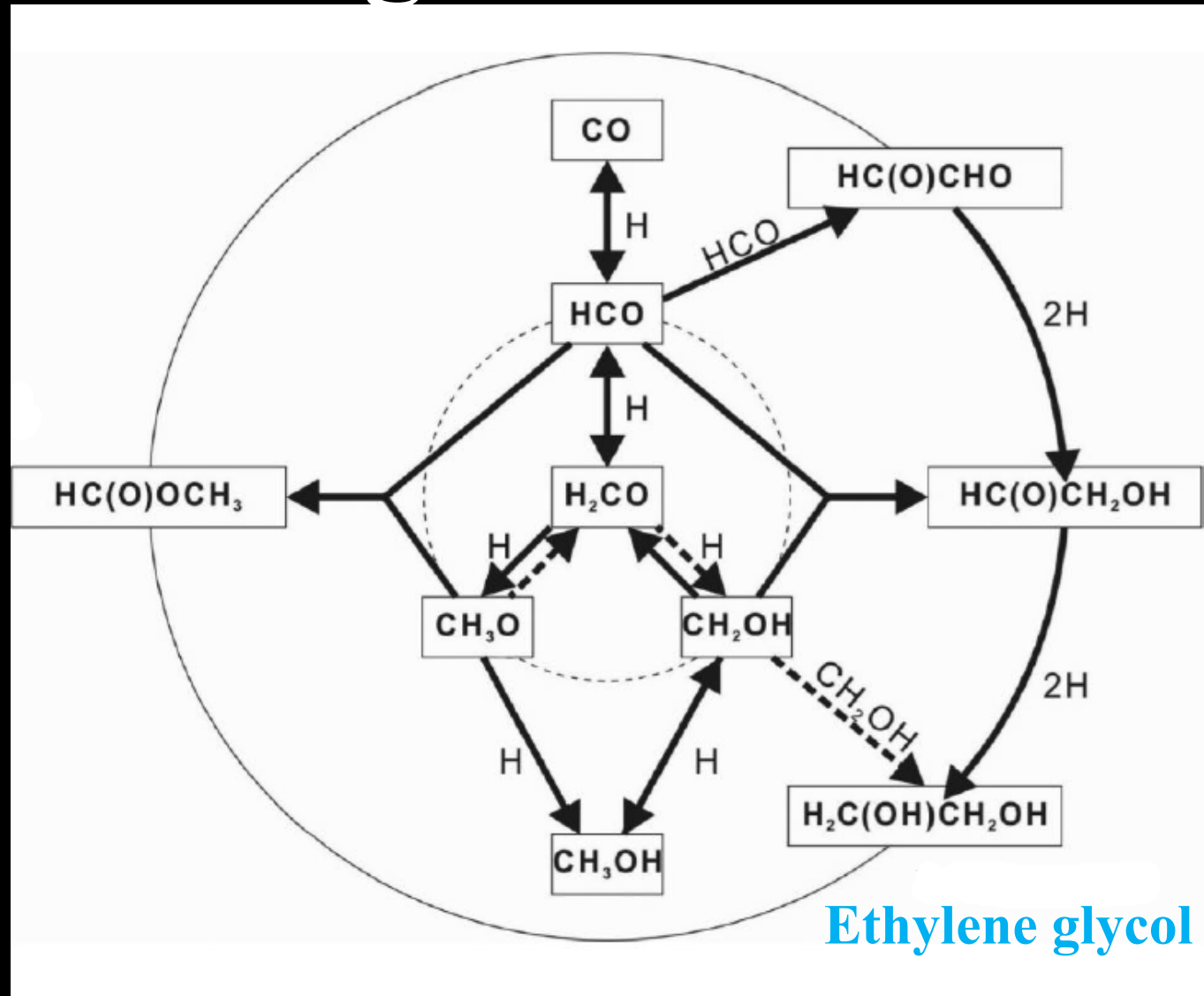


Watanabe et al. 2005
Fuchs et al. 2009
Öberg et al. 2009
Fedoseev et al. 2014, 2015

- CO freeze-out in dense cores $\rightarrow$ HCO $\rightarrow$... $\rightarrow$ CH₃OH
- Radical – radical reactions at low T
- UV-induced chemistry
- Heating by protostar $\rightarrow$ Thermal desorption $\rightarrow$ Observe with ALMA

Making COMs at low T

Methylformate



Glycolaldehyde

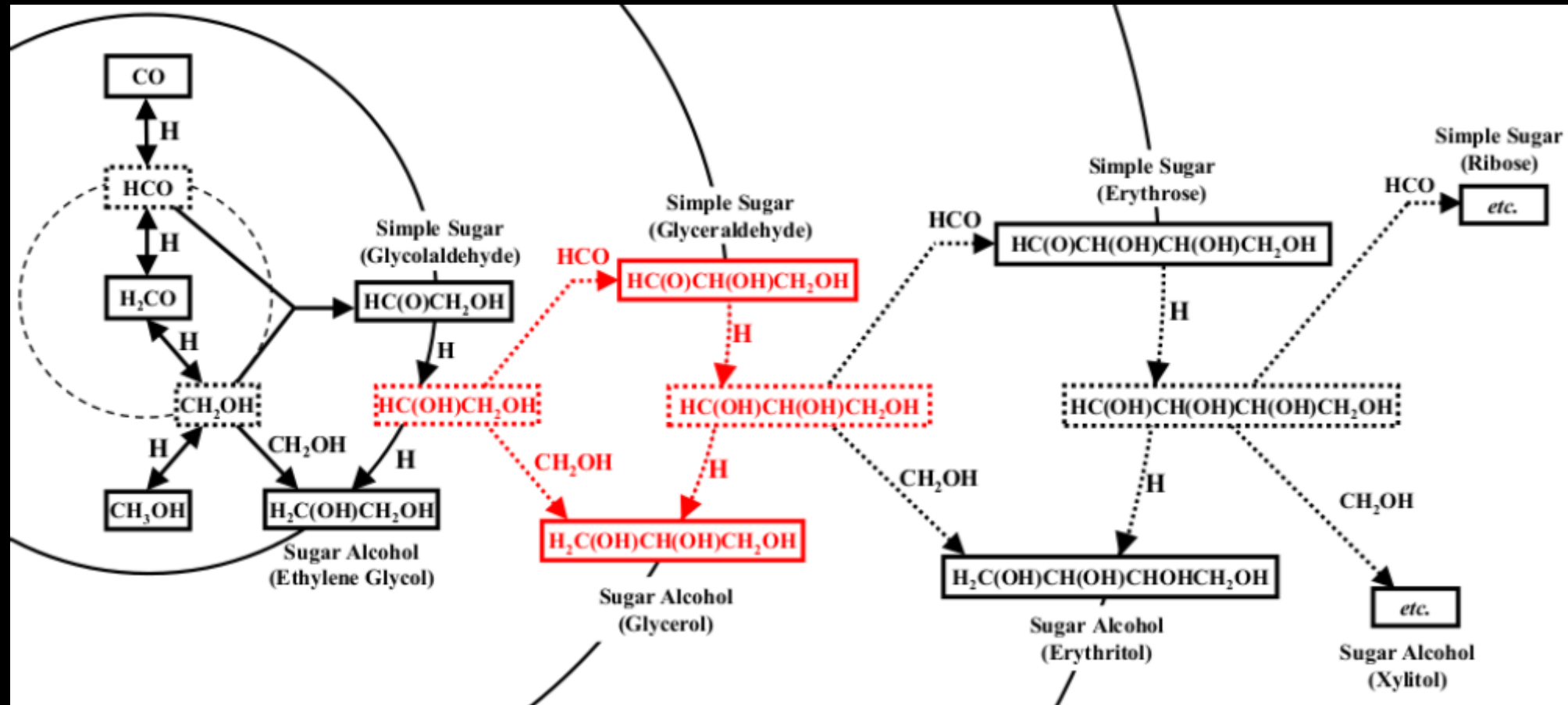
Ethylene glycol

Fedoseev et al. 2015
Chuang et al. 2016, 2017

- Reactions proceed at 15 K, even without need for UV

How far can complexity go?

Starting from CO hydrogenation

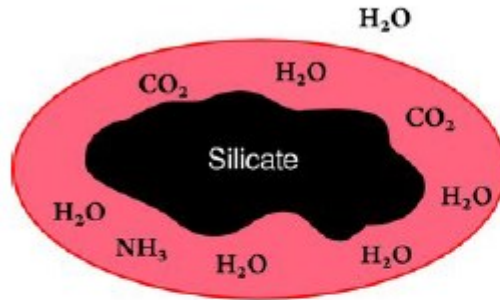


- Can even make glycerol and real sugars without heating or UV!*

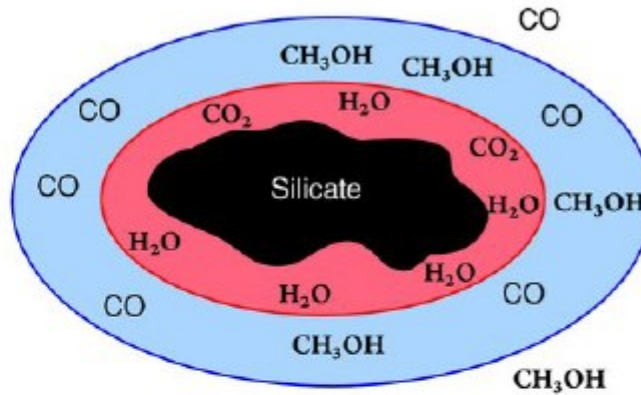
Complex ice chemistry

0th generation

(a) Early ice formation



(b) Cloud core ice formation

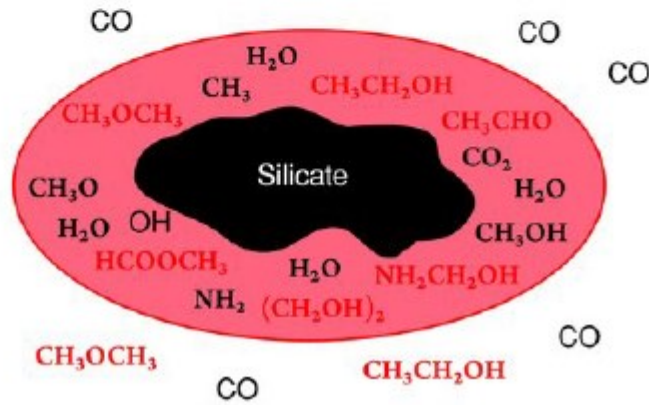


(c) Cold (<20 K) UV processing

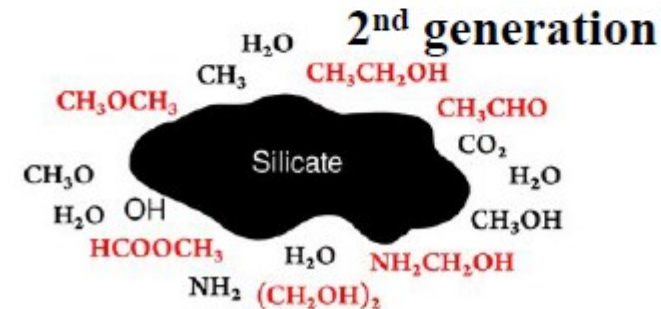


1st generation

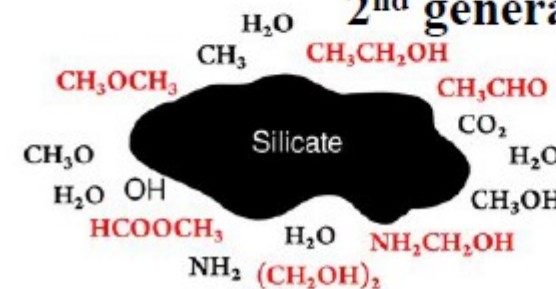
(d) Lukewarm protostellar envelope >20 K



(e) Protostellar hot core >100 K

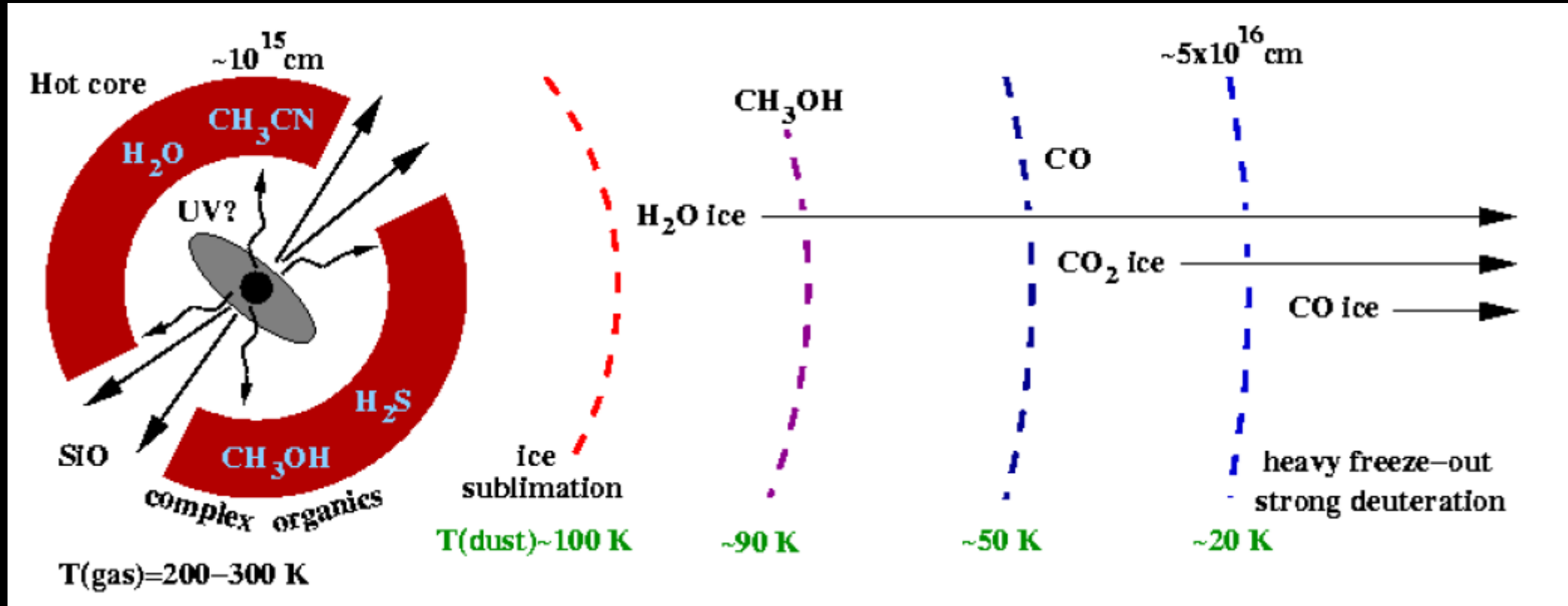


2nd generation



- Differences in 0th and 1st generation COMs no longer so clear

Chemical structure of envelope



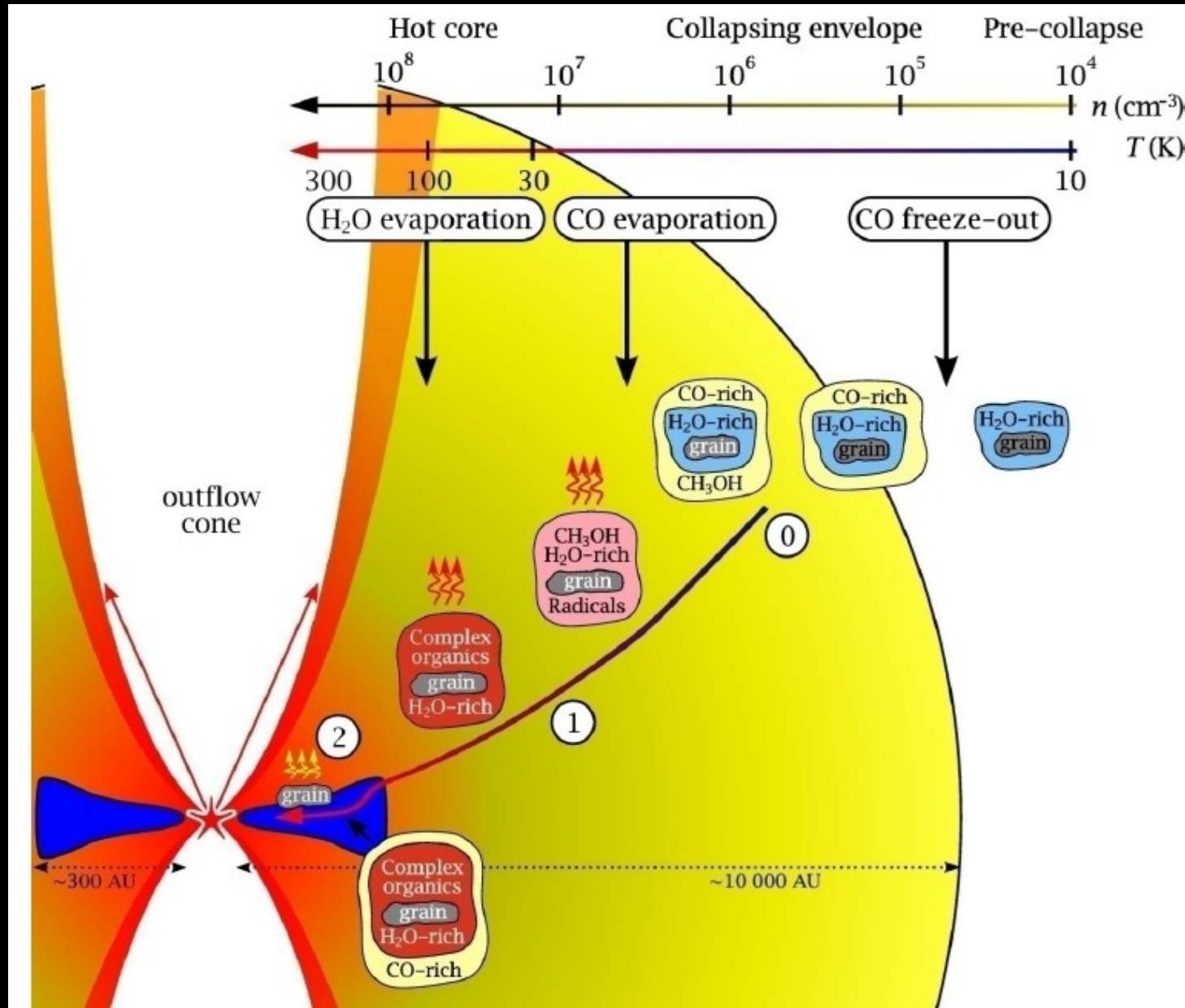
2nd generation

0th-1st generation

Chemical scenario

- Heavy freeze-out of molecules onto grains in cold pre-stellar phase
- Grain-surface reaction produce new species
- Protostar heats surroundings
 - Ice sublimation
 - Hot core chemistry
- Fraction of ices and gas ends up in disks; remainder dispersed

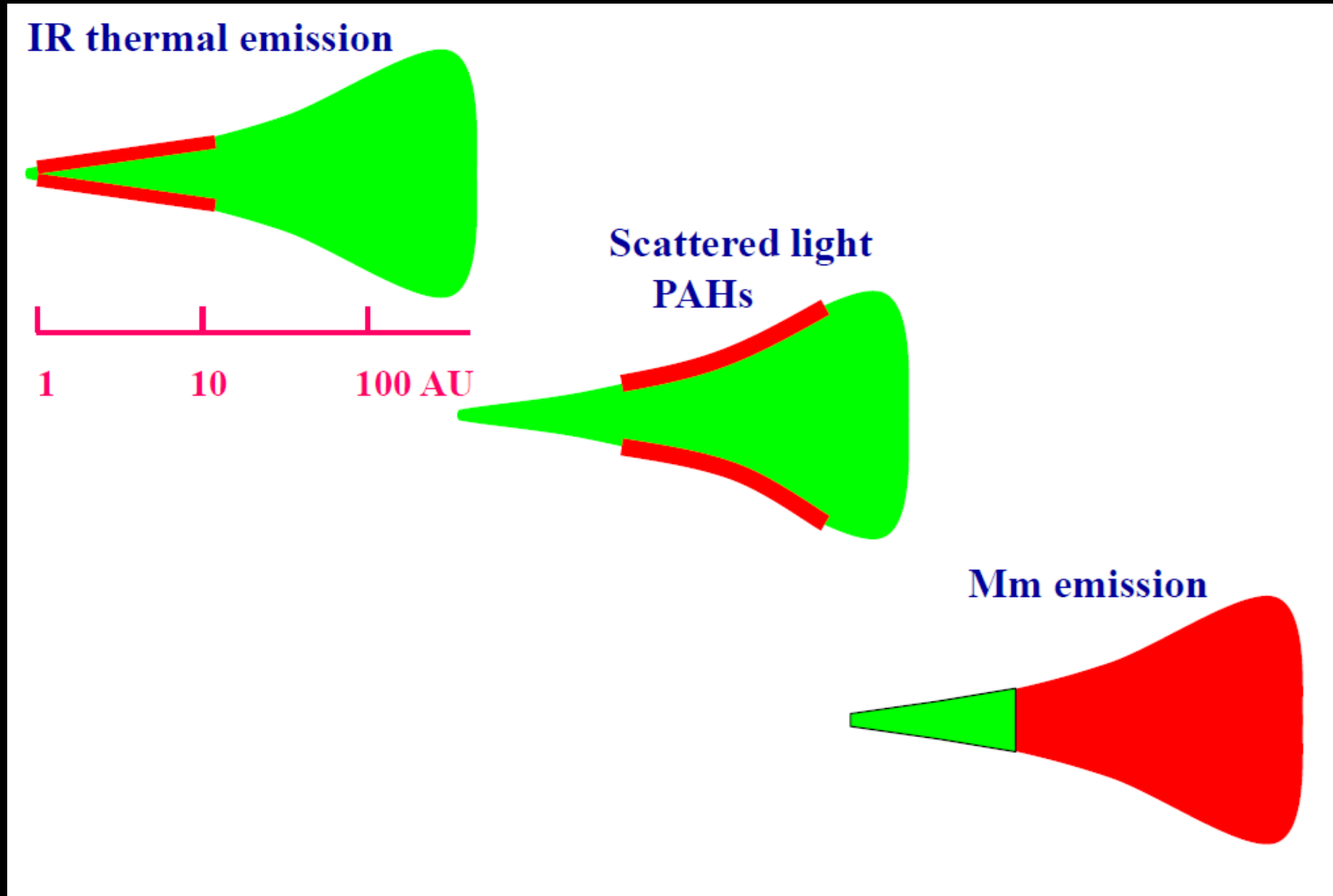
Follow journey of parcel from cores to disk



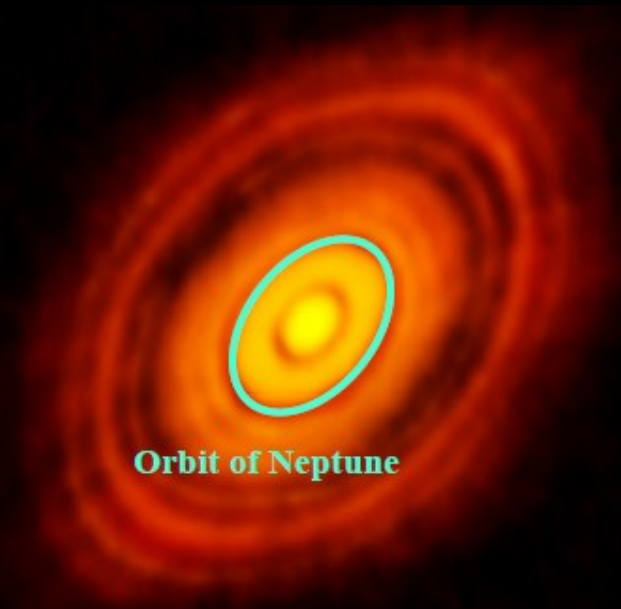
Herbst & van Dishoeck 2009

Visser et al. 2009, 2011

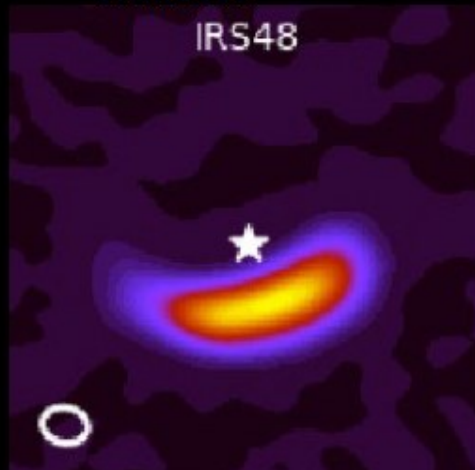
7.7 Outer disk chemistry – probing different parts of disks



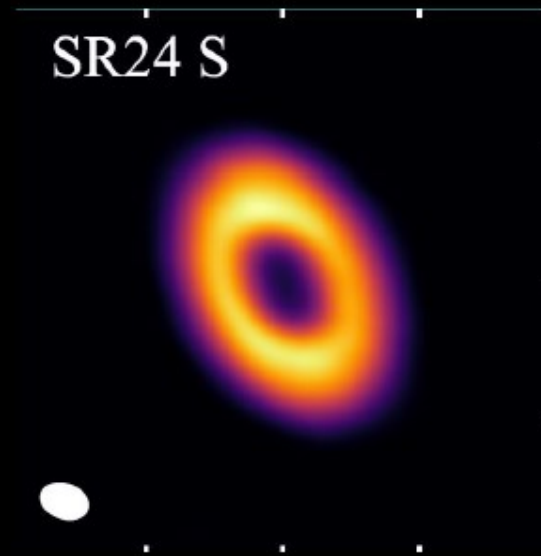
New era of observational planet formation



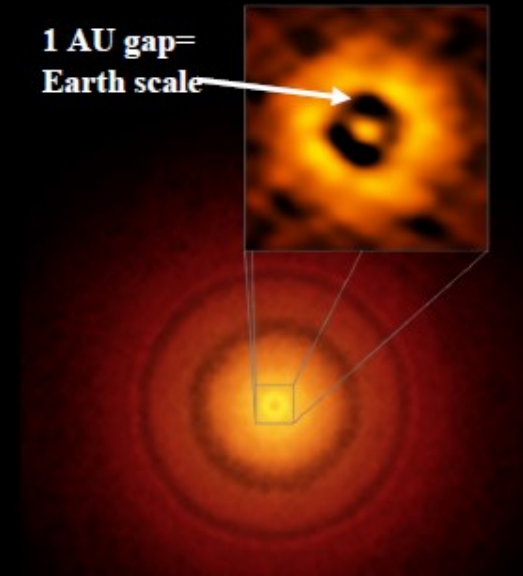
HL Tau young disk
ALMA partnership
et al. 2015



ALMA: van der Marel et al. 2013, 2016



ALMA: Pinilla et al. 2017



ALMA TW Hya
Andrews et al. 2016

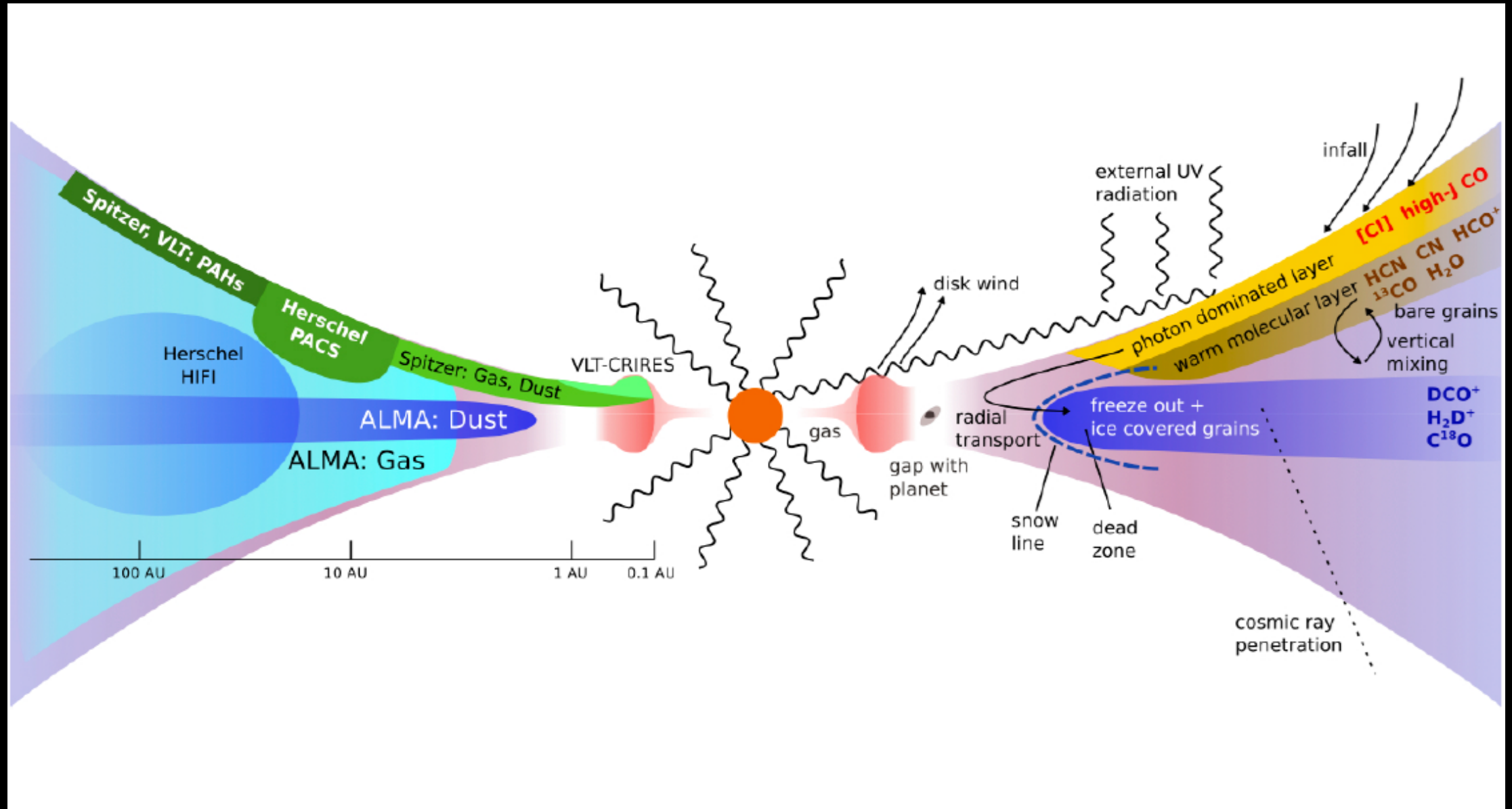


VLT-Sphere, Gemini-GP
Stolker et al. 2016



ALMA:
Fedele et al. 2017

IR inner disk, mm outer disk

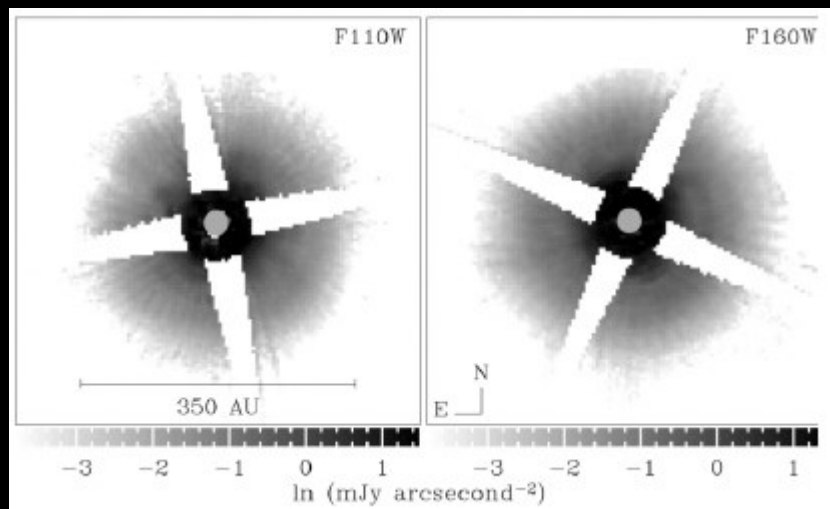


Chemistry in outer disks: mm data

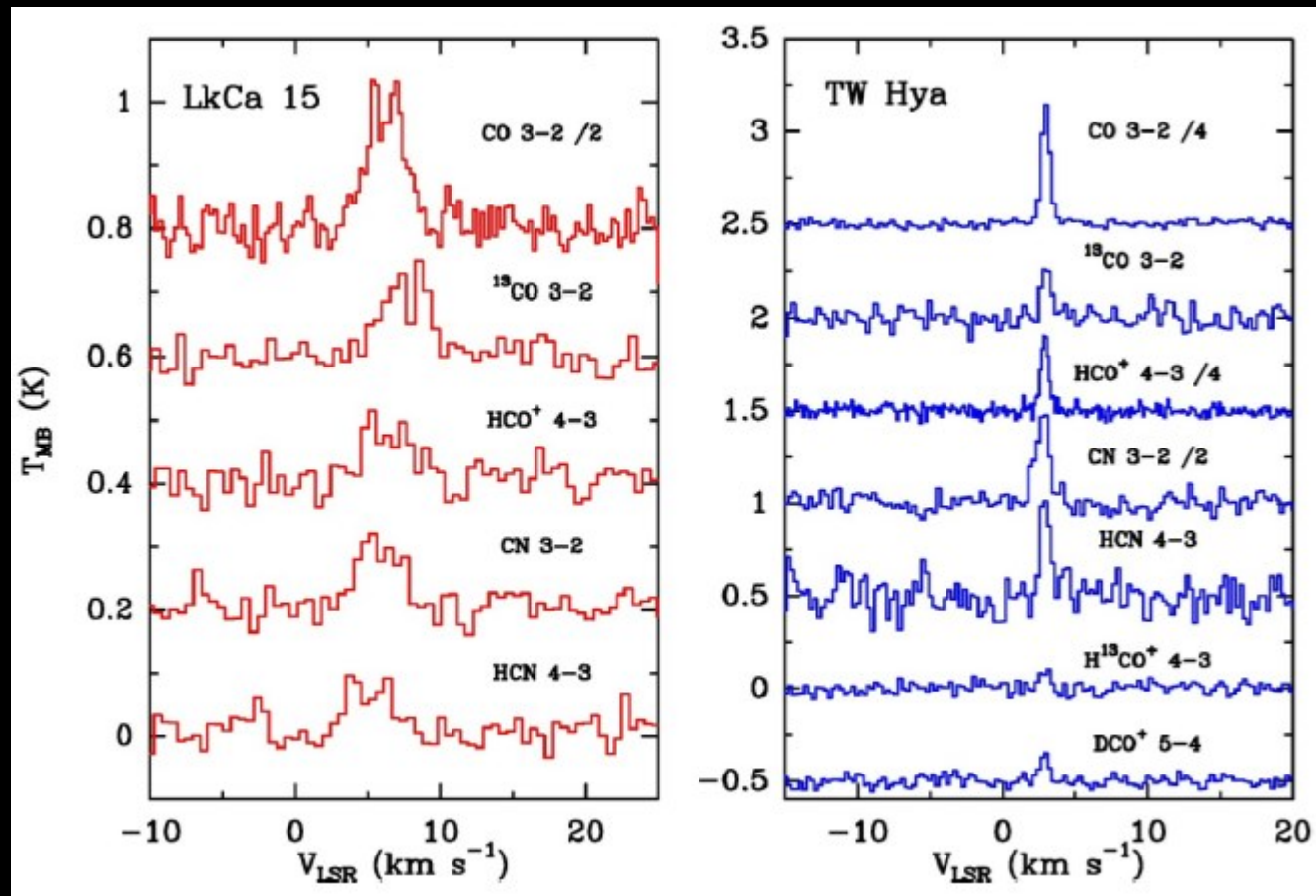
TW Hya face-on disk

1.1 μm

1.6 μm



Scattered light $\Rightarrow$ radius 200 AU



van Dishoeck et al. 2003

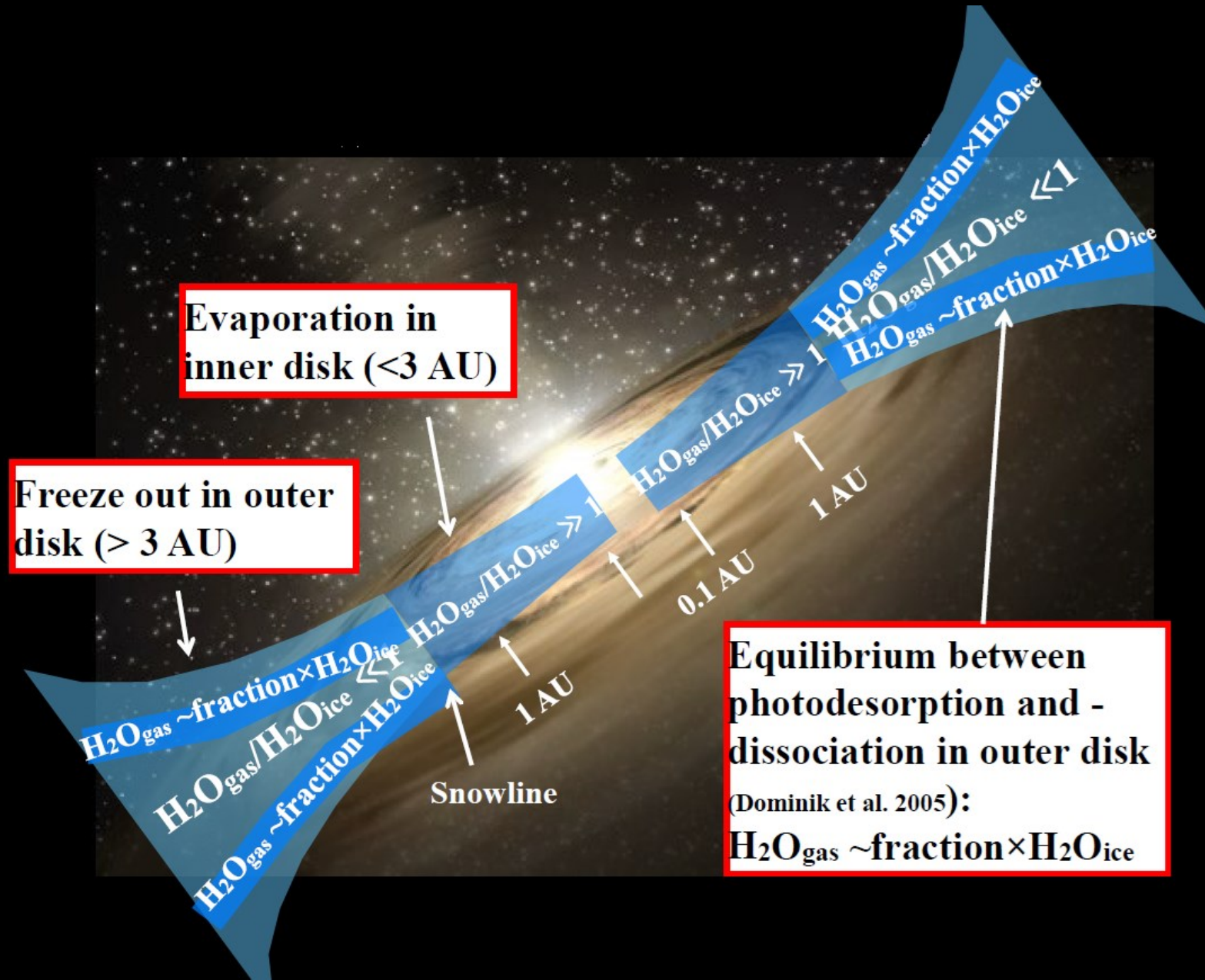
Thi et al. 2004

Kastner et al. 1997

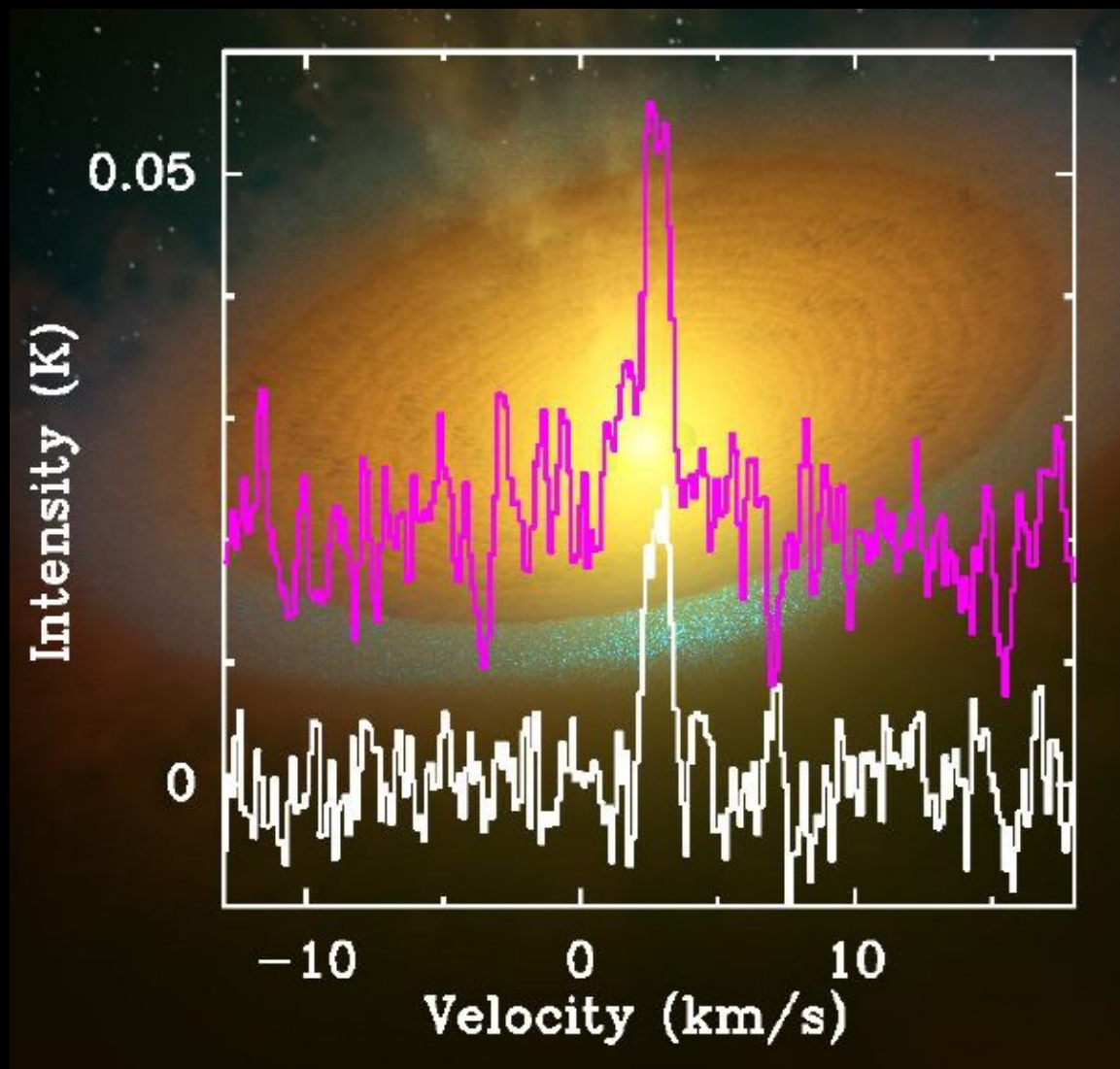
Weinberger et al. 2002

- Small molecules detected, including deuterated species
- Instruments prior to ALMA did not have sensitivity to search for complex molecules

Snowlines in disks



Cold H₂O reservoir in disks



TW Hya
Herschel-HIFI

p-H₂O 1₁₁-0₀₀
1113 GHz

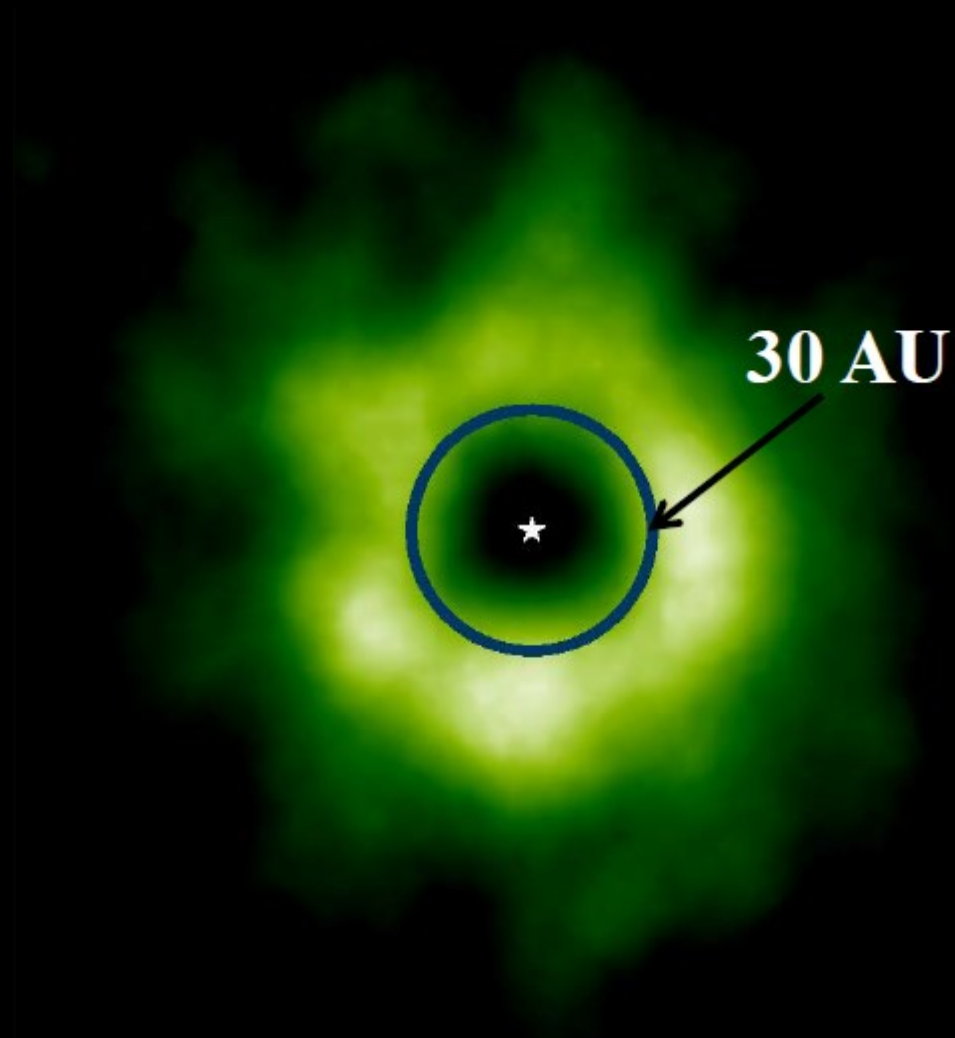
o-H₂O 1₁₀-1₀₁
557 GHz

Elevated o/p ratio in H₂O
compared to comets

Hogerheijde et al. 2011

6000 oceans of water ice in disk

Imaging the CO snowline



TW Hya

Face-on disk

d = 68 pc

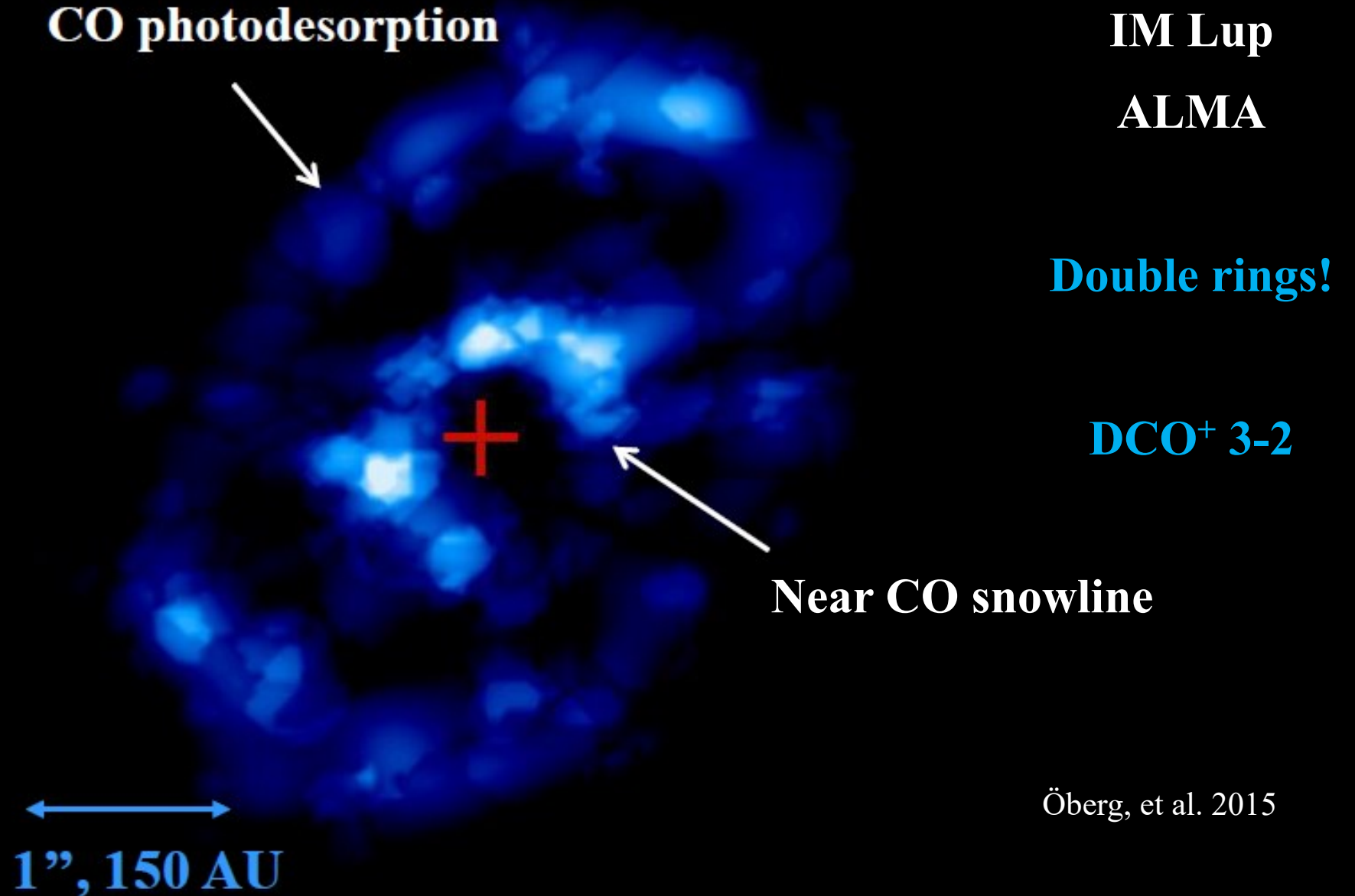
N_2H^+ 4-3

Qi, Öberg, et al. 2013

N_2H^+ appears when CO freezes out

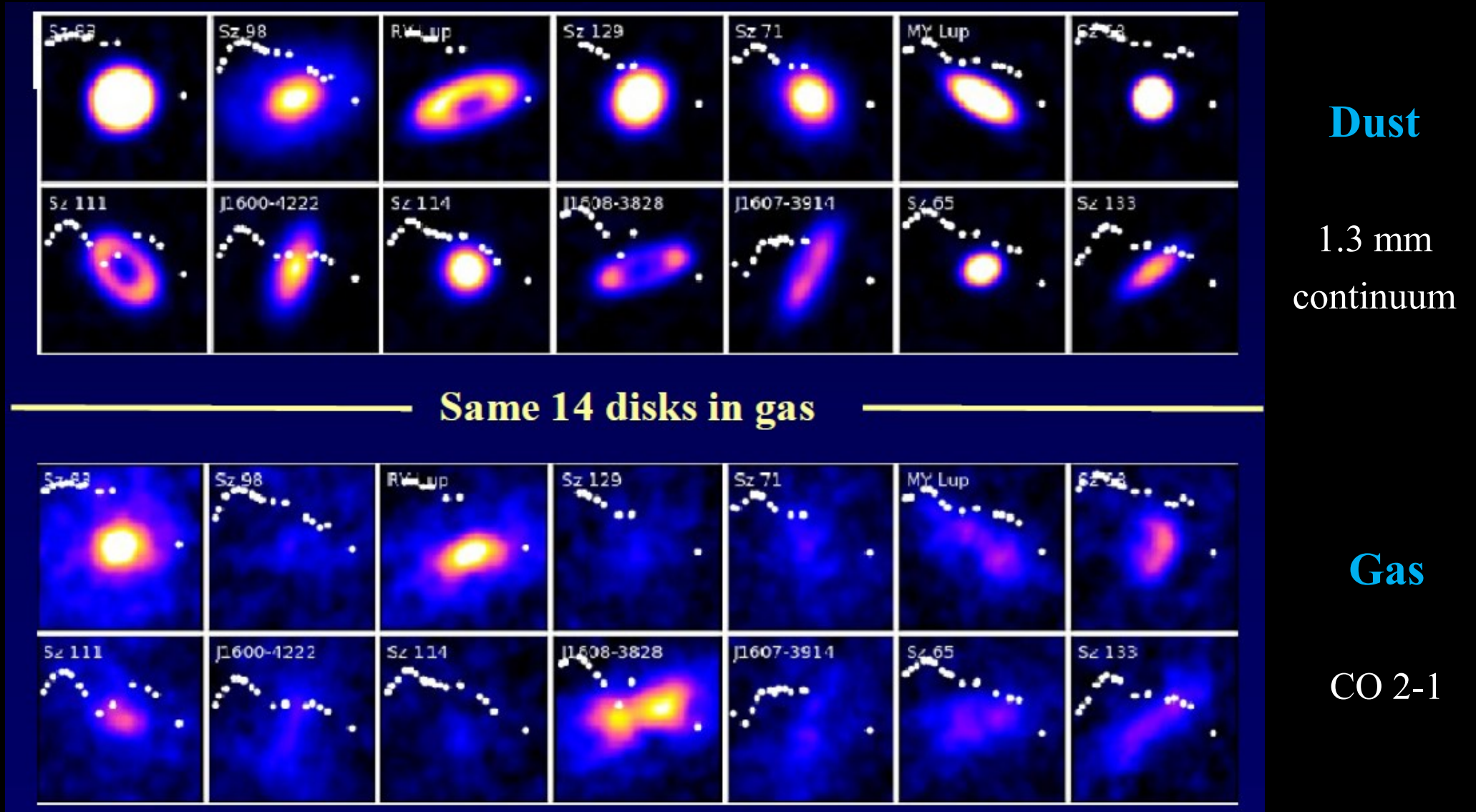
→ Tracer of snowline

Tracing the CO snowline



Öberg, et al. 2015

Puzzle: CO much weaker than expected



Why is CO weak?

- CO frozen out **Included**
- CO (isotope-selective) photodissociation **Included**
- CO locked up in large bodies in midplane **Possible**
- CO / volatile carbon transformed **???**
 - Gas-phase chemistry
 - Ice chemistry $\rightarrow$ CH₃OH, CO₂, CH₄, C₂H₆

Key parameters: temperature, time, cosmic ray ionization rate ζ

Aikawa et al. 1999

Eistrup et al. 2016, 2018

Wu et al. 2016, 2017

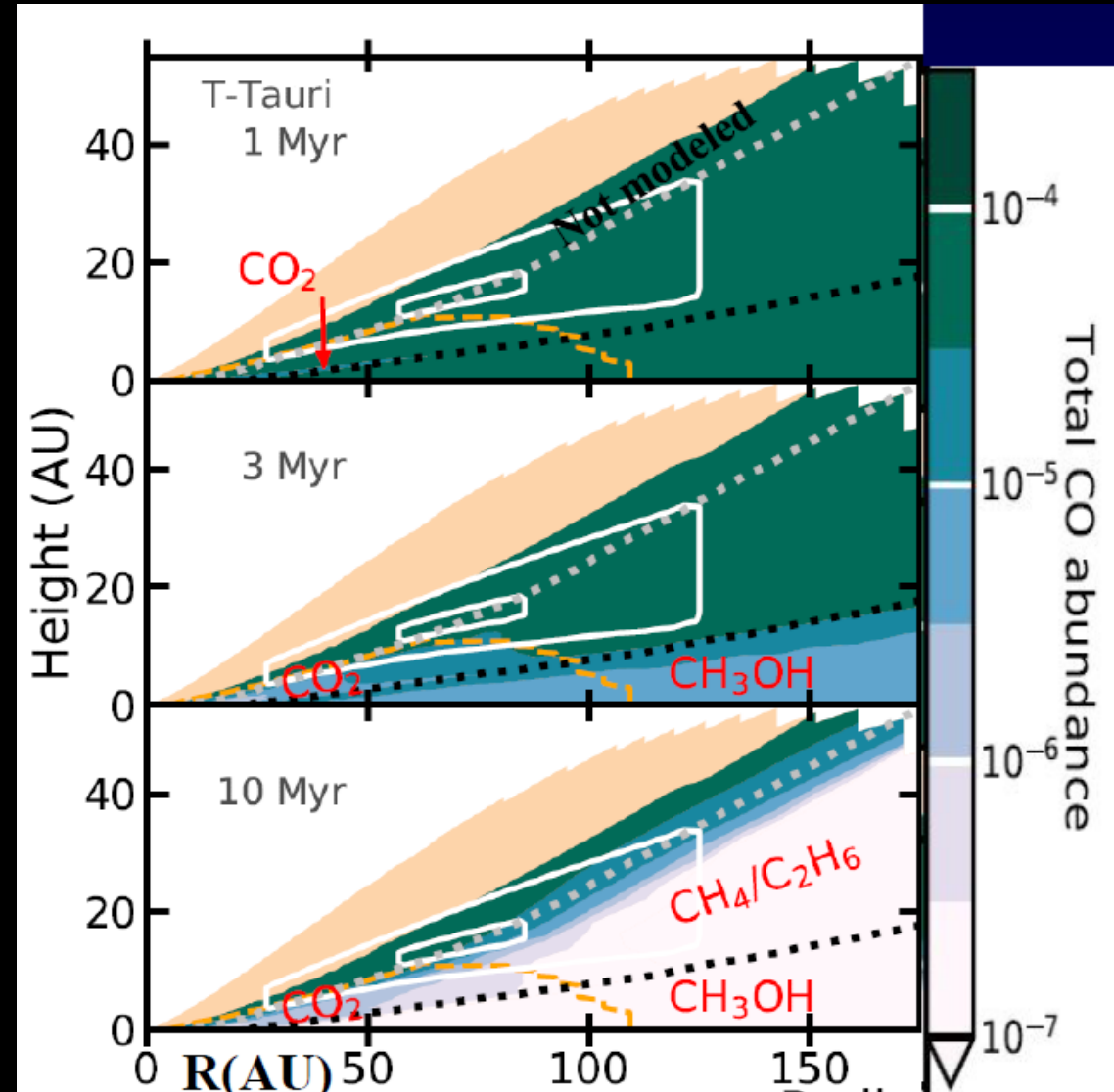
Molyarova et al. 2017

Schwarz et al. 2018

Bosman et al. 2018

CO depletion models

Young disk



$$\zeta = 10^{-17} \text{ s}^{-1}$$

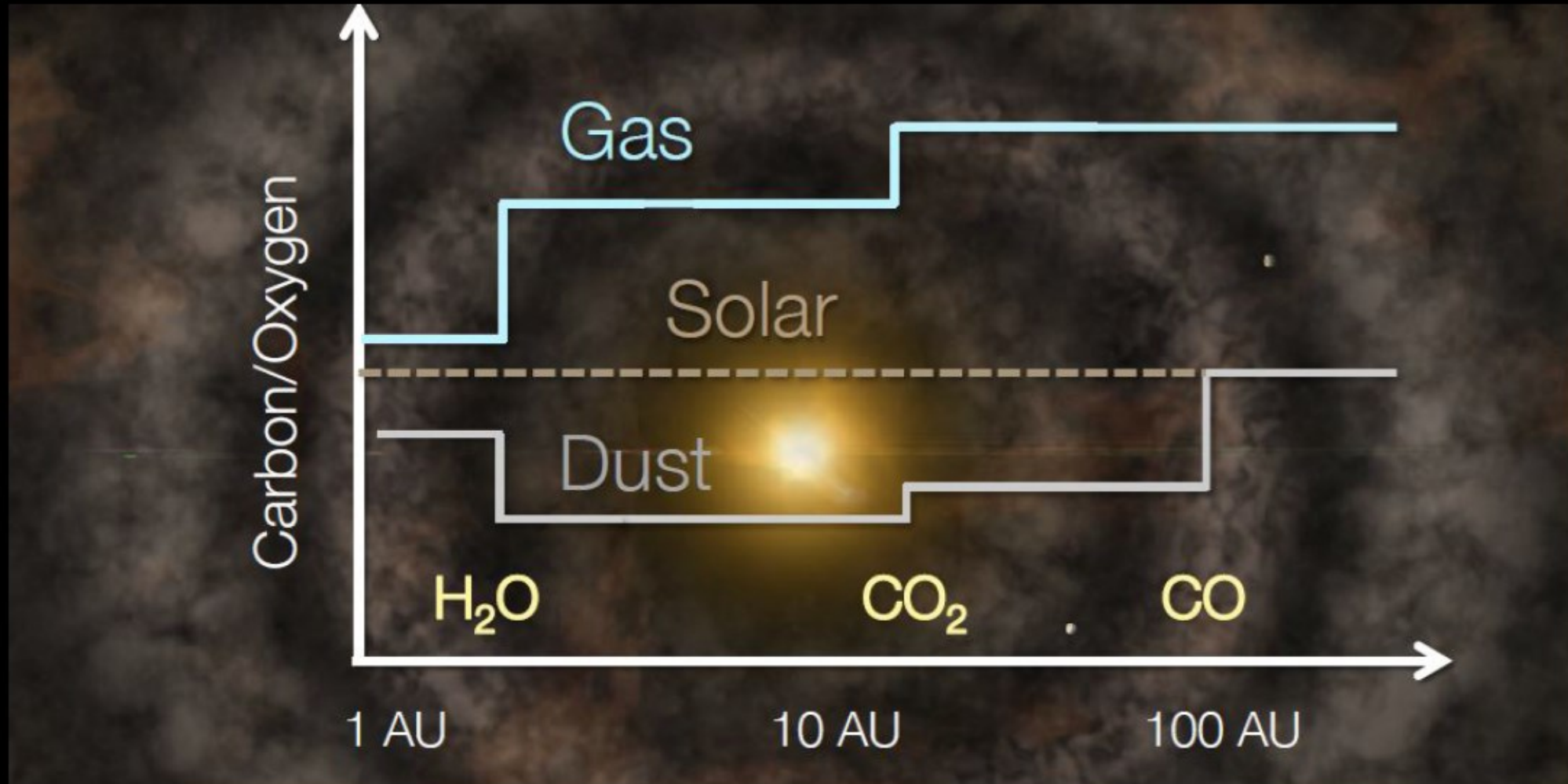
White contours:
25-75% emission

Bosman et al. 2018

Old disk

- Young disks show little CO transformation, in old disks most CO is missing

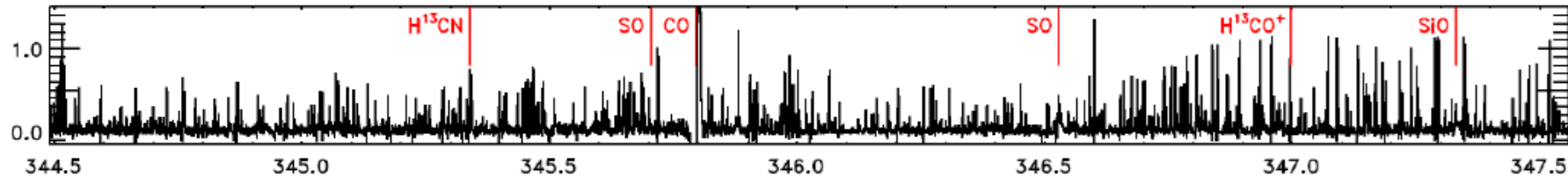
Changing C/O ratio with radius



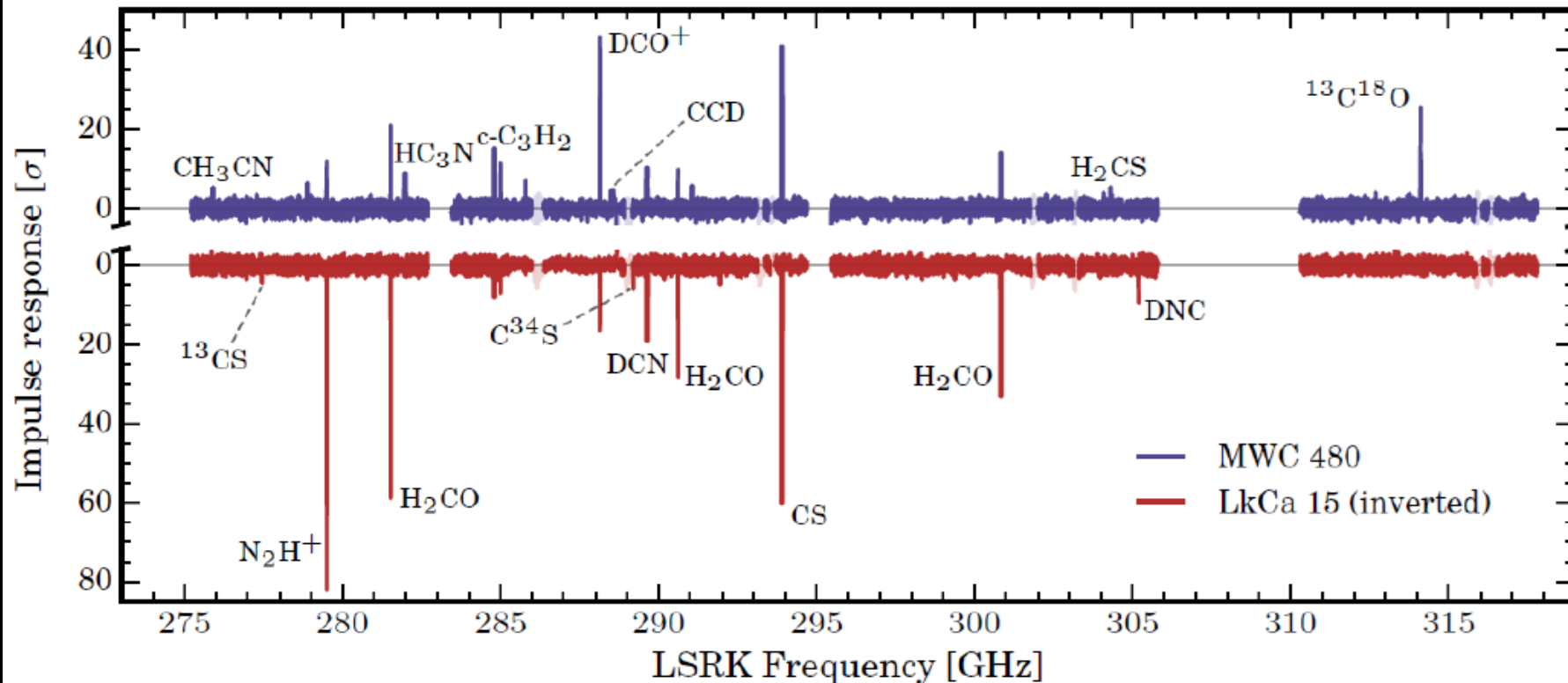
- Ices are oxygen rich
- Gas is carbon rich
- But both C and O are depleted

Mature disks have few COMs

Class 0: IRAS16293B PILS: 3 GHz, ~1000 lines

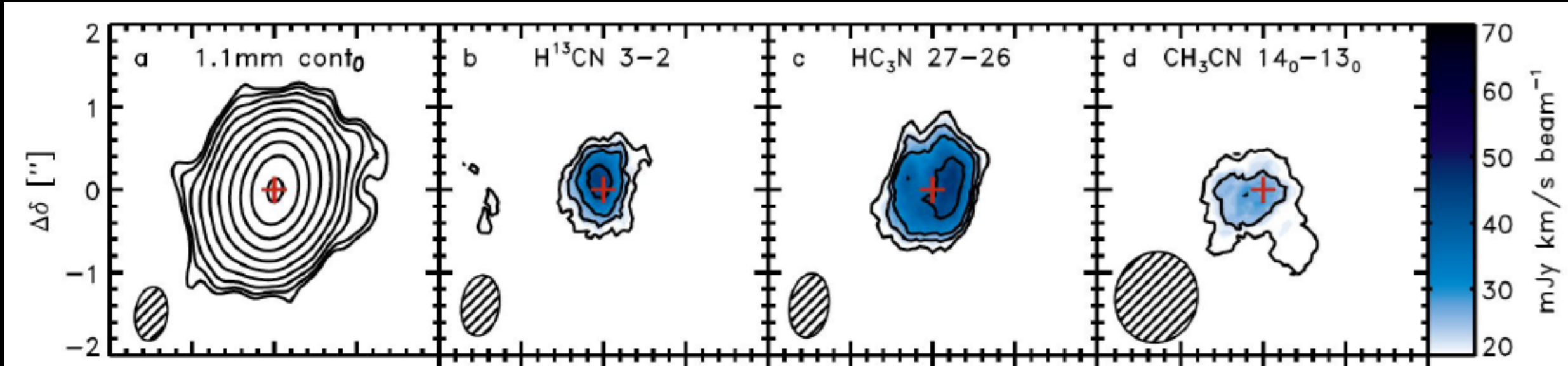


Class II: MWC 480: 30 GHz, ~20 lines



Complex molecules in disks

MWC 480 Herbig disk with ALMA: CH₃CN

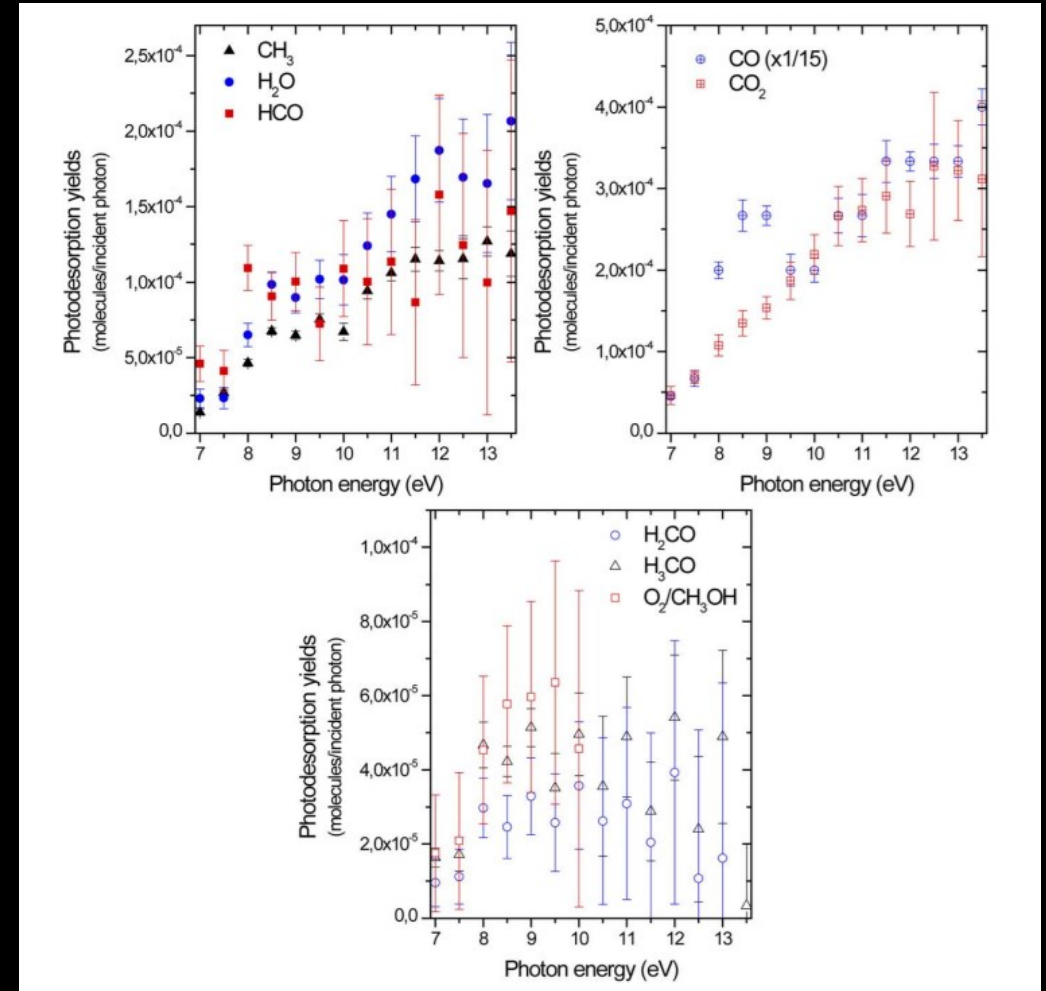
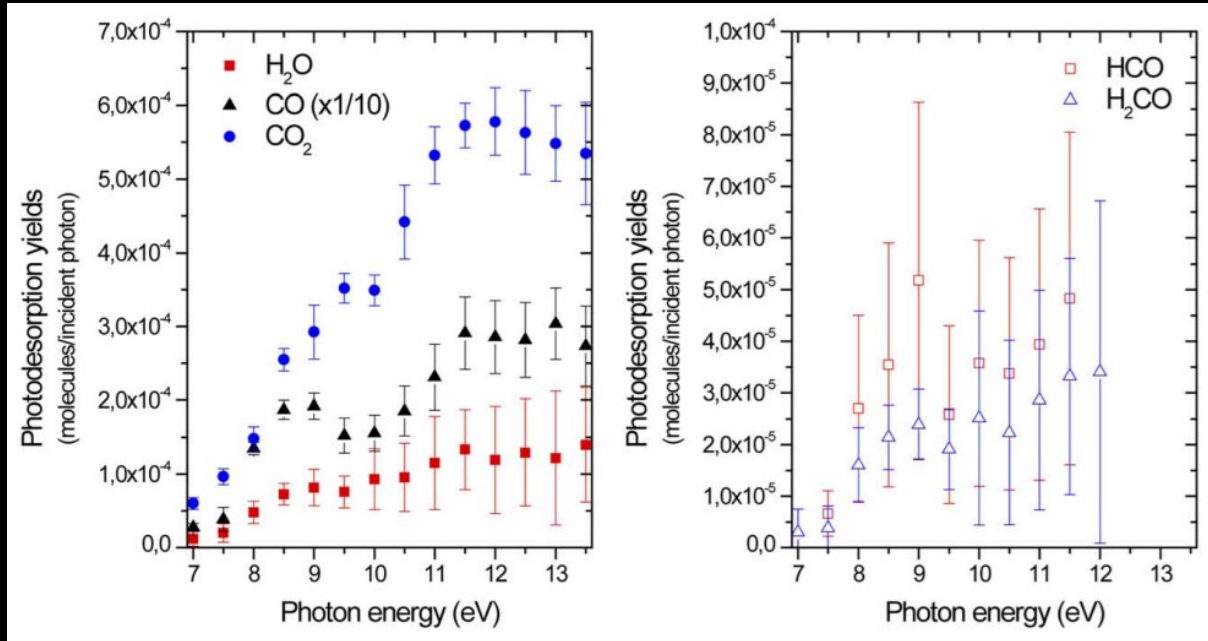


- $\text{HC}_3\text{N}/\text{HCN} \sim 0.4$, $\text{CH}_3\text{CN}/\text{HCN} \sim 0.05$ at 30 AU
- CH_3OH detected in TW Hya disk, but weak

Öberg et al. 2015
Walsh et al. 2016

Insight from ice experiments

UV photodesorption destroyed O-bearing COMs but retains cyano-COMs



Bertin et al. 2023

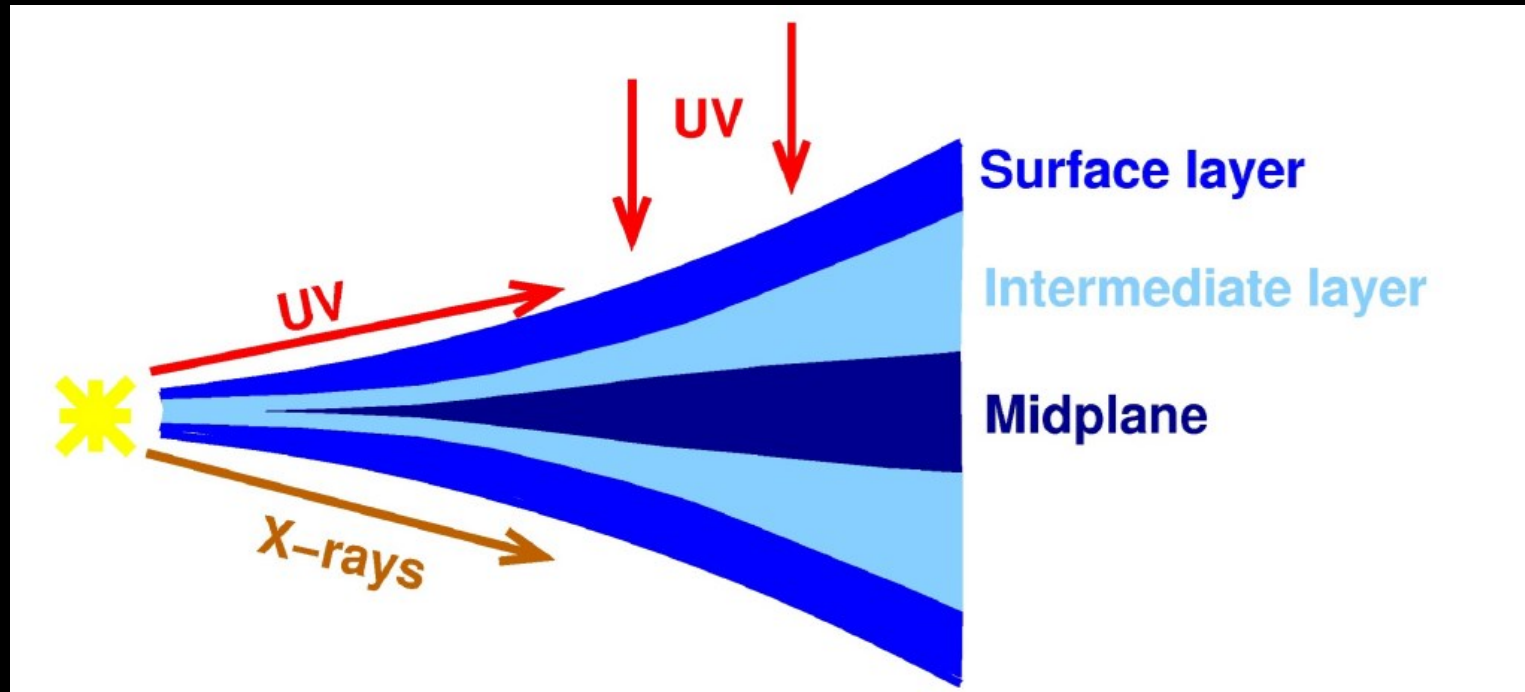
- Pure HCOOH and HCOOCH₃ ices – only fragments desorb
- CH₃CN photodesorbs intact!

Observational findings

- Simple gas-phase molecules observed (>50 AU)
 - Ion-molecule reactions (HCO^+)
 - Photon processes (high CN/HCN)
 - High deuterium fractionation (DCO^+)
 - Low abundances of complex species (H_2CO)
- Disk-averaged abundances are “depleted” by factor of 5-100 (using mass from dust continuum and assuming $\text{gas}/\text{dust} = 100$) due to freeze-out in cold midplane and photodissociation at the surface
- Transformation of CO to CO_2 , CH_3OH , hydrocarbons at low T
- Solid CO and H_2O detected in edge-on disks
 - Not yet CH_3OH ice
- Absence of PAHs – now confirmed with JWST! (Arulanantham et al. 2024)
 - Most T Tauri disks do not show PAH emission, abundances < 0.1 ISM

Chemical structure of 2D flaring disks

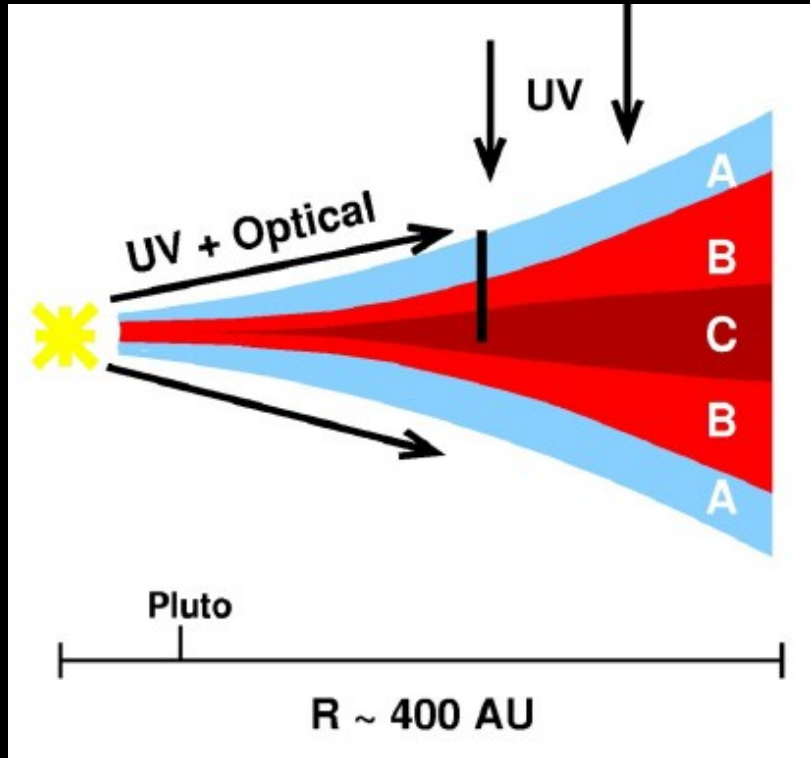
- Surface layer: molecules dissociated by UV photons
- Warm intermediate layer: molecules not much depleted, rich chemistry
- Cold midplane: molecules heavily frozen out



- PDR-like structure but with large density gradients and very intense UV (up to 10^6 x ISM)

Aikawa et al. 2002
Bergin et al. 2007, PPV +
many other groups

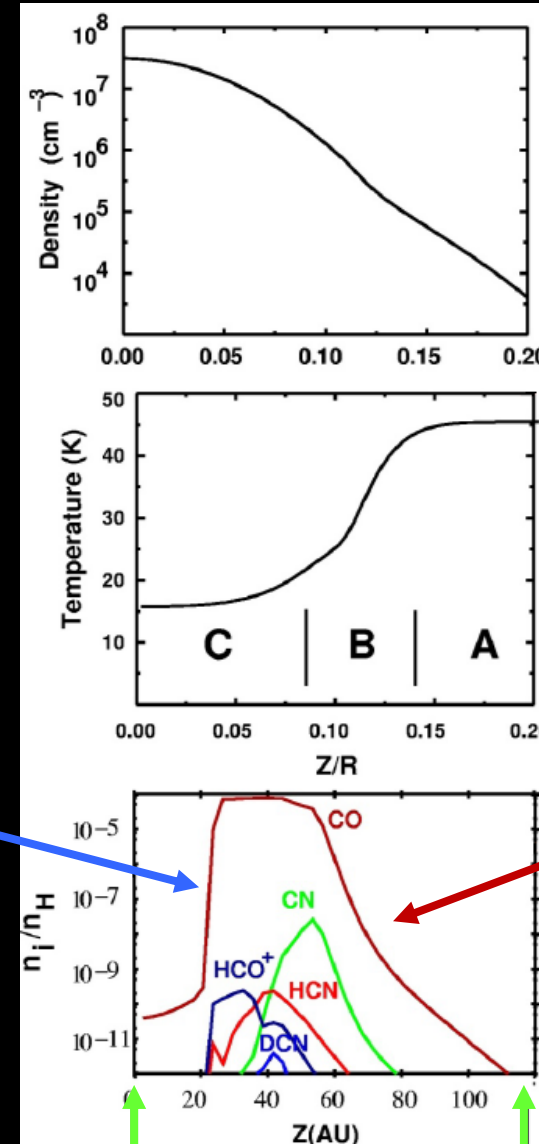
Vertical structure



Ionization fraction:

- Surface: 10^{-4} (C^+)
- Intermediate: 10^{-9} (HCO^+)
- Midplane: $\sim 10^{-11}$ (H_3^+ , H_2D^+)

Freeze-out



$R = 200 \text{ AU}$

T_{gas} is larger

Photodissociation

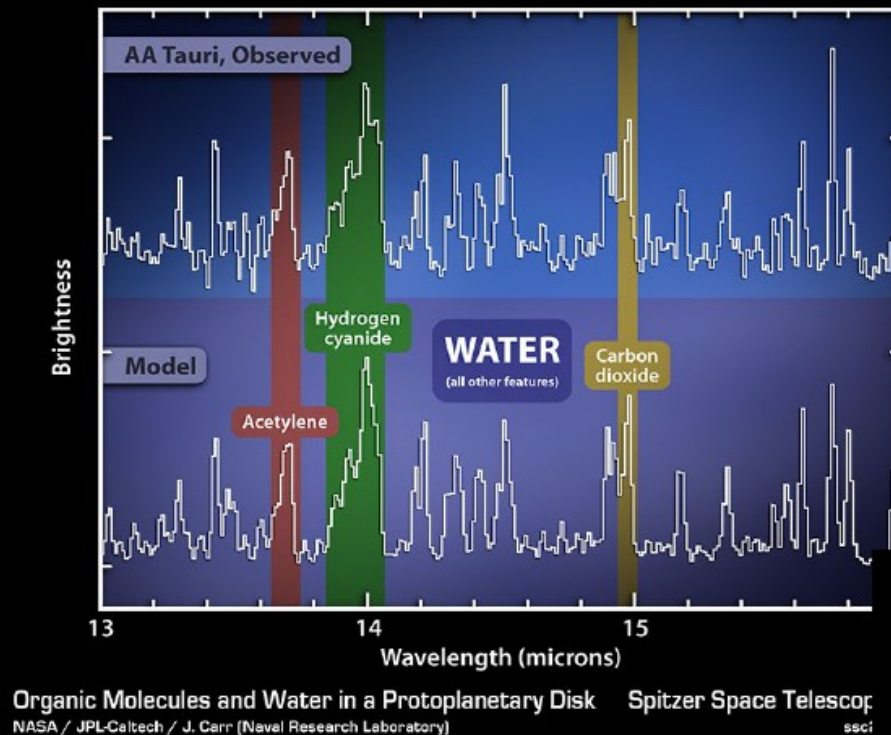
midplane

surface

7.8 Inner disk (<10 AU)

- Surprising detections of mid-IR observations of CO, H₂O, OH, HCN, C₂H₂, ...
- Many sophisticated thermo-chemical models being developed
 - Hollenbach & Gorti, Glassgold, Meyerink et al., Woitke, Kamp & Thi (PRODIMOO), Agúndez et al., Bruderer (DALI), ...

Hot water and organics in inner few AU of disks

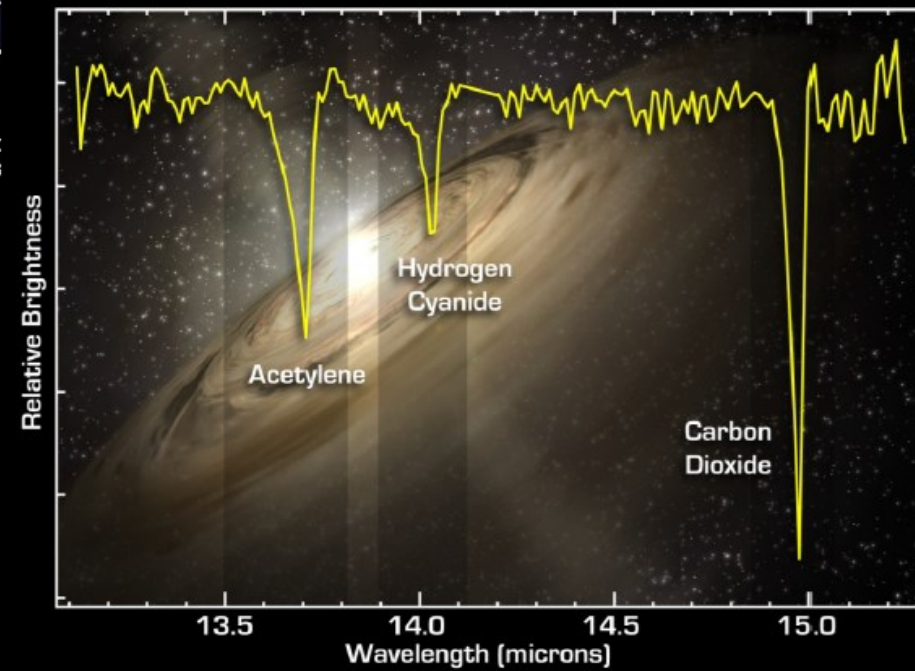


Carr & Najita 08
Salyk + 08, 11
Pontoppidan+10

$$T_{\text{ex}} = 300\text{--}700 \text{ K}$$

**Spitzer observations of
T Tauri stars show rich spectra
of simple molecules
 H_2O , HCN , C_2H_2 , CO_2**

Lahuis +06

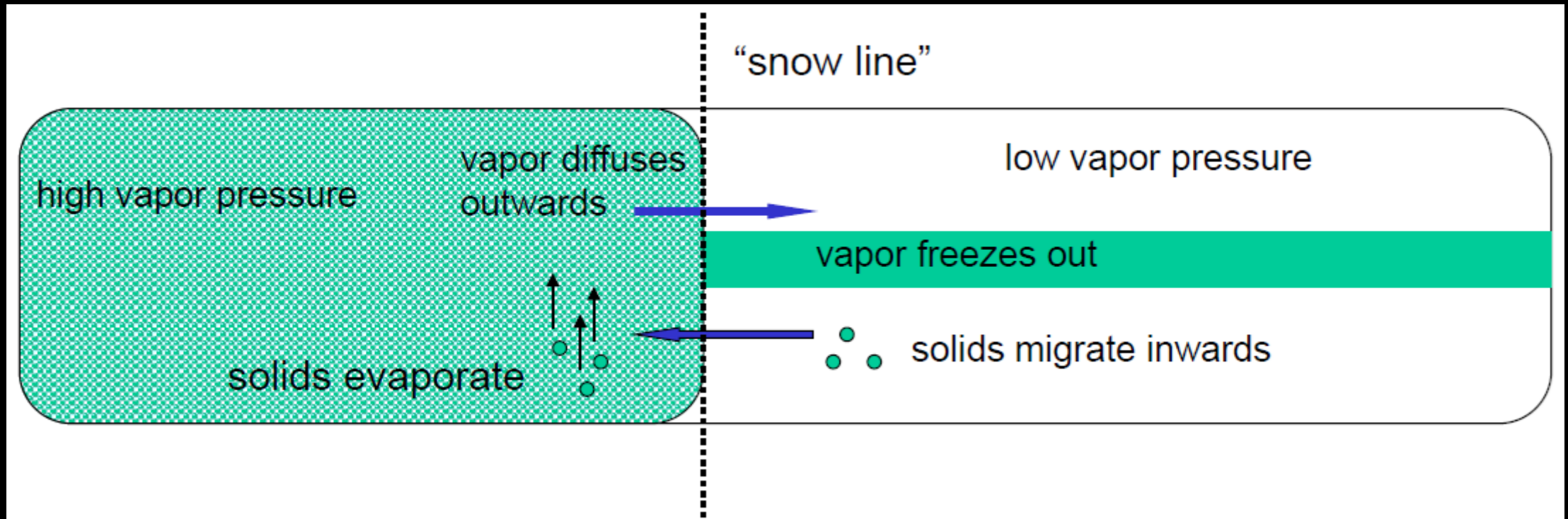


Implications

- Water gas is found well inside the ‘snow’ line (estimated at ~ 3 AU)
- Inward radial migration or upward mixing of icy grains and planetesimals, followed by evaporation?

How does disk water vapor depend on disk location and time?

- Water vapor contents may be set by a balance between vapor diffusion and icy planetesimal migration



- Inner disk water vapor content may reflect disk evolution

7.9 Cometary chemistry

- Comets are thought to have formed in the outer part of the disk which formed our solar system
- Comets contain the least-modified original interstellar material
- Early optical observations probed mostly daughter molecules (CH, CN, C₂, CO⁺, OH, OH⁺, ...)
- IR and (sub)mm observations provide direct information on parent molecules, i.e., *original composition of ices*
- Lots of data from comets Halley, Hyakutake, and Hale-Bopp

Parent-daughter model

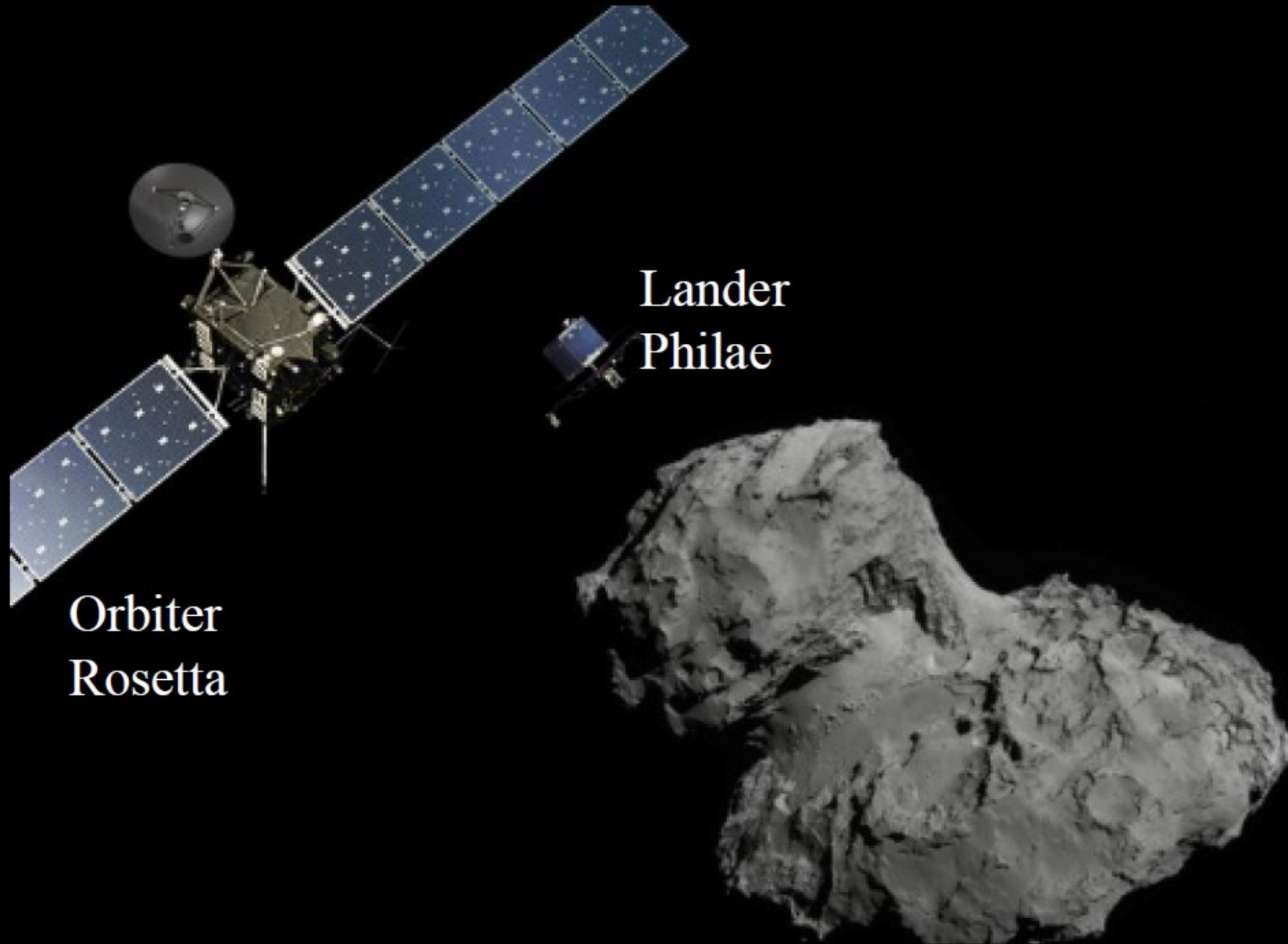
- As comet approaches Sun, ices heat up and evaporate. Radiation from the Sun photo-dissociates molecules:
 - $\text{H}_2\text{O} \rightarrow \text{OH} \rightarrow \text{O}$
 - $\text{NH}_3 \rightarrow \text{NH}_2 \rightarrow \text{NH}$
 - $\text{CH}_4 \rightarrow \text{CH}_2/\text{CH}_3 \rightarrow \text{CH}$
 - $\text{HCN} \rightarrow \text{CN}$

Ice abundances

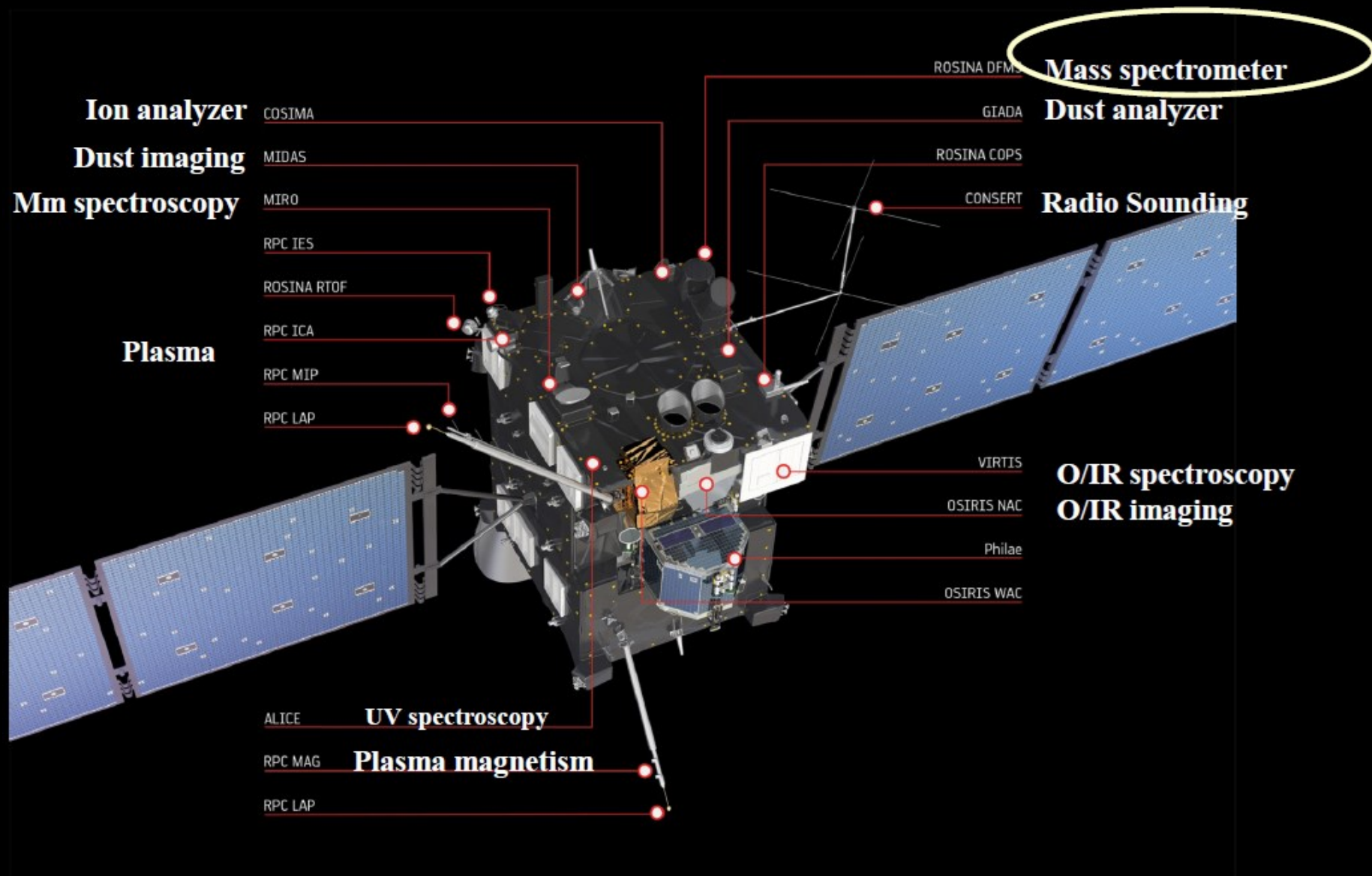
Species	Protostar ices	Comets
H ₂ O	100	100
CO	3-10	6-30
CO ₂	10-35	2-20
CH ₃ OH	1-25	2
H ₂ CO	1-7	0.2-1
HCOOH	0.4-2	0.1
NH ₃	1-10	0.5-2
CH ₄	0.5-2	1

Similar abundances => some fraction of cometary ices are unaltered ISM ices?

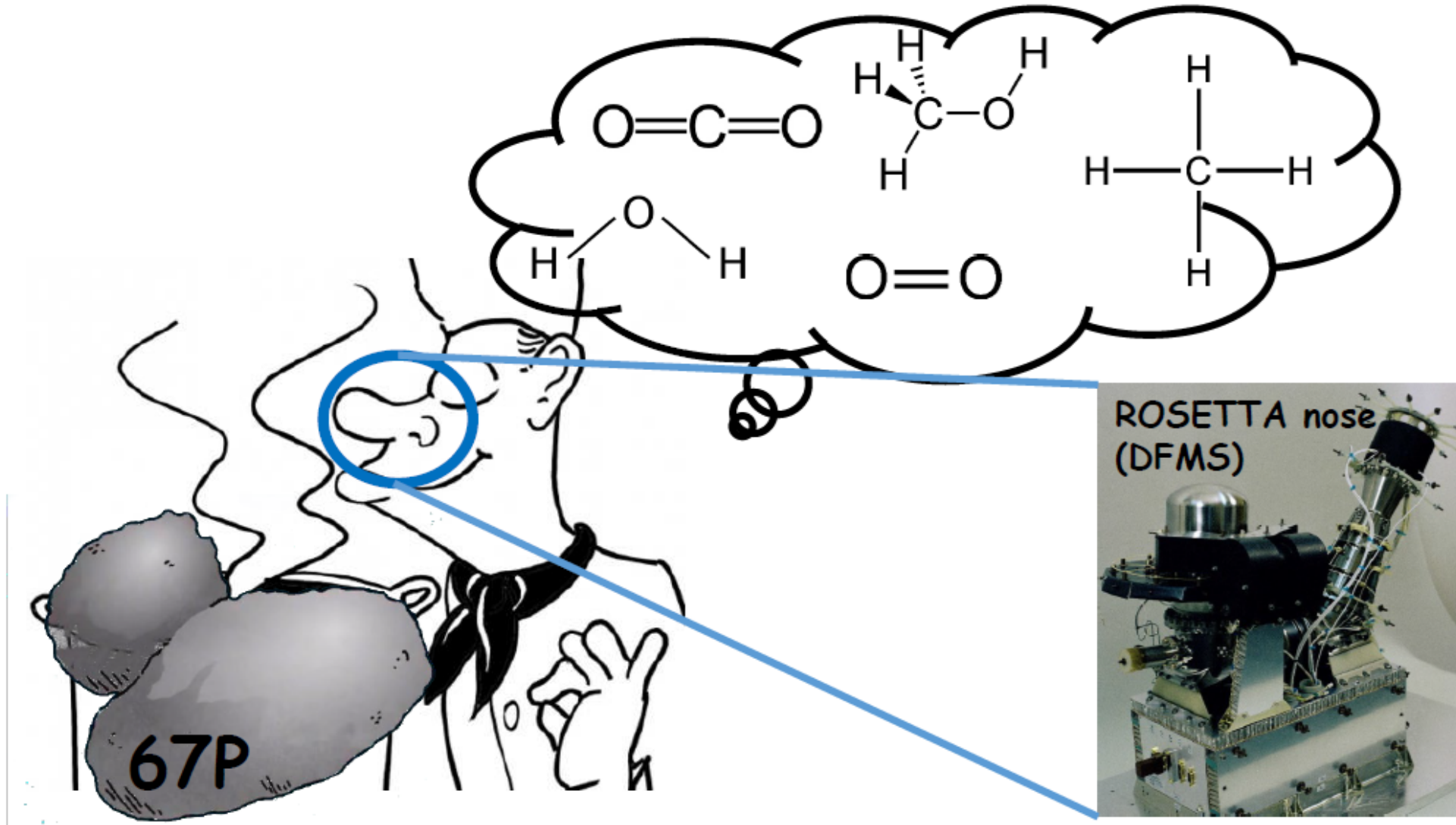
Rosetta and the origin of our solar system



Instruments on orbiter

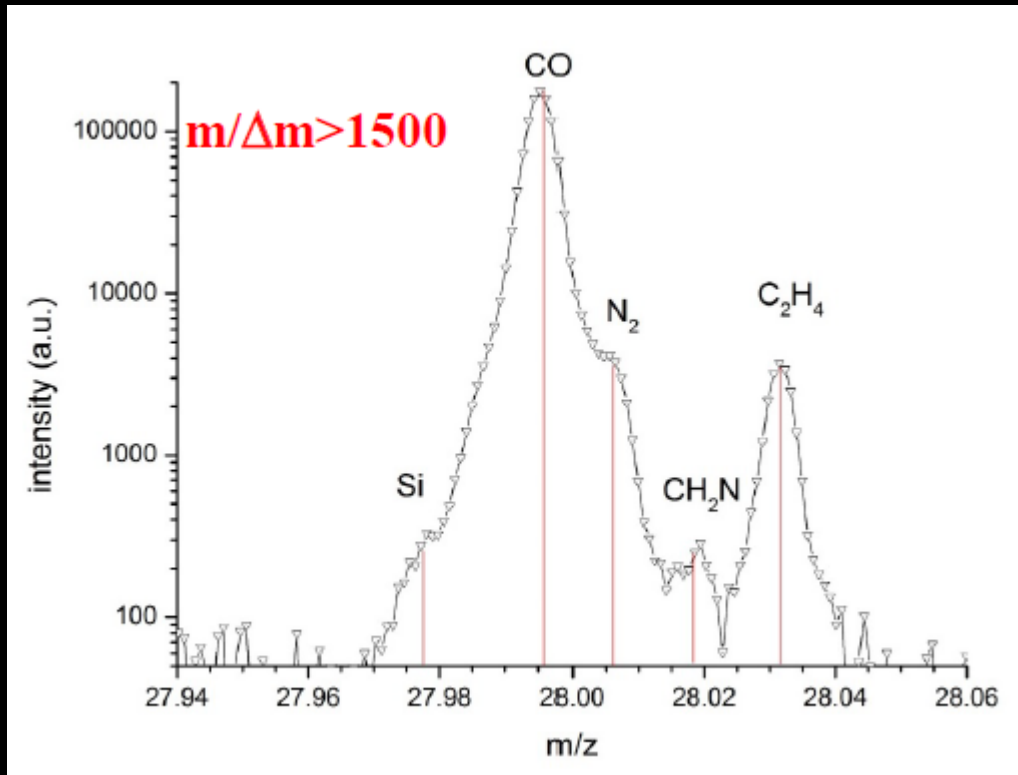


ROSETTA has a very good nose (DFMS)

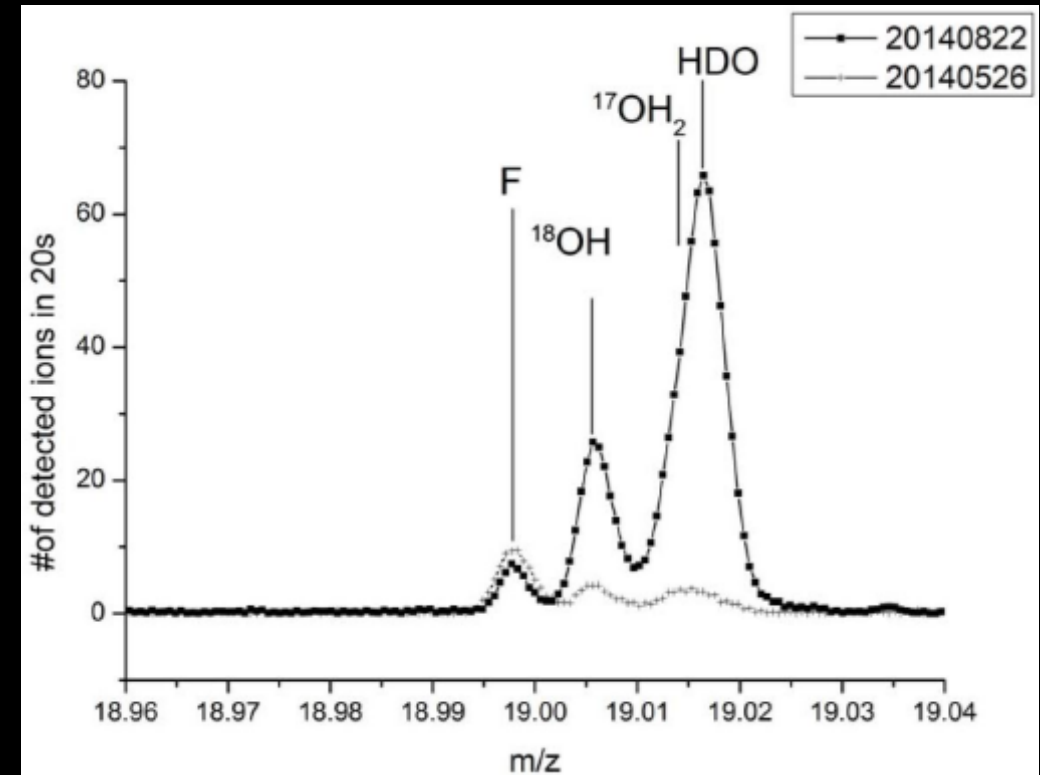


High mass resolution → unique identification

m/z 28: CO vs N₂

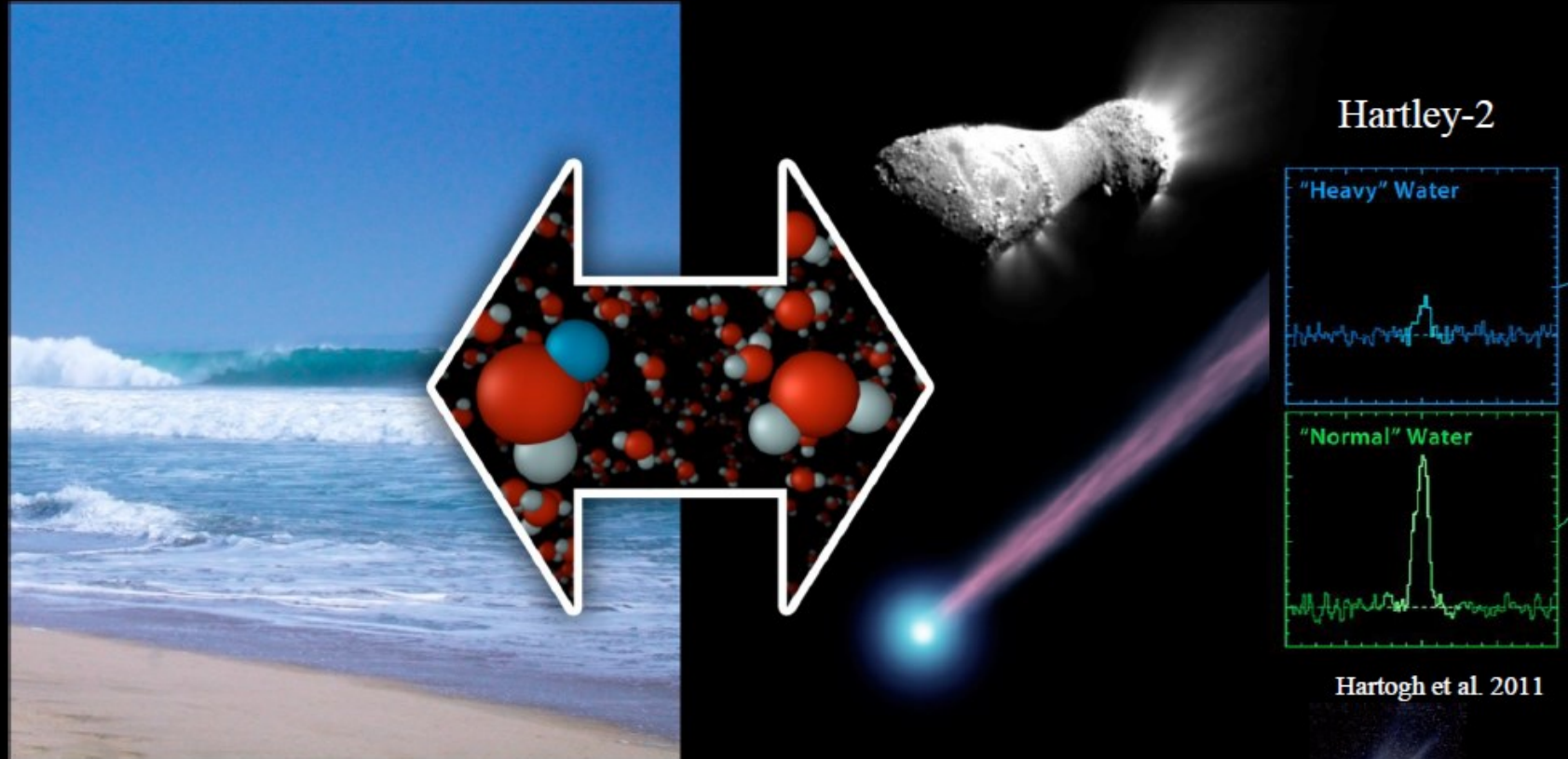


m/z 19: HDO vs H₂¹⁷O vs ¹⁸OH



ROSINA:
Altwegg et al. 2014
Rubin et al. 2015

Did comets bring water to Earth?

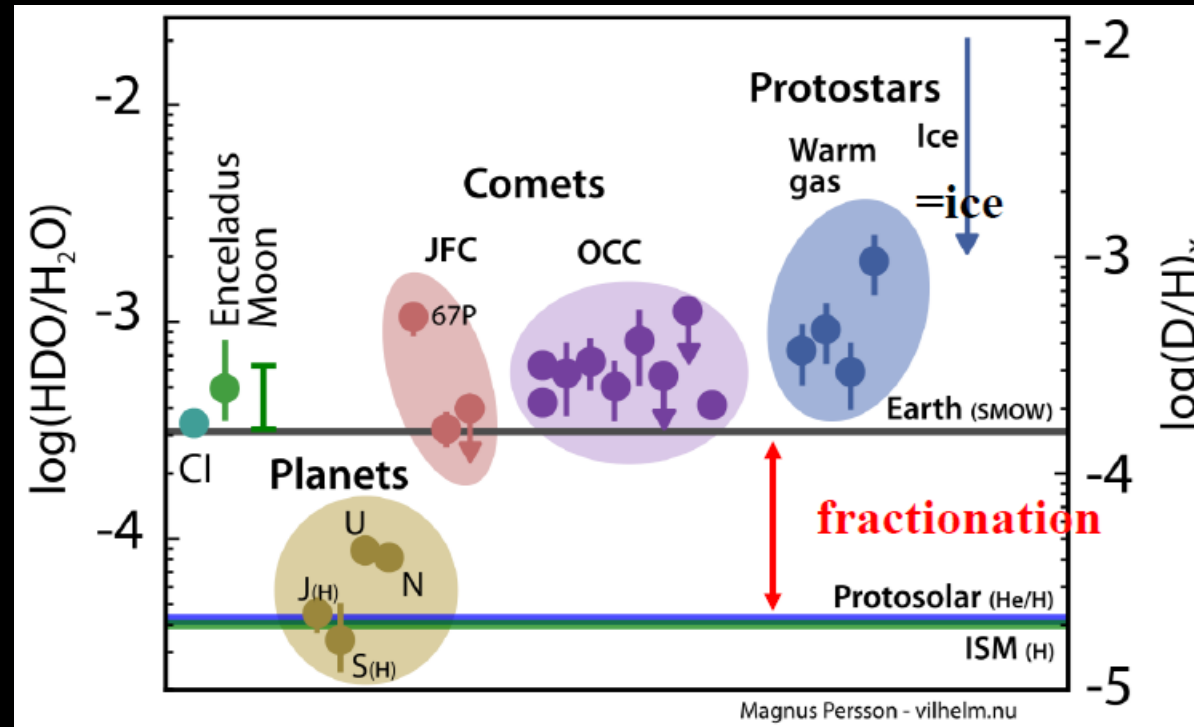


HIFI: There is at least one family of comets with same abundance of heavy water:

$$\text{HDO}/\text{H}_2\text{O} = 1.5 \times 10^{-4}$$



HDO/H₂O tracing history of Solar system?

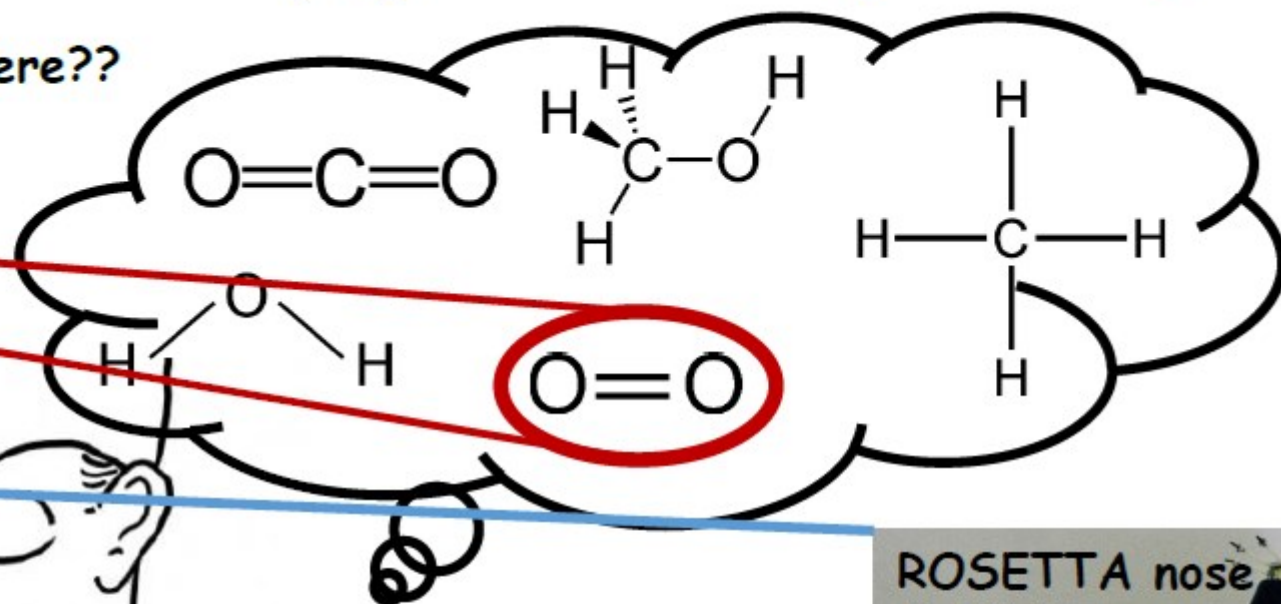
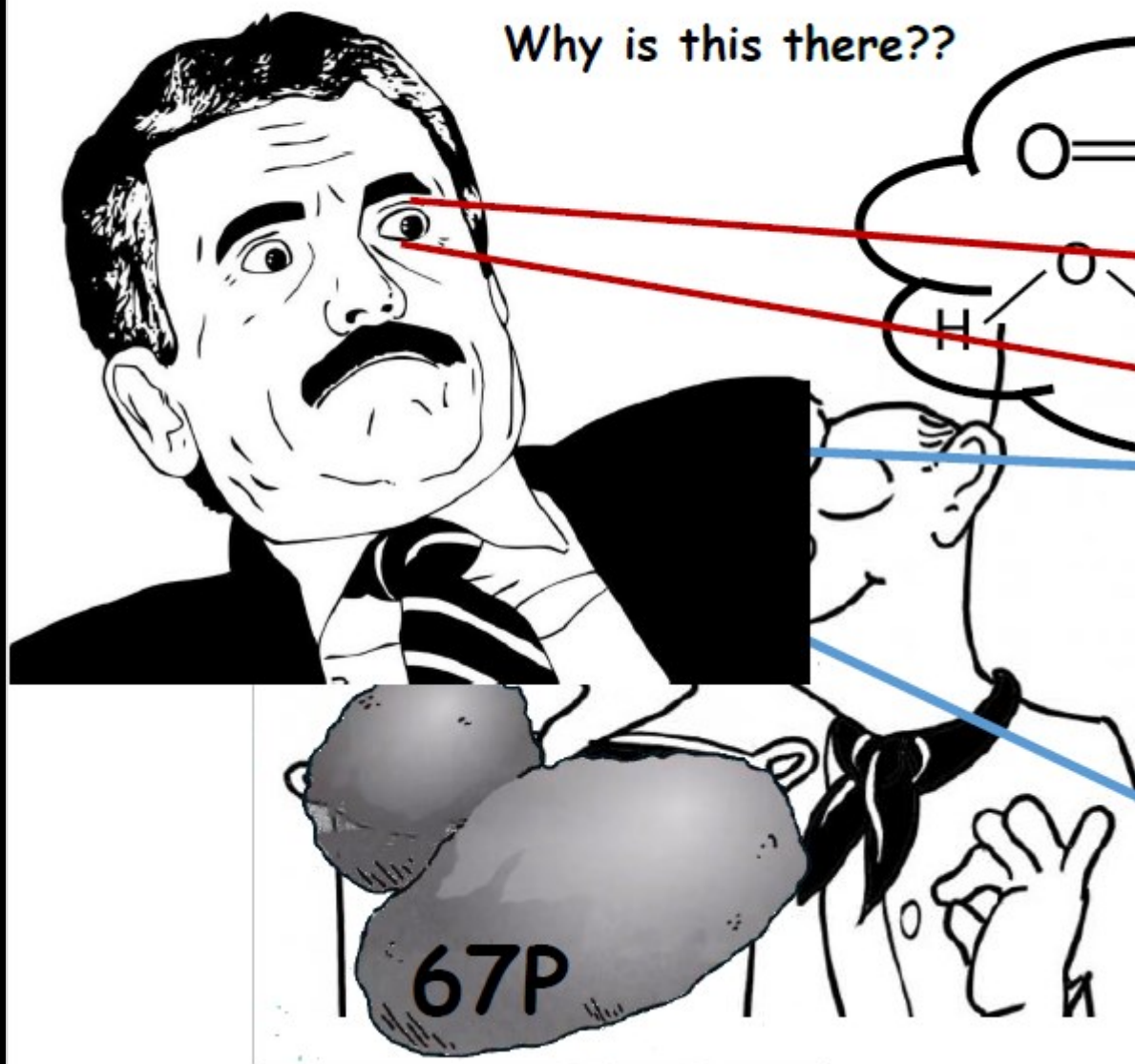


Persson et al. 2014
Bockelée-Morvan et al. 2014
Ceccarelli et al. 2014
Altwegg et al. 2014

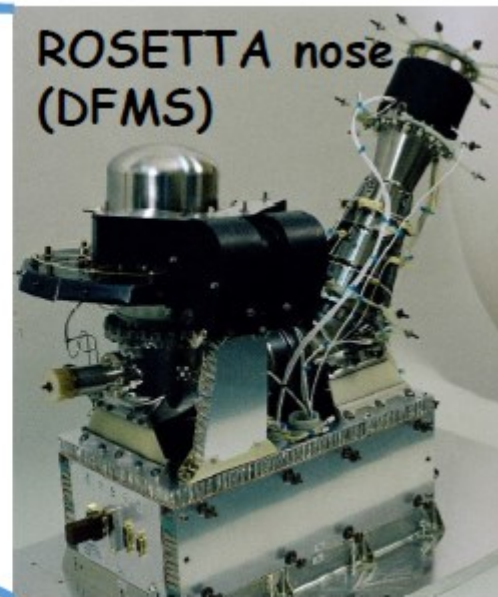
- Similarity of comets and interstellar ices could reflect inheritance of pre-stellar ices in large planetesimals that formed early
- Mixing in early solar system brought some of these water-rich bodies toward the terrestrial planet forming zone
- But comets do not have uniform D/H ratios whereas meteorites do.
- Current consensus: Comets probably contribute, but are not the sole originators of H₂O

ROSETTA has a very good nose (DFMS)

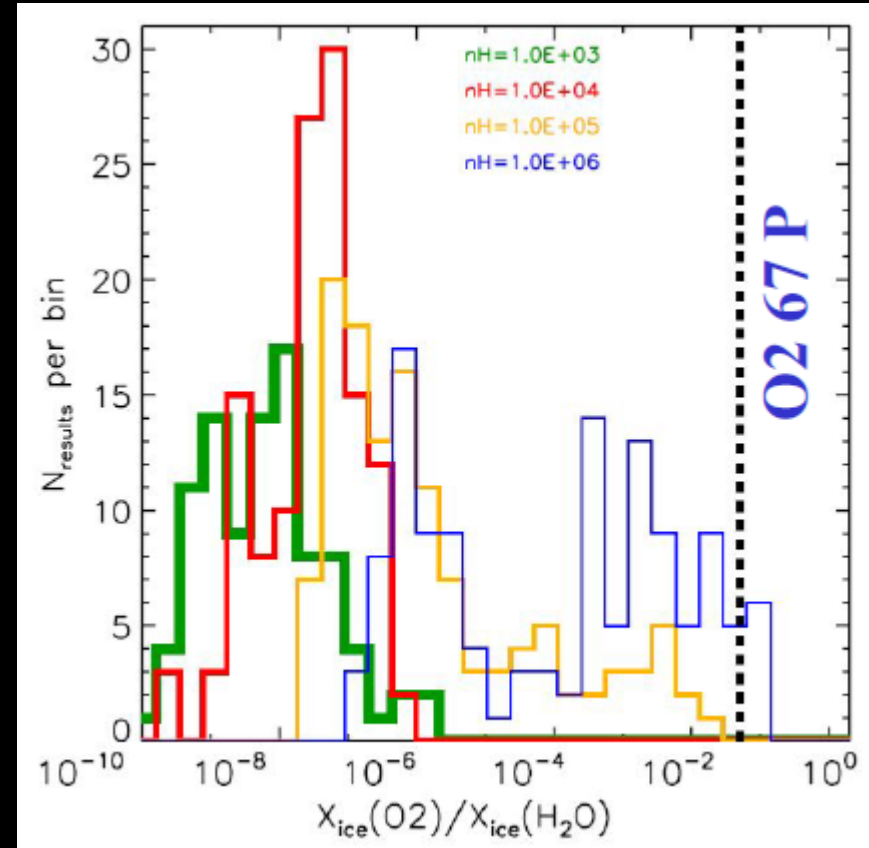
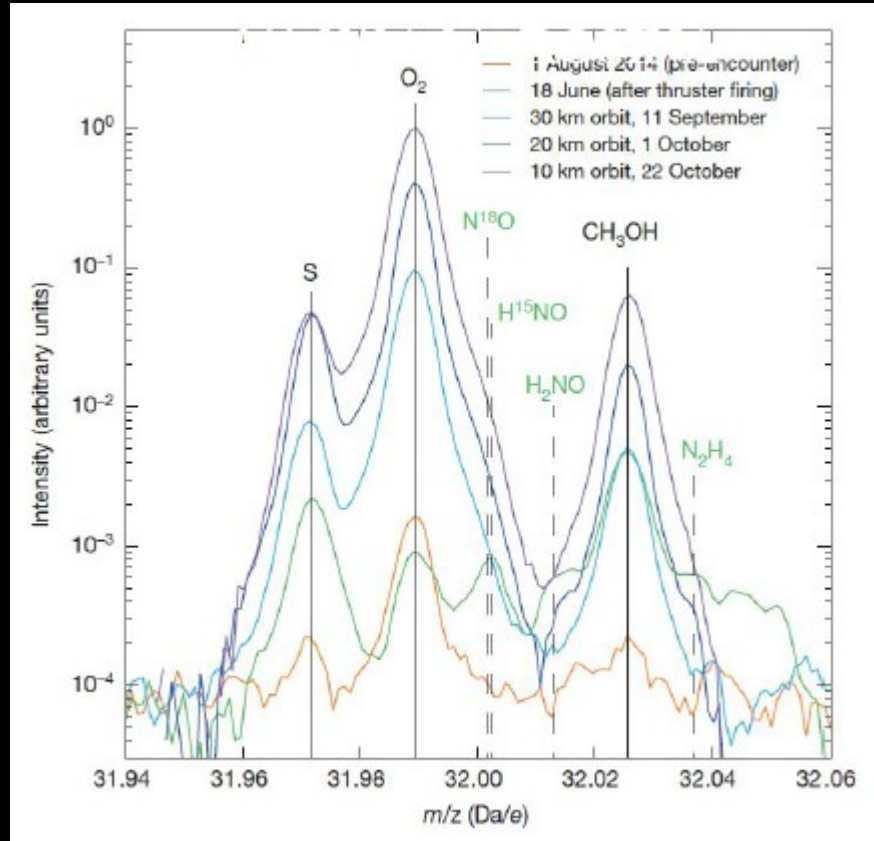
Why is this there??



ROSETTA nose
(DFMS)



Abundant O₂ mystery



Taquet et al. 2016
Walsh et al. 2015

O₂/H₂O = 3.7 ± 1.5%, No O₃!

- Problem: O₂ readily transformed to H₂O in ice (remember video!)
- Only models with low H/O ratio (high density) and relatively warm conditions ($T \sim 20\text{-}30\text{ K}$) can reproduce high observed O₂/H₂O ice ratio.

Organics on the surface

Table 1. The 16 molecules used to fit the COSAC mass spectrum.

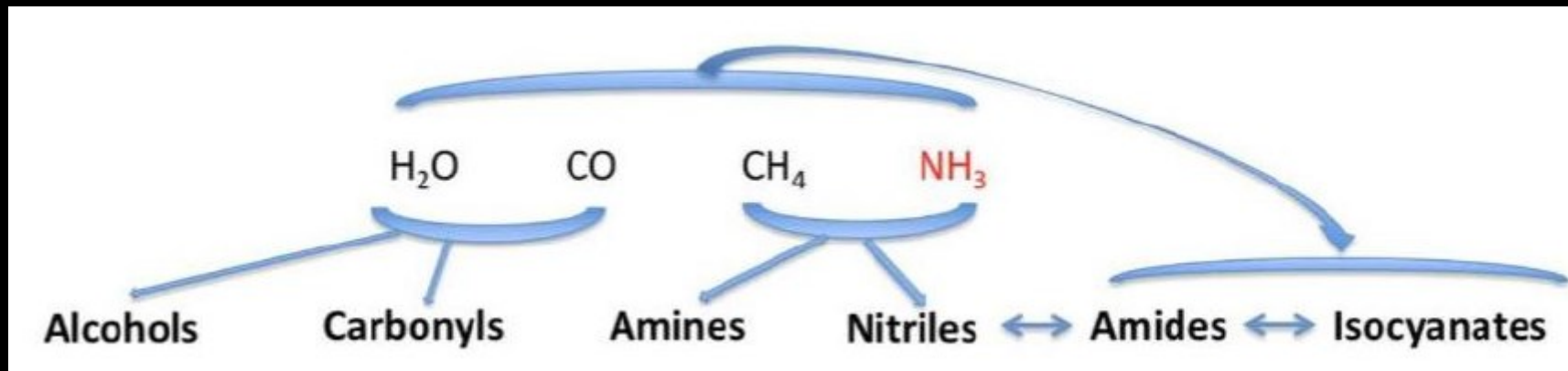
Name	Formula	Molar mass (u)	MS fraction	Relative to water
Water	H ₂ O	18	80.92	100
Methane	CH ₄	16	0.70	0.5
Methanenitrile (hydrogen cyanide)	HCN	27	1.06	0.9
Carbon monoxide	CO	28	1.09	1.2
Methylamine	CH ₃ NH ₂	31	1.19	0.6
Ethanenitrile (acetonitrile)	CH ₃ CN	41	0.55	0.3
Isocyanic acid	HNCO	43	0.47	0.3
Ethanal (acetaldehyde)	CH ₃ CHO	44	1.01	0.5
Methanamide (formamide)	HCONH ₂	45	3.73	1.8
Ethylamine	C ₂ H ₅ NH ₂	45	0.72	0.3
Isocyanomethane (methyl isocyanate)	CH ₃ NCO	57	3.13	1.3
Propanone (acetone)	CH ₃ COCH ₃	58	1.02	0.3
Propanal (propionaldehyde)	C ₂ H ₅ CHO	58	0.44	0.1
Ethanamide (acetamide)	CH ₃ CONH ₂	59	2.20	0.7
2-Hydroxyethanal (glycolaldehyde)	CH ₂ OHCHO	60	0.98	0.4
1,2-Ethanediol (ethylene glycol)	CH ₂ (OH)CH ₂ (OH)	62	0.79	0.2

COSAC instrument on Philae

“sniffing the ground truth”

Goesmann et al. 2015

Note: COSAC has $m/\Delta m \sim 1$: cannot distinguish species with same mass



ROSINA Z

Methane
Ethane
Propane
Butane
Pentane
Hexane
Heptane



Glycine

(Aminoacid)

S₂
S₃
S₄



Methanethiole (CH₃SH)

Ethanethiol (C₂H₅SH)

Thioformaldehyde
(CH₂S)

Ammonia

Methylamine

Ethylamine



Benzene

Toluene

Xylene

Benzoic acid

Naphthalene



Formic acid

Acetic acid

Acetaldehyde

Ethylenglycol

Propylenglycol

Butanamide



Argon

Krypton

Xenon

Acetylene

HCN

Acetonitrile

Formaldehyde



HF

HCl

HBr

P

CH₃Cl



N≡N

Nitrogen

Oxygen

Hydrogenperoxyd

Carbon monoxide

Carbon dioxide



Methanol

Ethanol

Propanol

Butanol

Pentanol



Cyanogen



Na, K, Si,
Mg

Hydrogensulfide

Carbonylsulfide

Sulfur monoxide

Sulfur dioxide

Carbon disulfide



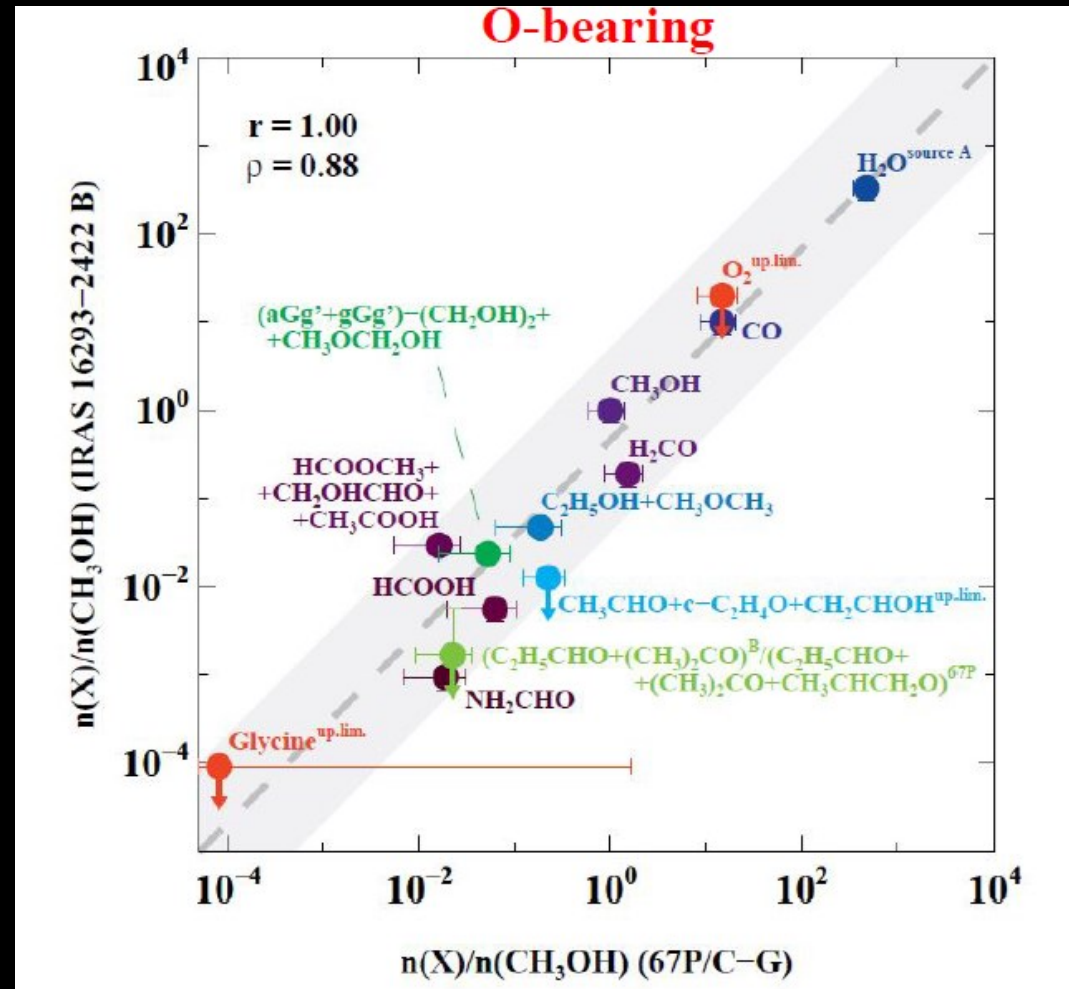
Detected in IRAS16293

Upper limit

Drozdovskaya et al. 2019
Altwegg et al. 2019, ARAA



Comparison of young disk and comet



- Overall similarity points to at least some preservation
- Comet somewhat richer in COMs than young disk
- (also holds for N- and S-bearing species)

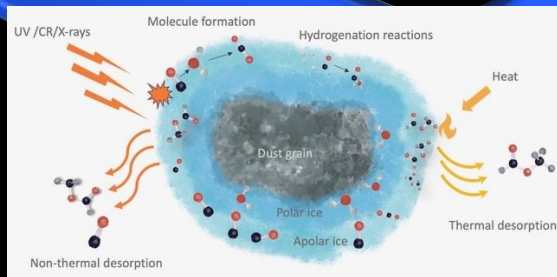
Building the Solar system's organic inventory

CO reservoir

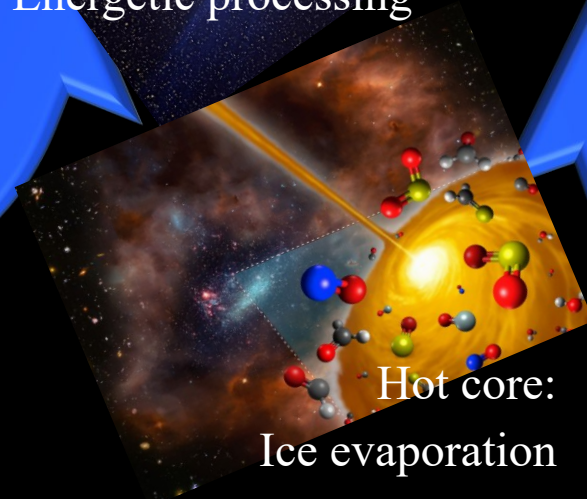


Gas:
ion-molecule reactions
Cosmic-ray photolysis

Ices:
Hydrogenation, photolysis, thermal
polymerization, ice-ion-molecule,
ice segregation



Comets:
Energetic processing



Hot core:
Ice evaporation
Ion-molecule reactions



Asteroids:
Aqueous alteration

PAH reservoir



Stars:
Soot chemistry
Shock chemistry

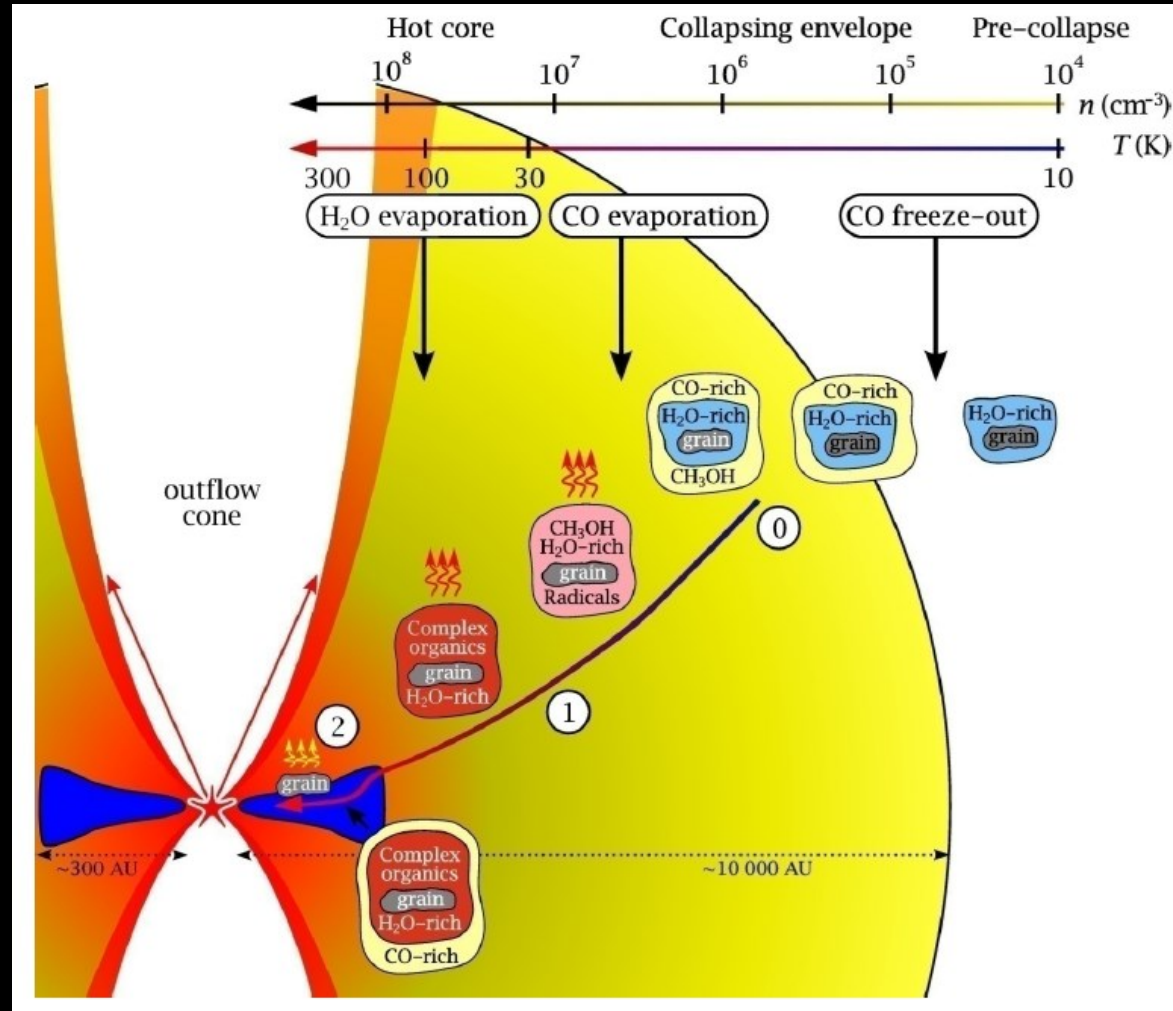
Nebulae:
UV & X ray photolysis
Radical reactions
Hydrocarbon chemistry
Fischer-Tropsch
Shocks, intermittent accretion,
diffusion

Right-hand branch not discussed in these lectures => [Astrochemistry 2!](#)

7.10 Summary

- Systematic trends in ices and gas-phase data allow chemical scenario during low- and high-mass star formation to be developed
 - ‘Quantitative astrochemistry has come of age’
- Chemistry in pre-stellar cores and cold envelopes driven by freeze-out
 - Directly observed through ices
 - Explains lack of O_2 , extreme deuteration, N_2H^+ ...
- Chemistry in inner warm envelope driven by sublimation of ices followed by a ‘hot core’ chemistry
 - Most complex organics produced as zeroth or first generation on surfaces?

Summary: molecules from cloud to disks



Herbst & van Dishoeck 2009

Visser et al. 2009, 2011

- 0th generation: cold ices: CH $_3$ OH + ?
- 1st generation: warm ices: more complex organics
- 2nd generation: high- T gas-phase chemistry

Summary (continued)

- Different IR and sub-mm features can be used as diagnostics of the earliest stages of star formation
 - Cold pre/proto-stellar cores: N_2H^+ , H_2D^+ , ices
 - Warm inner envelopes: CH_3OH
 - Hot cores: Complex organics
 - Shocks: SiO , SO , SO_2

Summary (continued)

- Chemistry in disks being studied
 - Inner disk: high T chemistry at high densities
 - Outer disk: layered chemical structure
- Comets provide clue to composition of material from which our solar system is made
 - Large similarity between cometary and interstellar ices
- Laboratory data and theoretical studies of basic molecular processes remain essential for further progress.

Future is bright with ALMA and JWST in full swing and ELT, Ariel, Habitable Worlds coming soon!

The ALMA (r)evolution

- New molecules
 - More complex (prebiotic) species
 - Weeding out known molecules
 - Known molecules outside Sgr B2 / Orion
- Spatial distribution
 - Resolve relevant physical-chemical scales
- New processes and excitation
 - Probe hotter gas, state-selective chemistry
- Extragalactic chemistry
 - High-z galaxies like Orion 40 years ago

The JWST (r)evolution

- CO₂ in exoplanet atmospheres
- Photochemistry in exoplanet atmospheres (SO₂)
- Detection of the methyl (CH₃⁺) cation in a star-forming region
- Complete inventory of ice chemistry (CH₃OH, OCS, numerous minor species)
- Rich carbon chemistry around young stars (C₂H₂, HCN, CO₂...) and hydrocarbons from dust processing close to host in carbon-rich protoplanetary disk (C₄H₂, benzene...)
- PAH features in T Tauri stars

JWST & ALMA are astrochemistry machines!

Thank you for
coming to lectures!